Euler-Euler and Euler-Lagrange Modeling of Wood Gasification in Fluidized Beds

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Overview

- Motivation
- Modeling approaches
 - Euler-Lagrange / Discrete Element Method (DEM)

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- Euler-Euler
- Results
- Summary and outlook



Motivation - simulation of wood gasification

- Renewed interest in producing gas from biomass
- Gasification in fluidized bed reactors
- Lack of detailed knowledge about the complex interactions between heterogeneous reactions and the fluid mechanics in fluidized beds

• Develop closure laws for large scale models



CFD modeling for particulate flow

Continuum models

Euler-Lagrange

DNS

tractable problem size



Euler-Lagrange model

Basis:

OpenFOAM (www.opencfd.co.uk/openfoam)
Modules:

- Solver for the gasphase (OpenFoam)
- 2 Lagrangian particle tracking / DEM (OpenFOAM)
- Particle combustion / gasification model
 (here: simple zero dimensional particle model)



Euler-Lagrange model – gasphase

Mass / momentum balance:

 $(\rho\epsilon)_t + \nabla \cdot \epsilon\rho \mathbf{U} = \epsilon \dot{\rho}_s$

$$(\epsilon \rho u)_t + \nabla \cdot \{\epsilon \rho uu\} + \nabla p - \nabla \cdot \tau + \epsilon \rho g = F_s$$

Energy / species balance:

$$(\rho \epsilon E)_t + \nabla \cdot [(\rho E + p)\epsilon u] + \nabla \cdot q = \dot{Q}_s$$
$$(\rho \epsilon u Y_\alpha)_t + \nabla \cdot [\rho \epsilon Y_\alpha u] - \epsilon \dot{w}_\alpha = \epsilon \dot{\rho}_{\alpha,s}$$





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Euler-Lagrange model – particle tracking

Equation of motion for every single particle

$$m_{i}\frac{\mathrm{d}\vec{v}_{i}}{\mathrm{d}t} + \vec{v}_{i}\frac{\mathrm{d}m_{i}}{\mathrm{d}t} = \sum_{j}\vec{F}_{ij}$$
$$J_{i}\frac{\mathrm{d}\vec{\omega}}{\mathrm{d}t} + \vec{\omega}\frac{\mathrm{d}J_{i}}{\mathrm{d}t} = \sum_{j}\vec{M}_{ij}$$

- Fluid dynamic forces (drag, buoyant, Magnus, etc.) with empirical correlations
- Contact forces via spring-damper modell
- Multiple particle contacts possible



Euler-Lagrange – contact forces by DEM

(Cundall & Strack, 1979)





Euler-Lagrange model – coupling



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Euler-Euler model

- Public Domain Code MFIX (www.mfix.org)
- 1 gas phase
- 3 solid phases: wood (4mm), 2 x charcoal (1mm, 2mm)

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- 5 homogeneous gas phase reactions (rev.),
 - 4 heterogeneous reactions carbon
- Models for primary and secondary pyrolysis



Gas phase chemistry

Simple reaction mechanism

Species: H₂, O₂, N₂, CO, CO₂, CH₄, H₂O, tar

1.	2 CO	+	O2	\rightleftharpoons	2 CO ₂
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- $2. 2 H_2 + O_2 \rightleftharpoons 2 H_2 O$
- $3. \hspace{0.2cm} 2 \hspace{0.1cm} CH_4 \hspace{0.1cm} + \hspace{0.1cm} 3 \hspace{0.1cm} O_2 \hspace{0.1cm} \rightleftharpoons \hspace{0.1cm} 2 \hspace{0.1cm} CO \hspace{0.1cm} + \hspace{0.1cm} 4 \hspace{0.1cm} H_2O$
- $\label{eq:holescaled} \begin{array}{rccccccc} 4. & CO & + & H_2O & \rightleftharpoons & H_2 & & + & CO_2 \end{array}$
- 5. CH_4 + H_2O \rightleftharpoons CO + $3H_2$



Pyrolysis / heterogeneous reactions

Primary pyrolysis (Grönli, Melan (2000))

 $wood \rightarrow H_2O(g), \ CO(g), \ CO_2(g), \ H_2(g), \ CH_4(g), \ tar(g), \ char(s).$

Secondary pyrolysis (Boroson et al. (1989))

$$tar \rightarrow CO(g), \ CO_2(g), \ H_2(g), \ CH_4(g), \ tar_{inert}.$$

Heterogeneous reactions

- $C + O_2 \rightarrow CO_2$
- $C + CO_2 \rightarrow 2CO$

 $C + 2H_2 \rightarrow CH_4$

$$C \hspace{0.2cm} + \hspace{0.2cm} H_2O \hspace{0.2cm} \rightarrow \hspace{0.2cm} CO \hspace{0.2cm} + \hspace{0.2cm} H_2$$

(Ross and Davidson(1982)) (Biggs and Agarwal(1997)) (Hobbs et al. (1992))

(Hobbs et al. (1992))

Reactor at the department



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air

Reactor at the department







Reactor - inlet boundaries



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Results - Euler-Euler model



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Results – exhaust gas composition



Image: A matrix



Results - Euler-Euler model





Euler-Lagrange / DEM - results

Model assumptions:

- proportional particle massloss and shrinking
- transient calculation of particle mass-loss and energy

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- no heterogeneous reaction (in progress)
- no particle-particle heat transfer

no radiation



Euler-Lagrange / DEM - results





Summary and outlook

- Euler-Lagrange and Euler-Euler model for wood gasification in fluidized beds
- Euler-Euler: reasonable agreement with experimental data
- Euler-Lagrange: allows detailed investigation of particle/chemistry/gas phase interactions
- Refined chemistry
- Parallelization of the DEM modell
- Quantitative comparison between Euler-Euler, Euler-Lagrange/DEM und experiments
- Coupling Euler-Lagrange with Euler-Euler methods
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Governing equations – Euler-Euler model

Species mass balance gas phase

$$\frac{\partial}{\partial t}(\epsilon_g \rho_g Y_{gs}) + \nabla \cdot (\epsilon_g \rho_g \vec{v}_g Y_{gs}) = \nabla \cdot (D_{gs} \nabla Y_{gs}) + R_{gs}$$

Mass balance solid phase

$$\frac{\partial}{\partial t}(\epsilon_m \rho_m Y_{ms}) + \nabla \cdot (\epsilon_m \rho_m \vec{v}_{sm} Y_{ms}) = \nabla \cdot (D_{ms} \nabla Y_{ms}) + R_{ms}$$

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Governing equations – Euler-Euler model

Momentum balance solid phase

$$\frac{\partial}{\partial t}(\epsilon_m\rho_m\vec{v}_m) + \nabla \cdot (\epsilon_m\rho_m\vec{v}_m\vec{v}_m) = -\varepsilon_m\nabla\rho_m + \nabla \cdot \overline{\overline{\tau}}_m - \sum_{\substack{l=1\\l\neq m}}^{n_m}\vec{l}_{ml} + \vec{l}_{gm} + \epsilon_m\rho_m\vec{g}$$

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Governing equations – Euler-Euler model

Energy balance gas phase

$$\epsilon_{g}\rho_{g}c_{\rho g}\left(\frac{\partial T_{g}}{\partial t}+\vec{v}_{g}\cdot\nabla T_{g}\right)=-\nabla\cdot\vec{q}_{g}+\sum_{m=1}^{N_{m}}\gamma_{gm}\left(T_{m}-T_{g}\right)-\Delta H_{g}$$

Energy balance solid phases

$$\epsilon_m \rho_m \boldsymbol{c}_{pm} \left(\frac{\partial T_m}{\partial t} + \vec{\boldsymbol{v}}_m \cdot \nabla T_m \right) = -\nabla \cdot \vec{\boldsymbol{q}}_m - \gamma_{gm} \left(T_m - T_g \right) - \Delta H_m$$

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