

The Contribution of FBC to Optimized Resource and Waste Management

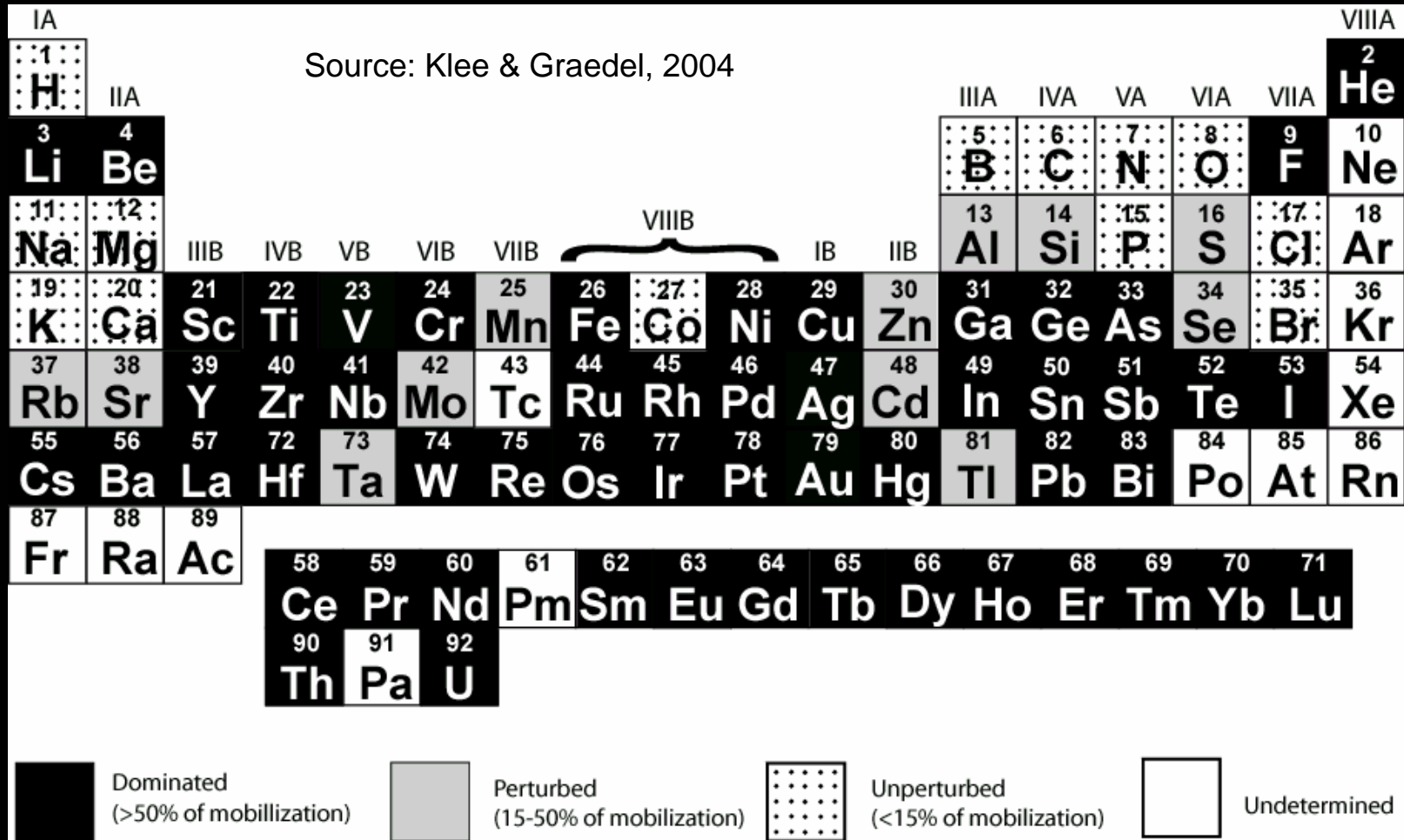
Future Challenges for Waste Combustion and Co-combustion in FBC
May 24 2004, Vienna



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Anthropogenic vs. geogenic mobilization

Source: Klee & Graedel, 2004

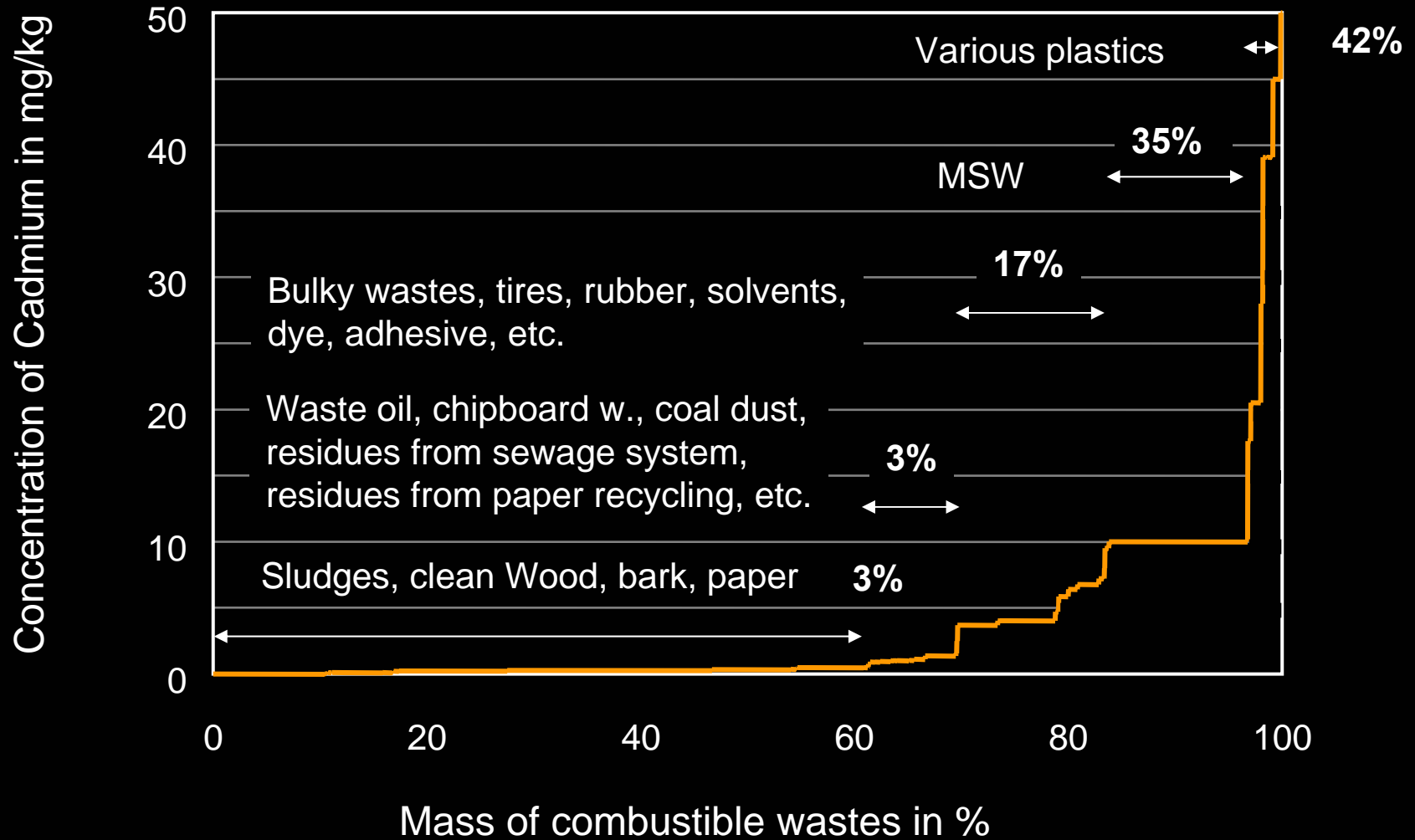


Generation of combustible wastes

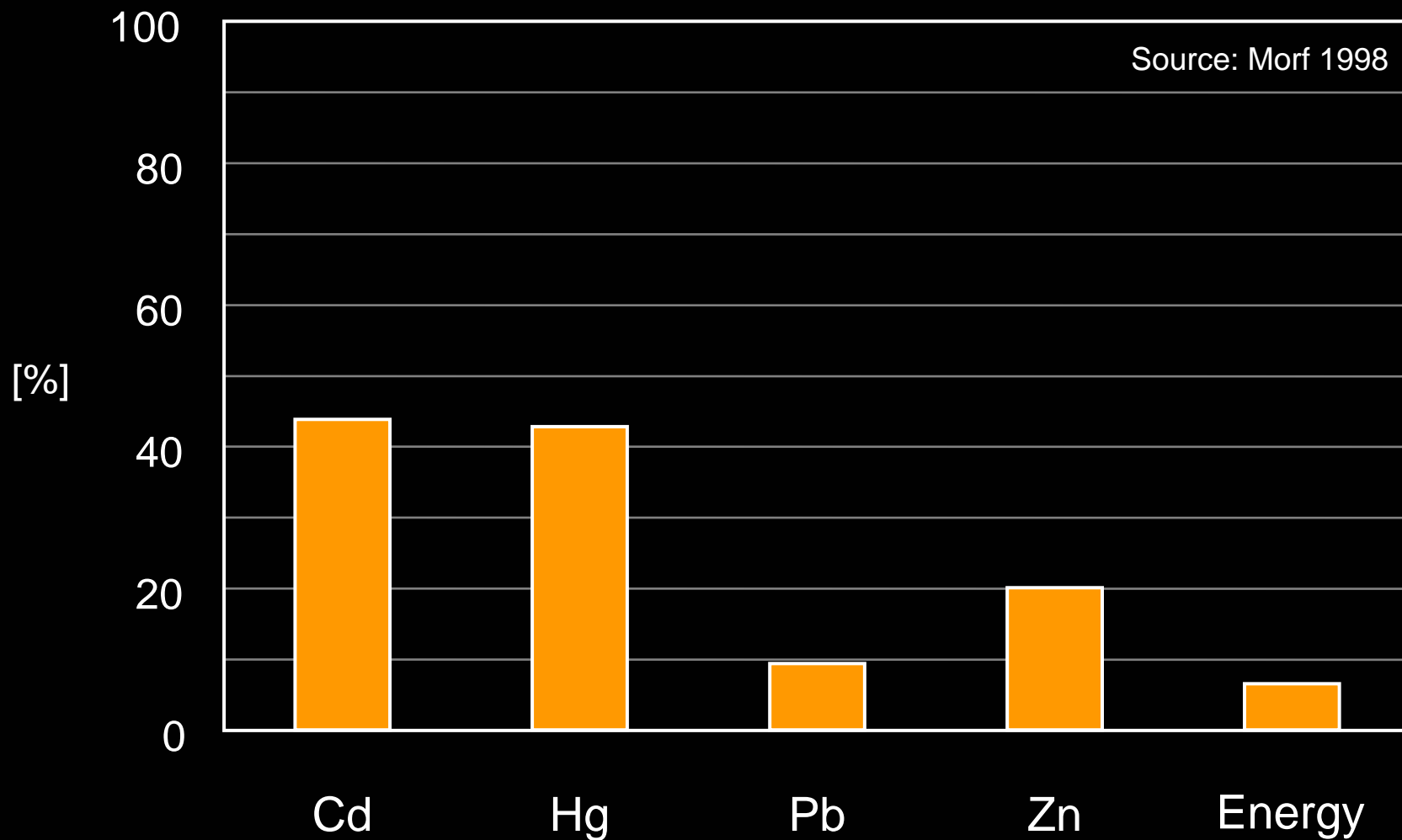
Waste categories	t/(c.yr)	% combustible	t/(c.yr)
• Municipal solid waste	0.31	87	0.27
• Construction & demolition waste	3.00	2	0.06
• Wastes of water purification & waste water treatment (34%DM)	0.29	41	0.12
• Wood & timber wastes	0.44	100	0.44
• Industrial wastes	1.00	14	0.14
• Hazardous wastes	0.14	22	0.03
Total	5.20	22	1.10

Source: Fehringer et al. 1997

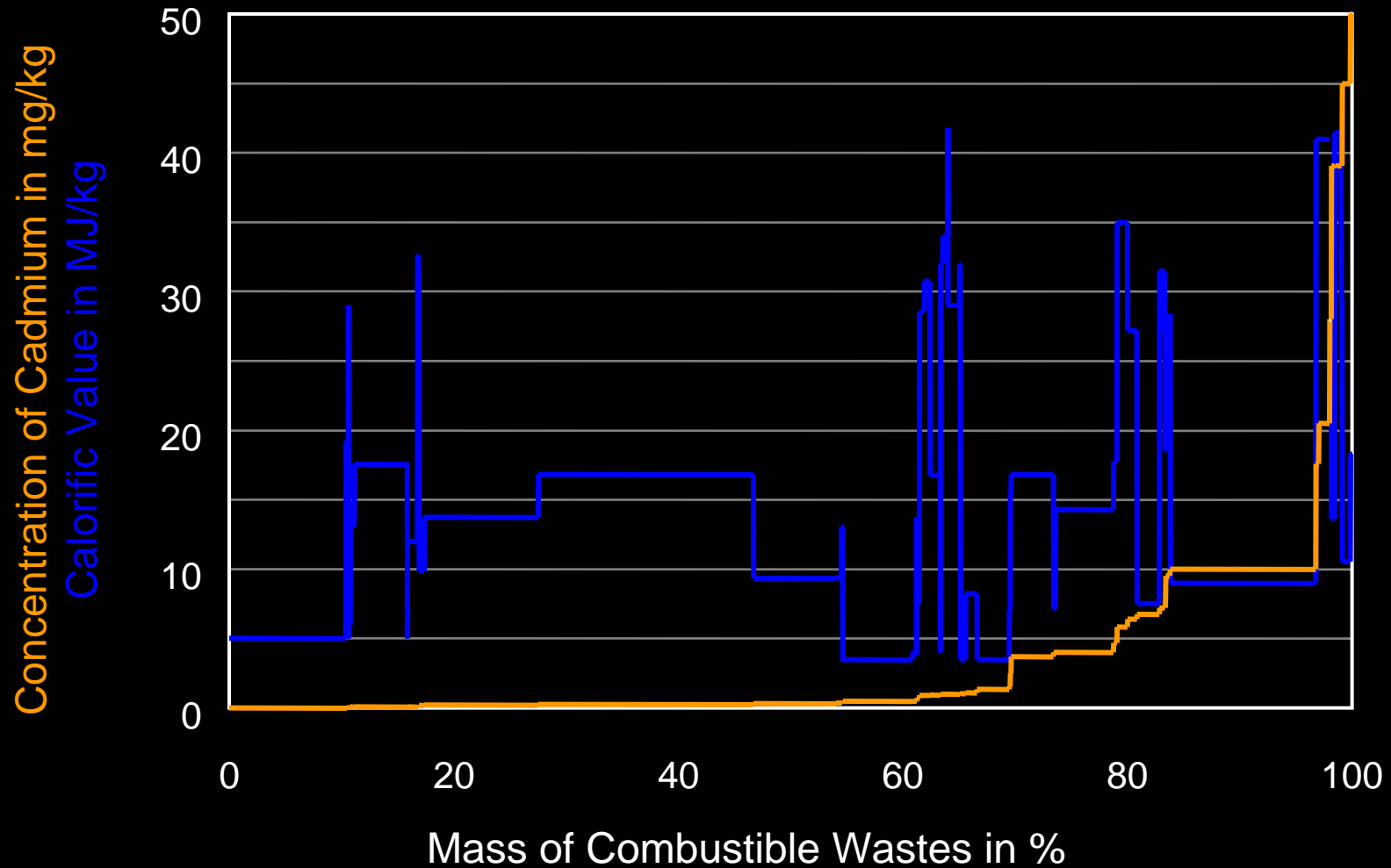
Contamination by Cadmium



Relative flow of HM via combustible wastes (100% = national consumption)



No correlation between HM and calorific value



Introduction to Statistical Entropy Analysis (SEA)

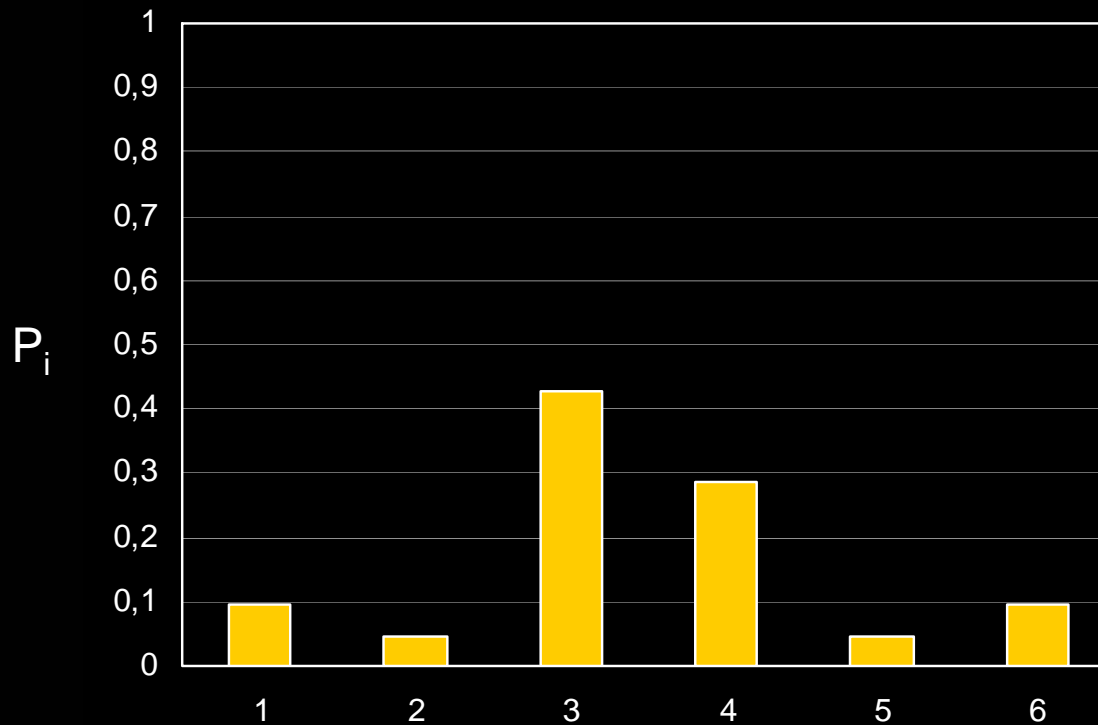
- **Statistical entropy quantifies distribution patterns (statistics)**
- **Statistical entropy \neq thermodynamic entropy**
- **SEA is published**

Rechberger, H., Brunner, H.P. A new, entropy based method to support waste and resource management decisions. *Environmental Science and Technology*, 2002, Vol. 34, No. 4, 809-816.

Rechberger, H., Graedel, T.E., The European copper cycle: statistical entropy analysis. *Ecological Economics*, 2002, Vol. 42, No. 1-2, 59-72.

Distribution patterns: statistics

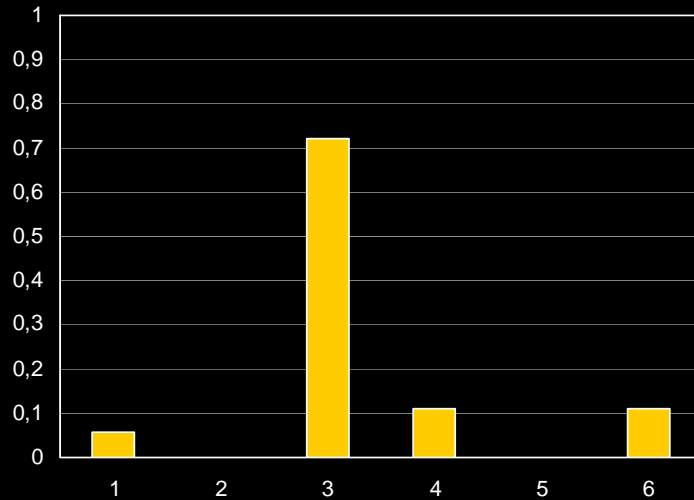
$$H_{\text{rel}} = - \sum_{i=1}^k (P_i \log_2(P_i)) / \log_2(k)$$



$$0 < H_{\text{rel}} < 1$$

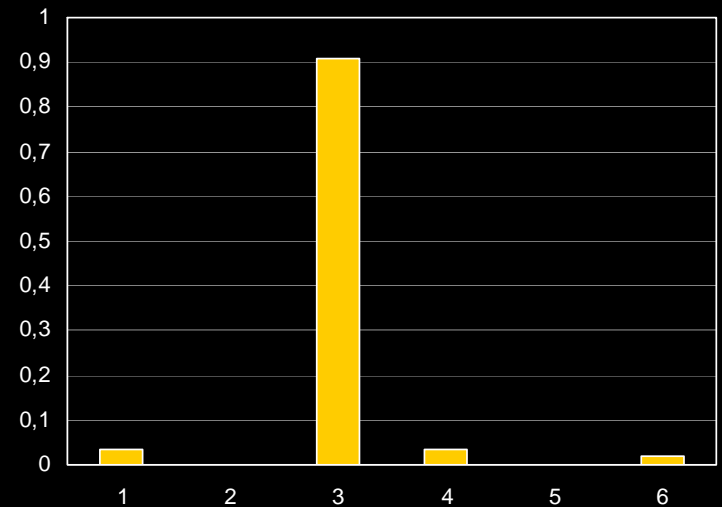
$$H_{\text{rel}} = 0.81$$

Distribution patterns: statistics

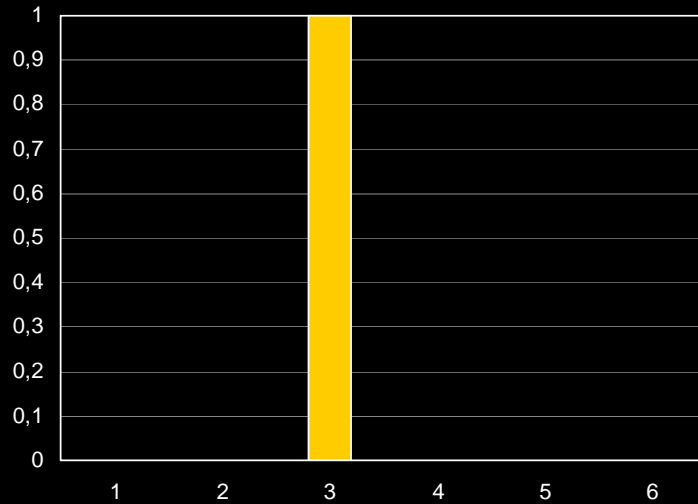


$$H_{rel} = 0.49$$

$$H_{rel} = 0.22$$

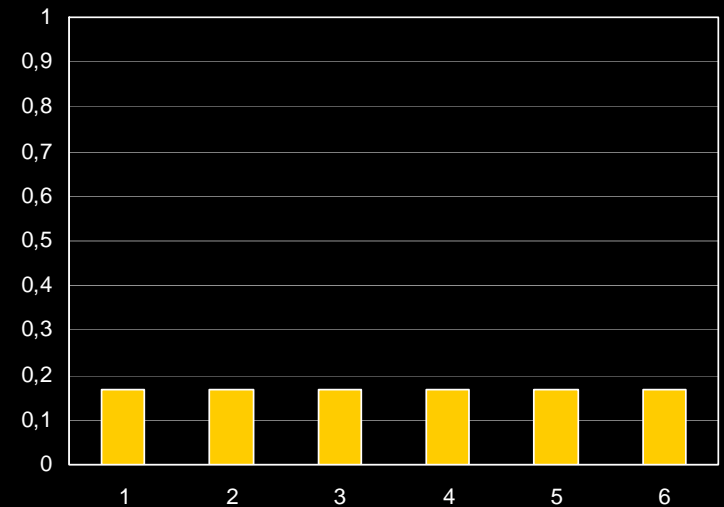


Distribution patterns: statistics



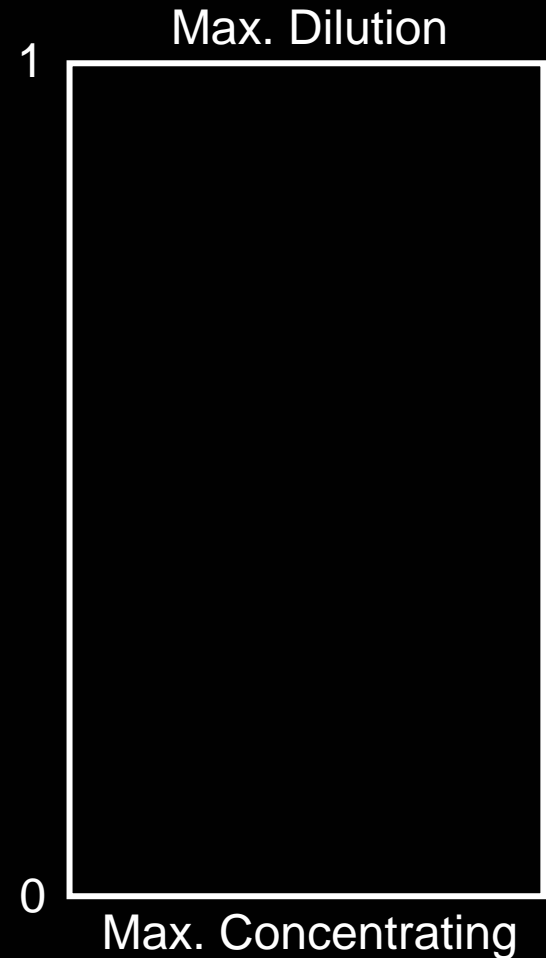
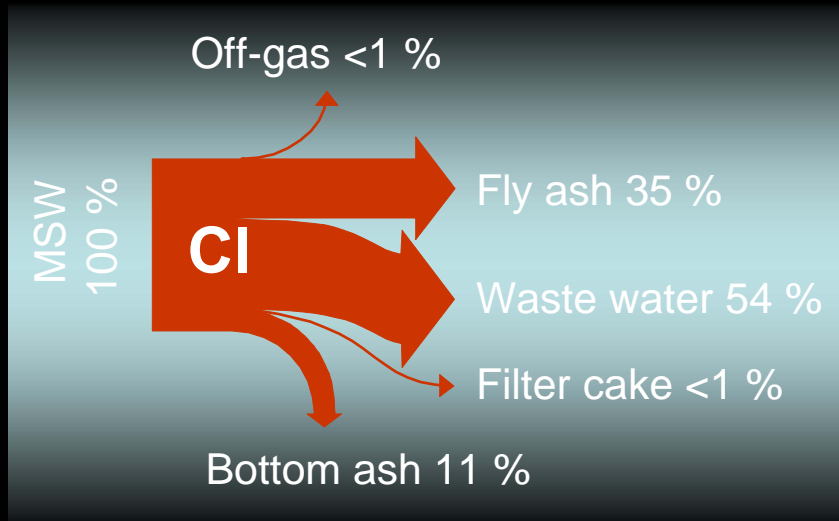
$H_{rel} = 0$
Minimum

$H_{rel} = 1$
Maximum



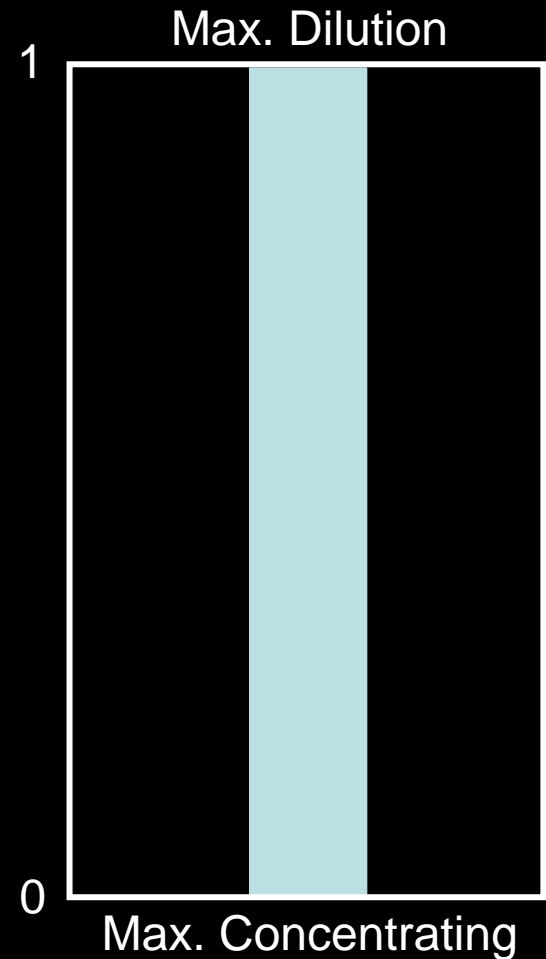
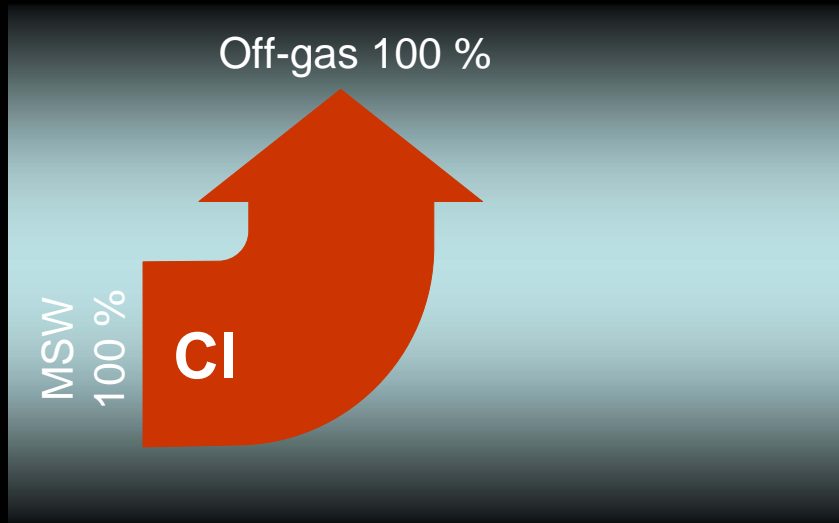
Distribution patterns: combustion

Example Chlorine



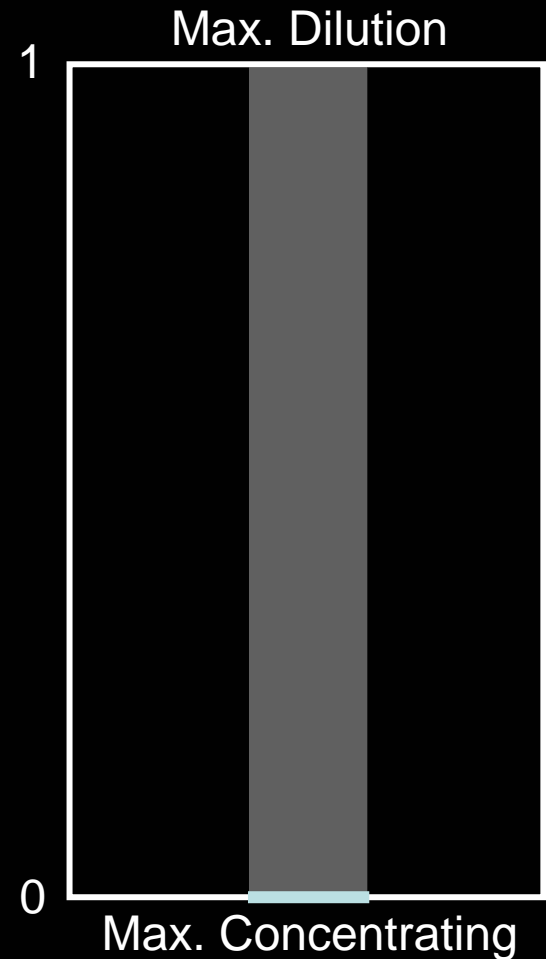
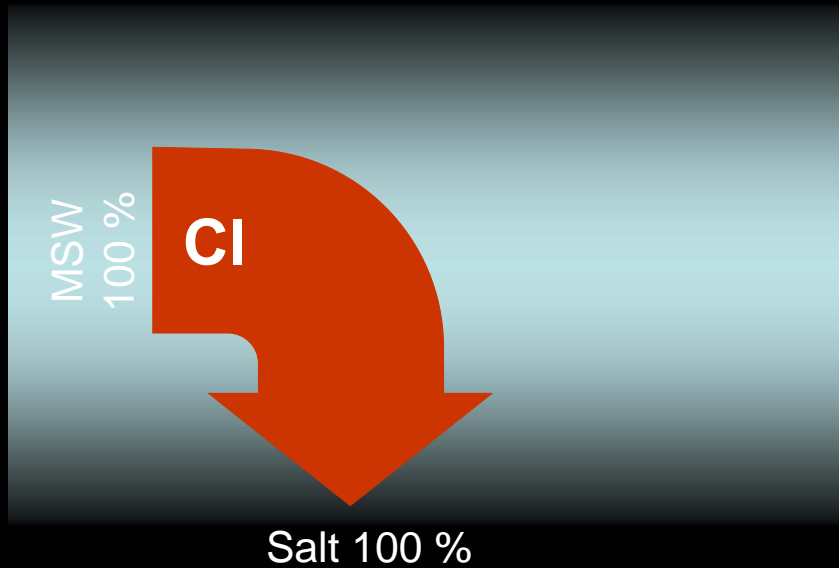
Distribution patterns: combustion

Example Chlorine



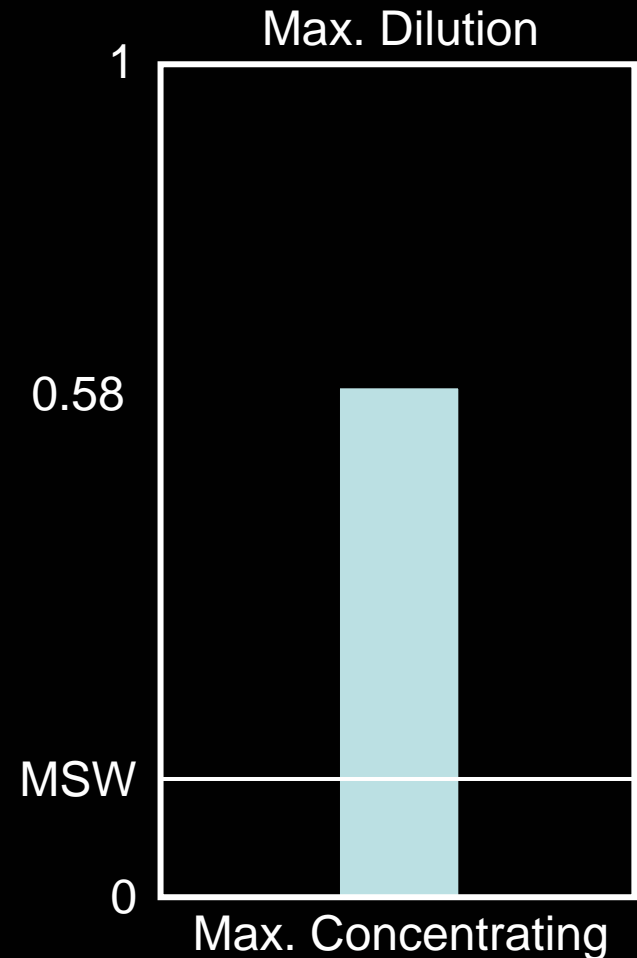
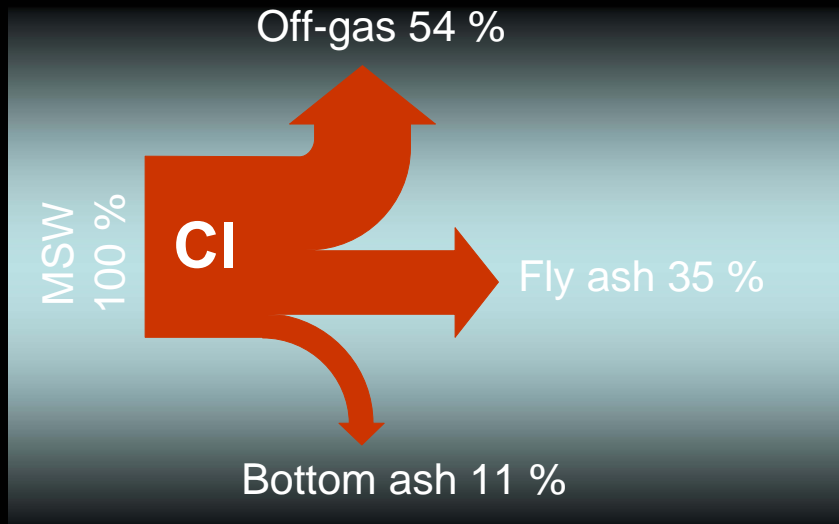
Distribution patterns: combustion

Example Chlorine



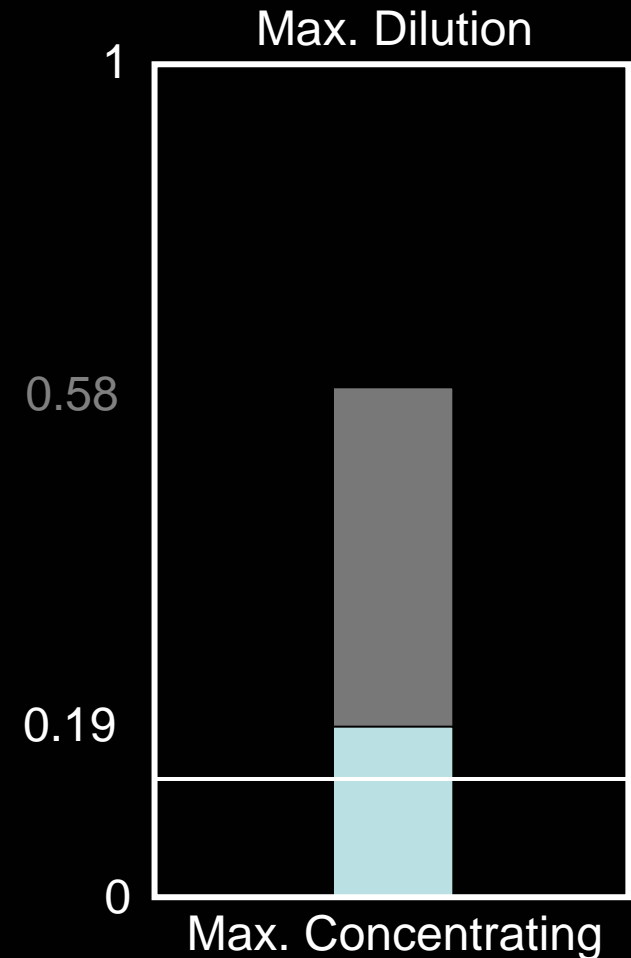
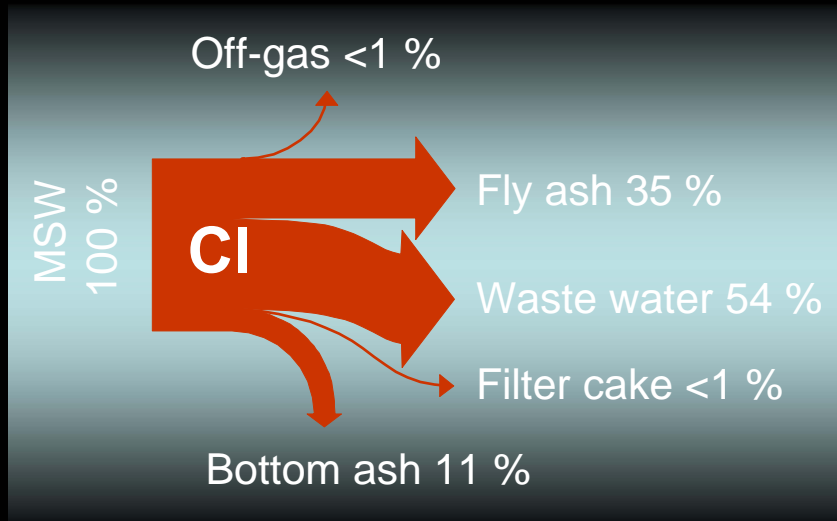
Distribution patterns: combustion

Example Chlorine



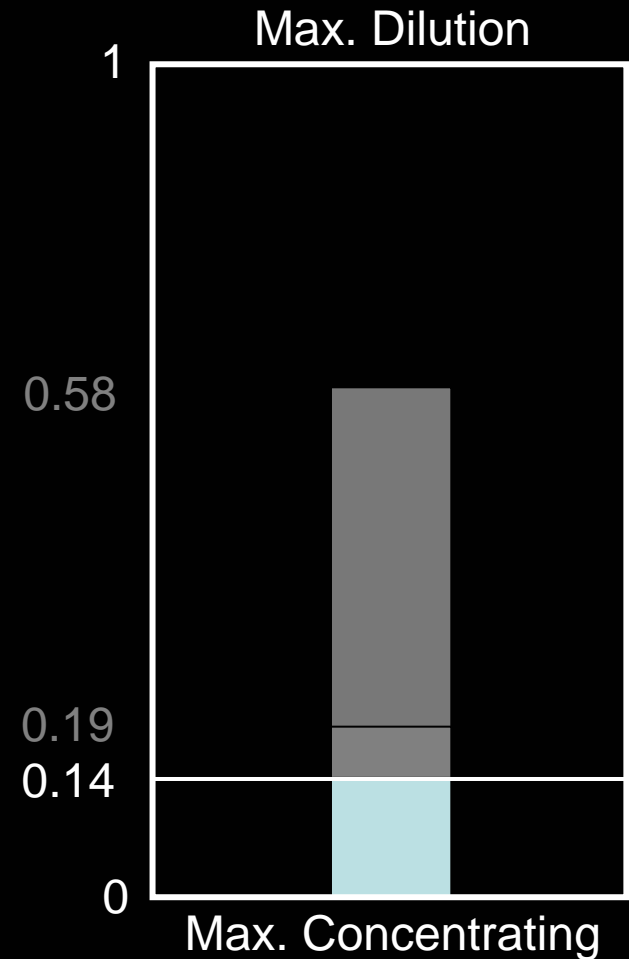
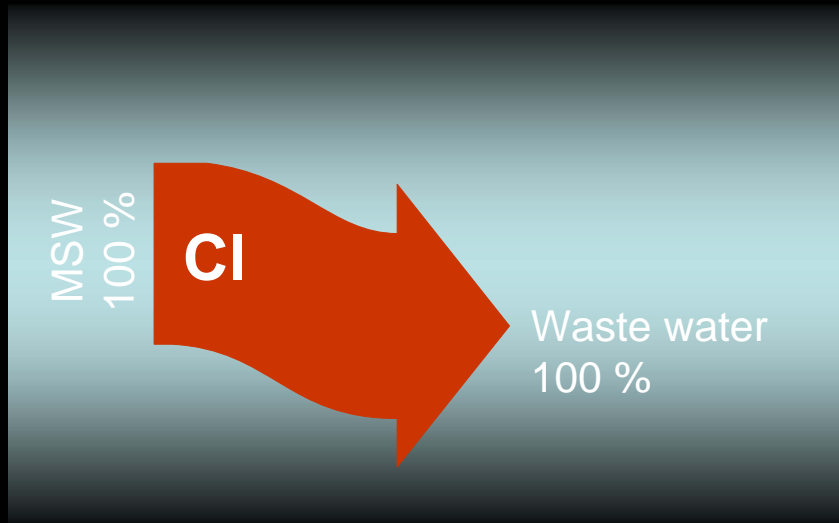
Distribution patterns: combustion

Example Chlorine



Distribution patterns: combustion

Example Chlorine



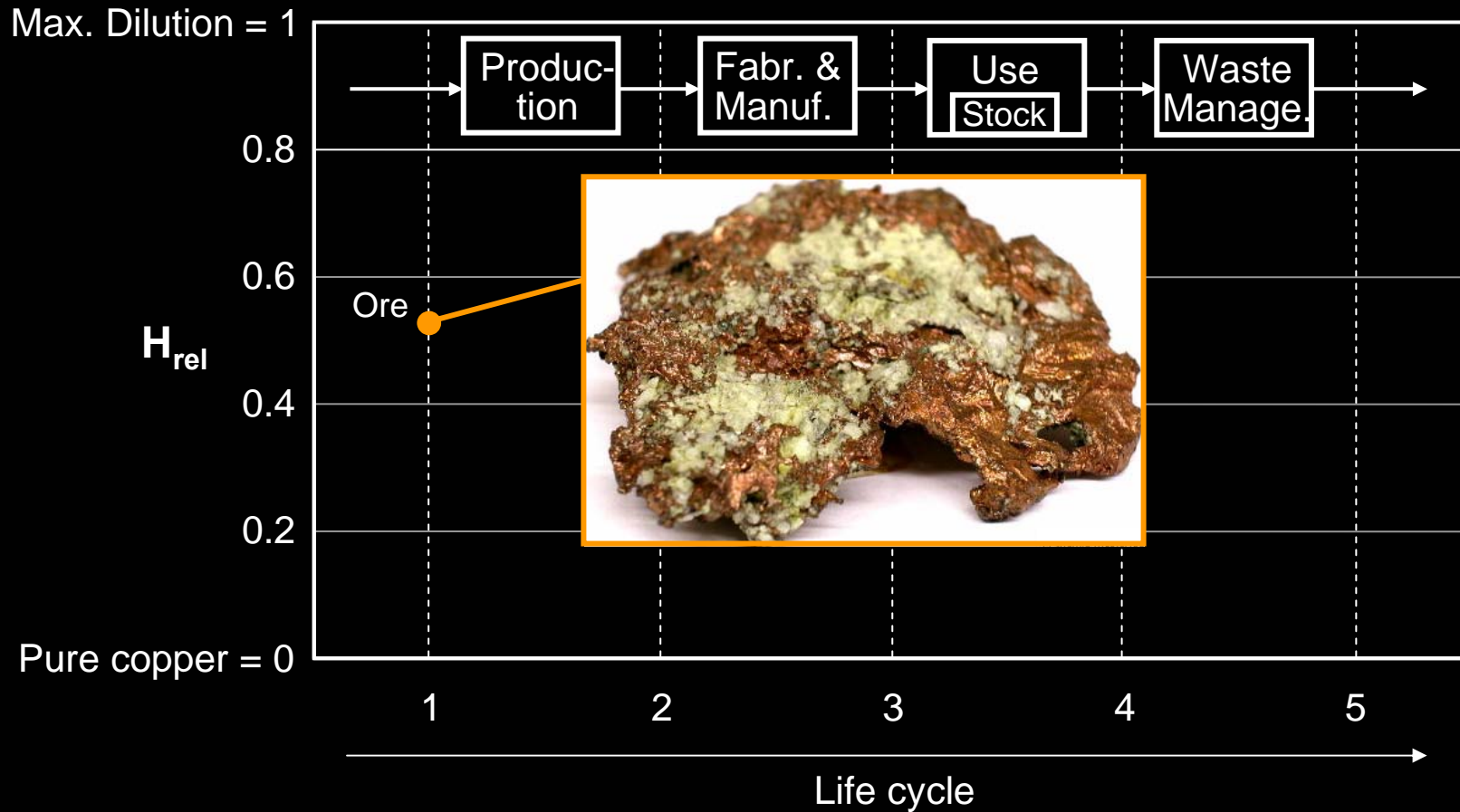
SEA of global cop

Max. Dilution = 1



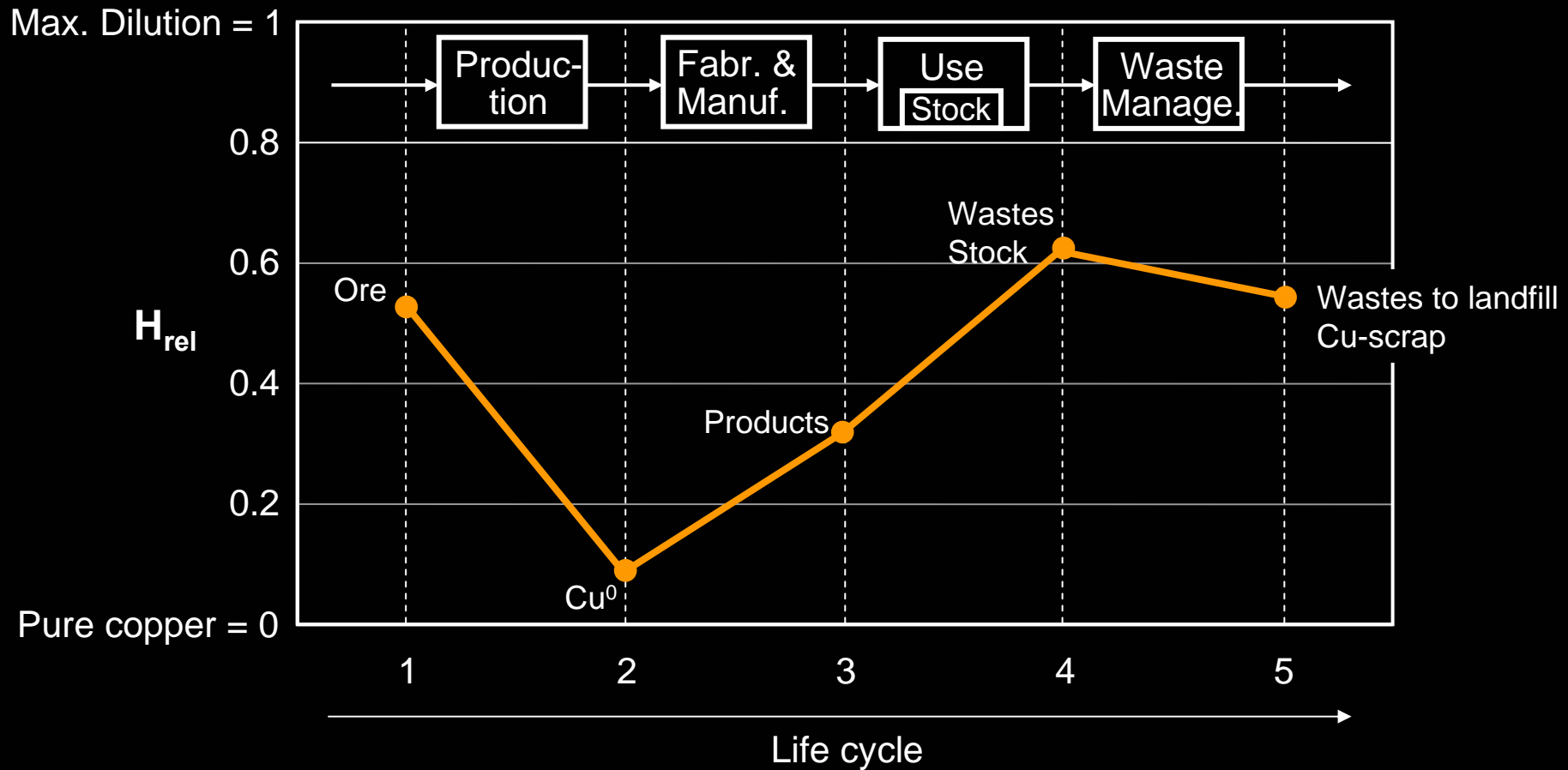
Source: Rechberger, 2003

SEA of global copper flows



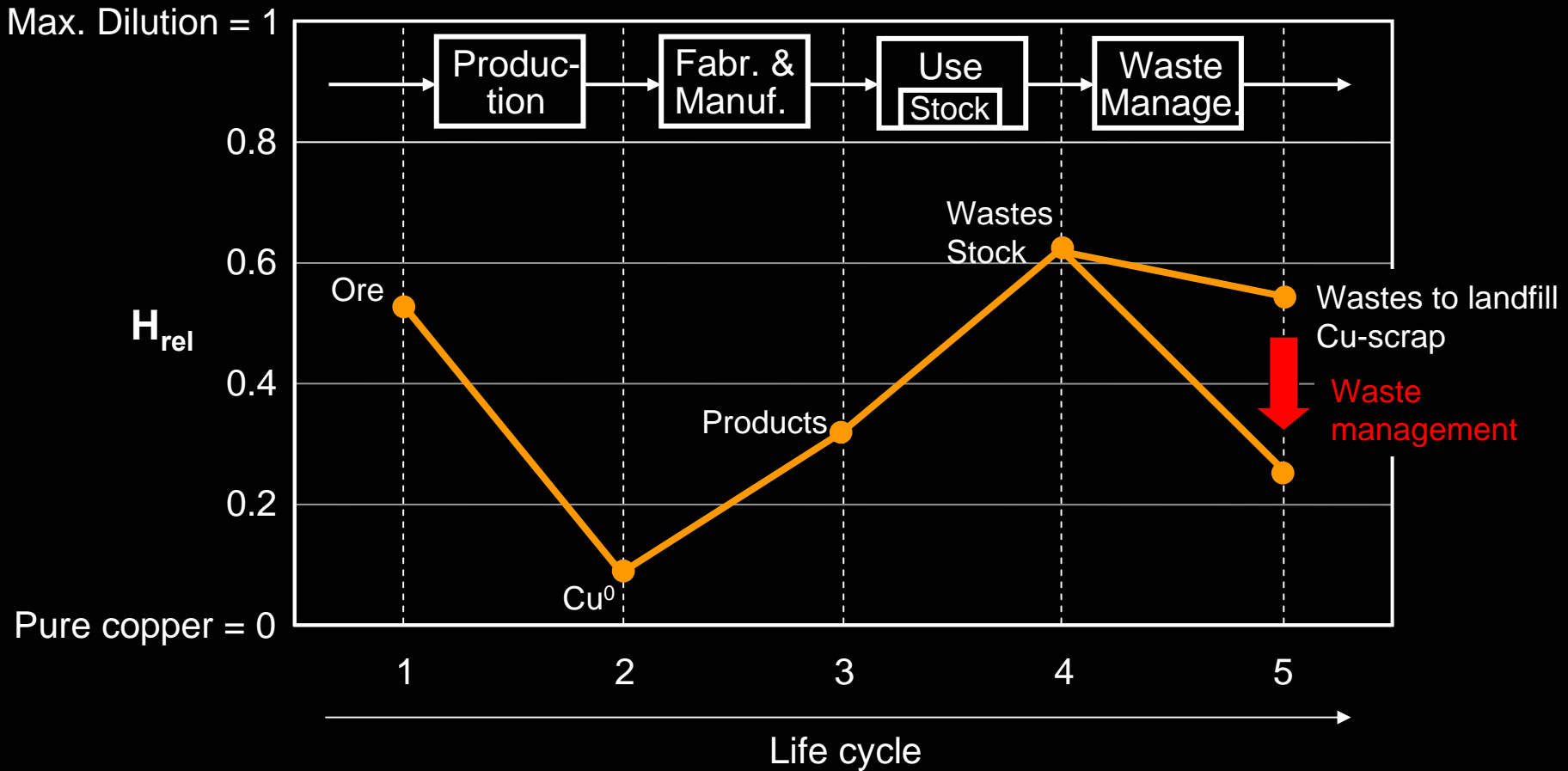
Source: Rechberger, 2003

SEA of global copper flows



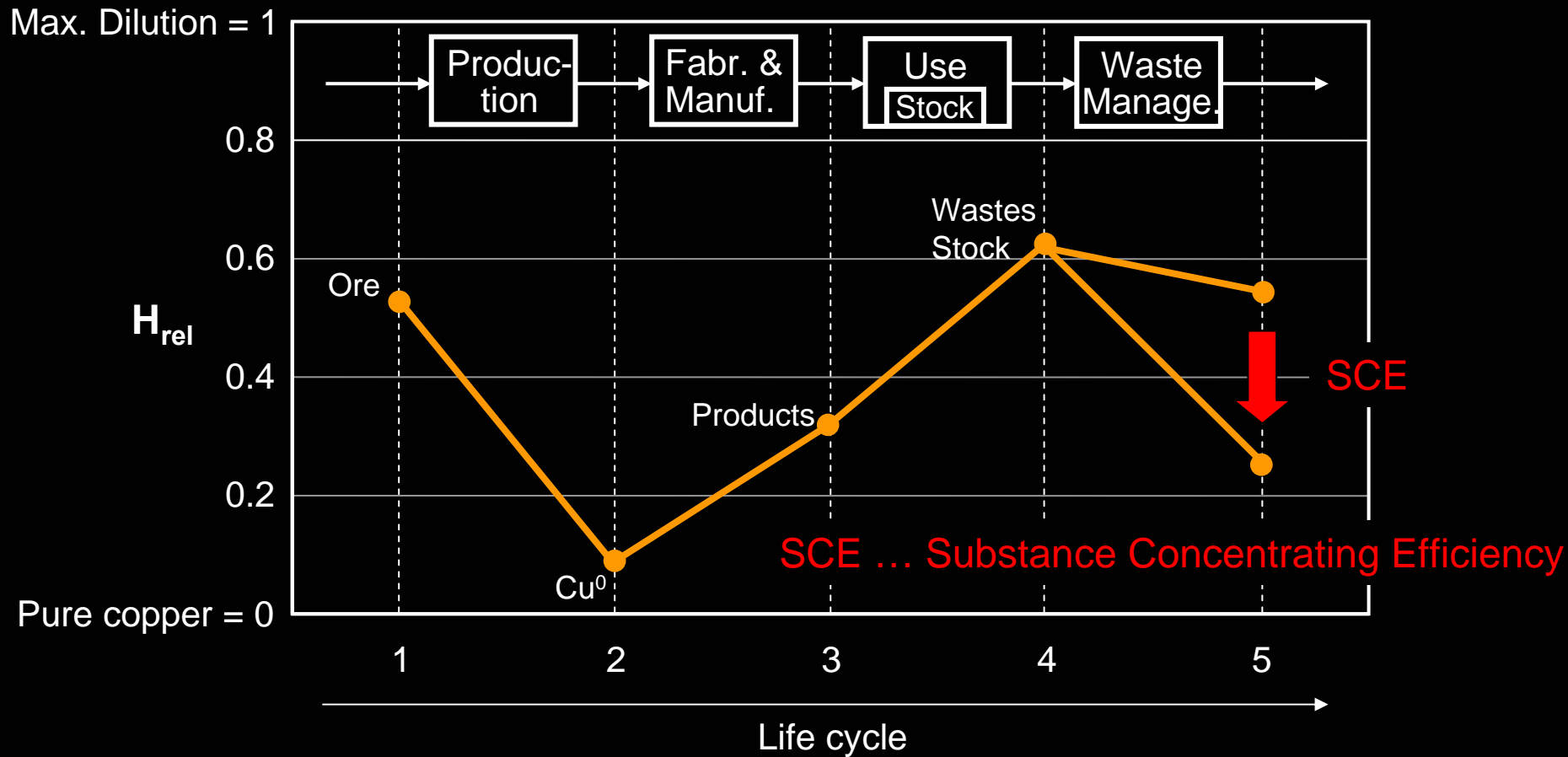
Source: Rechberger, 2003

SEA of global copper flows



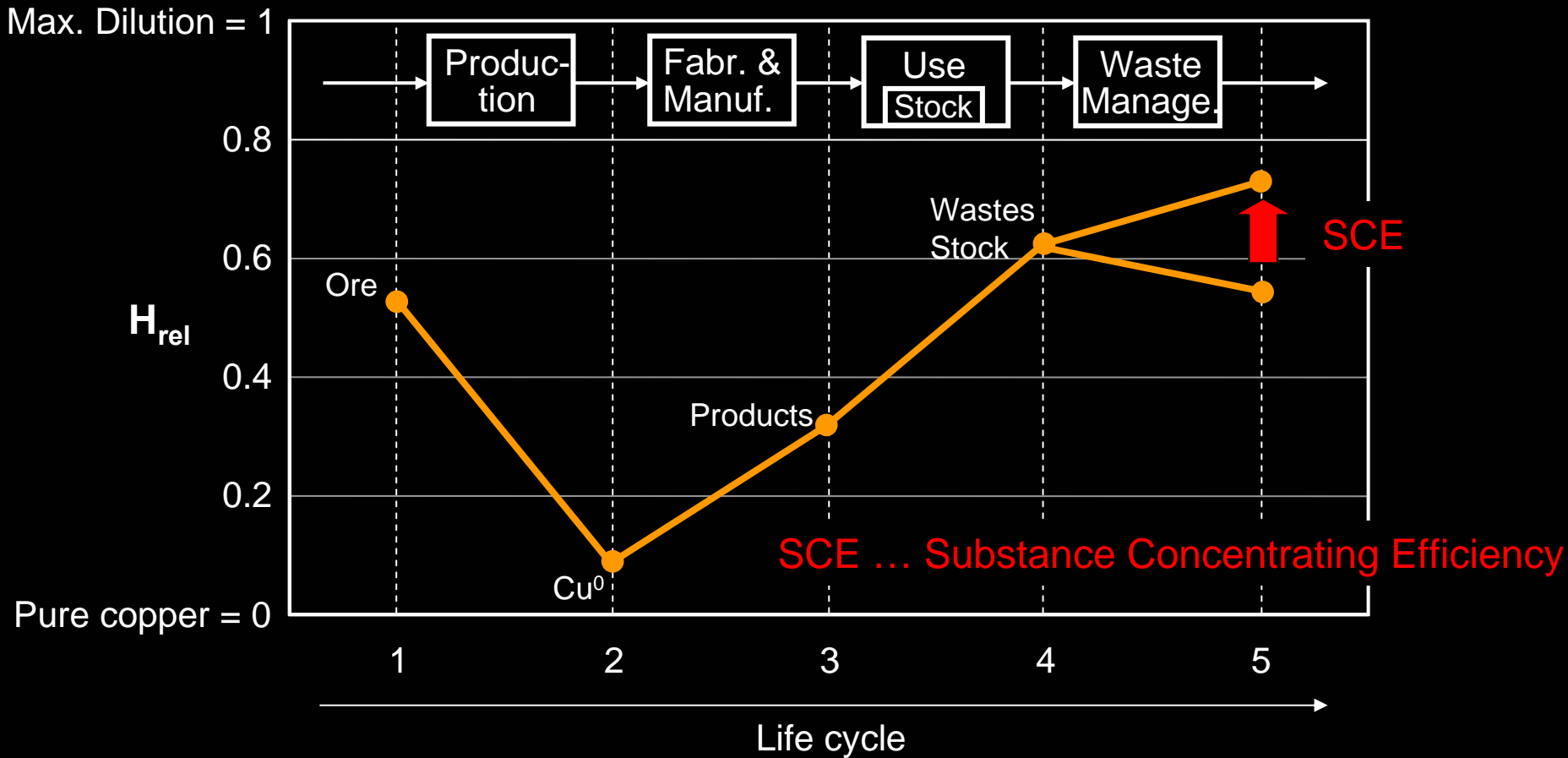
Source: Rechberger, 2003

SEA of global copper flows



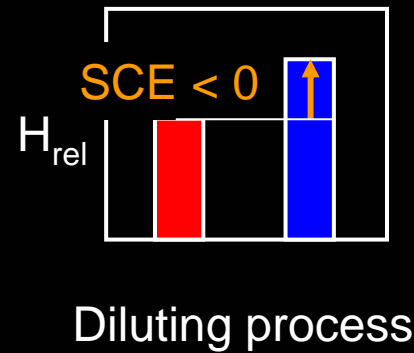
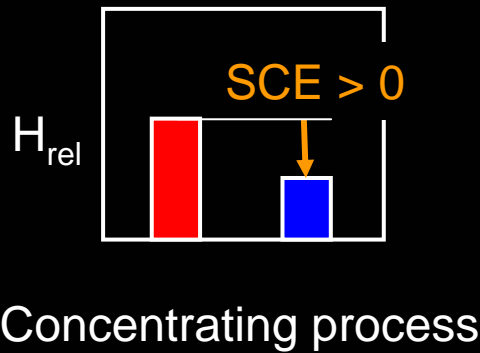
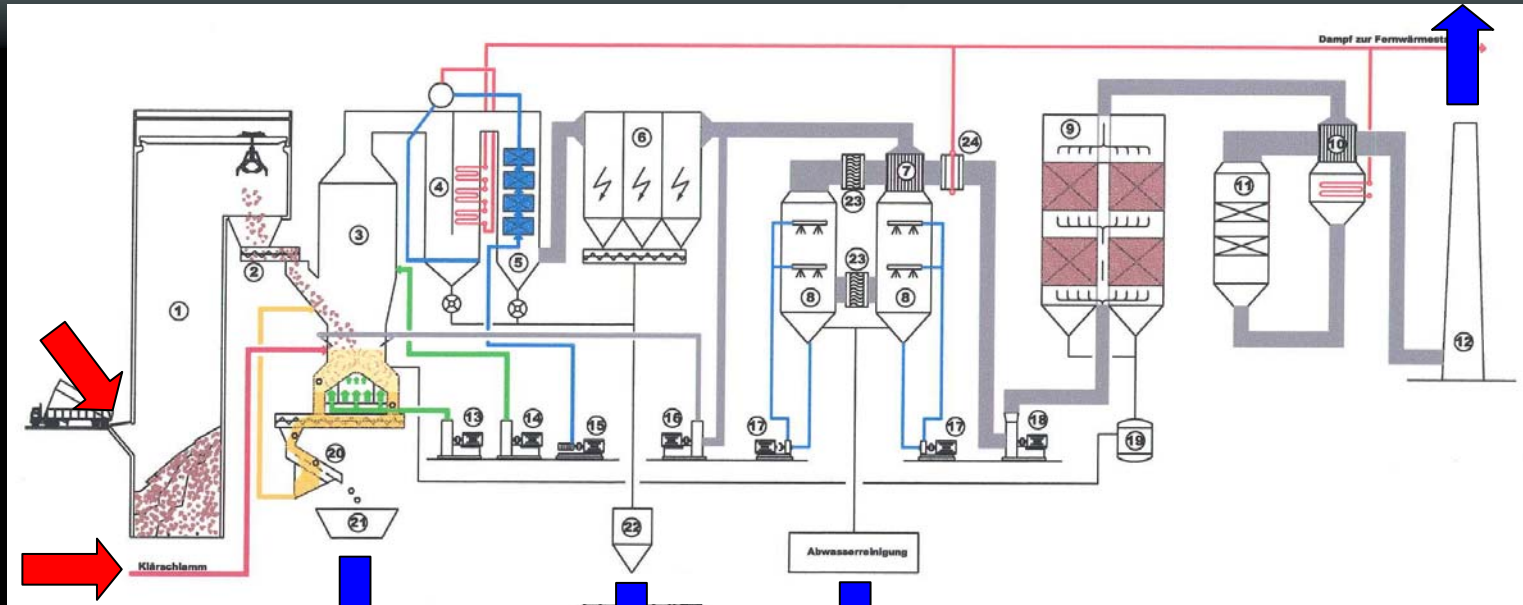
Source: Rechberger, 2003

SEA of global copper flows



Source: Rechberger, 2003

Substance concentrating efficiency (SCE)



Requirements for high SCE-values

- **Low emissions to the environment**
- **High-concentrated residues (low volume)**
- **Earth-crust like residues**

Benefit of the indicator SCE

- **Quality of substance mgmt. is quantified**
- **Comparability of technologies**
- **Indicator for efficiency**

Efficiency of technologies

