

Introduction

Fluidized Bed Combustion

at the

Dresden University of Technology

Co-combustion

A. HILLER, N. T. QUANG

Investigation (with) system

TU Dresden Pilot Rig

Co-fuels

Mathematical modelling

Calculations for optimising of combustion

- **Secondary air distribution**
- **Mixture proportions**
- **Charge spots of Co-fuels**

Modelling Data sheets 1D

Experimental Profiles 2D

Outlook





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Test Rig Circulating Fluidized Bed II

System height: 6,7 m

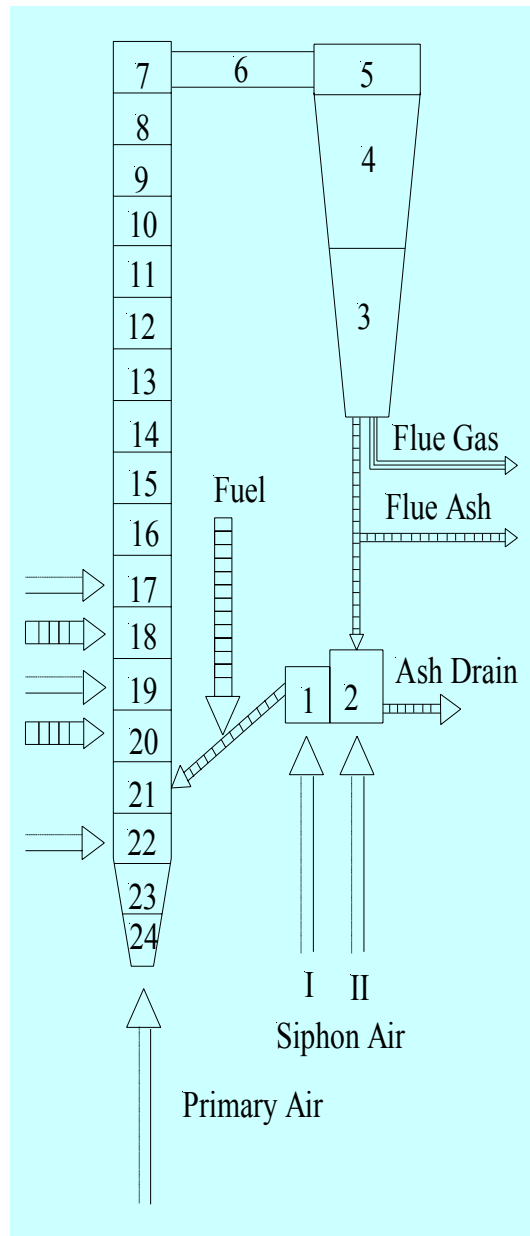
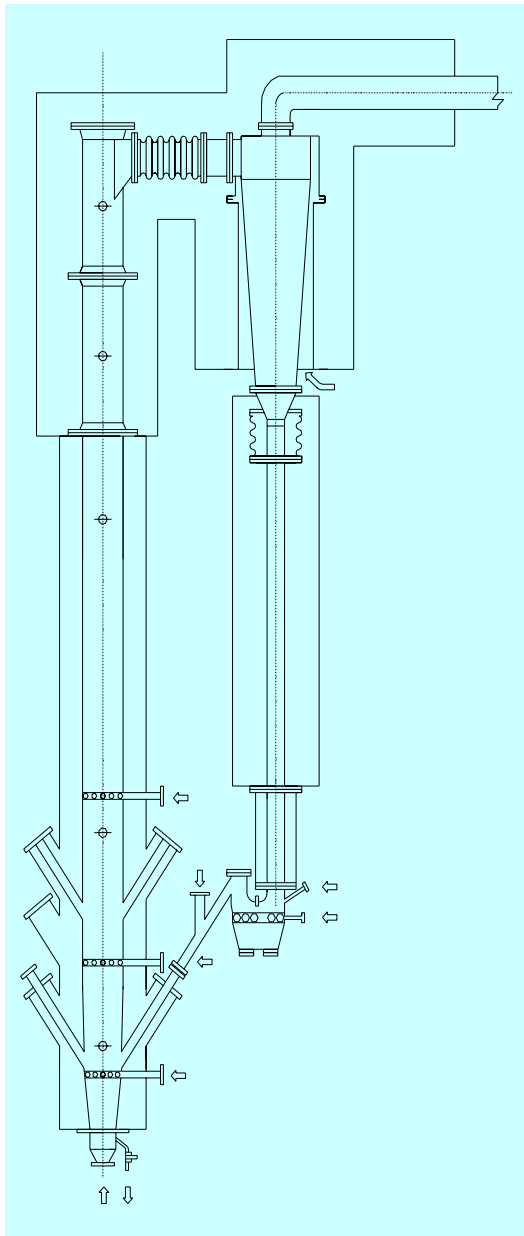
ø: 0,32 m

Sectional view

24 Elements

thermal Load: 300 kW

Cell Distribution for Modelling



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Outline of substances, processes and reactions considered in the simulation model

Air O₂ + N₂

Fuels: Coal + Waste

Elementary Analysis: C,H,O,N,S,A,W
 Caloric Value: H_u; H_{ul}
 Particle Size Distribution: p₀(d_C); p_{0a}(d_{Ca})

Limestone

Elementary Analysis
 CaCO₃, H₂O, Minerals
 Particle Size Distribution: p_{0k}(d_{Ck})

Drying Process (Water Outlet, kinetic)

Dry Fuels + Water

Calcination

CaO, CO₂, H₂O, Ash

Degassing (kinetic)

Coke
 CO, CO₂, H₂, CH₄, C_xH_y, H₂O, NH₃, HCN, H₂S

Particle Size Modification

(Contraction, Fragmenting, Attrition, Burning Rate)
 p₀(d_C) → p₁(d_{C1}); p_{0a}(d_{Ca}) → p_{1a}(d_{Ca1})

Particle Size Modification

(Abrasion)
 p_{0k}(d_{Ck}) → p_{1k}(d_{C1k})

Combustion Reactions (stoichiometrically)

1 H₂ + 0,5 O₂ → 1 H₂O
 1 H₂S + 1,5 O₂ → 1 H₂O + 1 SO₂
 1 CH₄ + 1,5 O₂ → CO + 2 H₂O
 1 HCN + 1,25 O₂ → NO + CO + 0,5 H₂O

Sulphation(kinetic, Chemical Balance)

1 SO₂ + 0,5 O₂ → 1 SO₃
 1 CaO + 1 SO₃ → 1 CaSO₄

Combustion Reactions (kinetic)

1 C + 1/φ O₂ → (2-2/φ) CO + (2/φ-1) CO₂
 1 CO + 0,5 O₂ → 1 CO₂
 1 NH₃ + 1,25 O₂ → 1 NO + 1,5 H₂O
 1 NH₃ + 0,75 O₂ → 0,5 N₂ + 1,5 H₂O
 1 NO + 1 CO → 0,5 N₂ + 1 CO₂
 1 NO + 2/3 NH₃ → 5/6 N₂ + 1 H₂O
 1 NO + 1 H₂ → 0,5 N₂ + 1 H₂O
 1 NO + 1 C → 0,5 N₂ + 1 CO
 1 C_xH_y + (0,5x+y/4) O₂ → x CO + y/2 H₂O

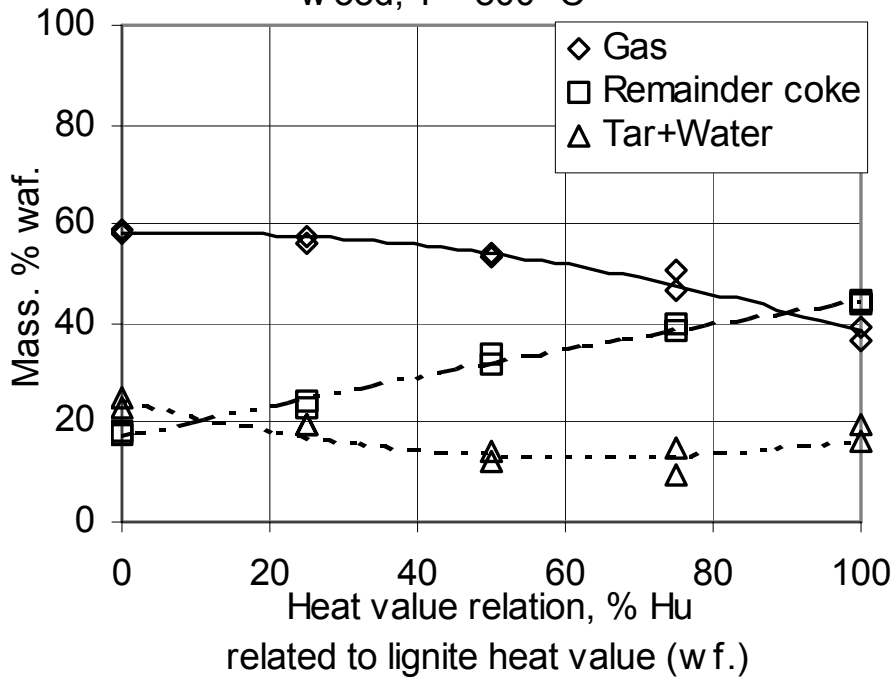


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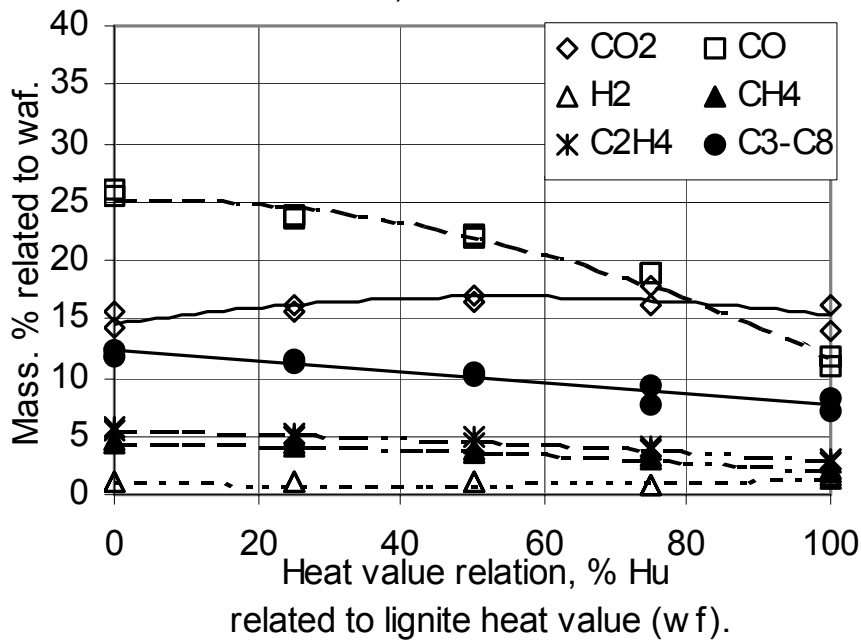
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Mass balance, pyrolysis of lignite and wood, $T = 800\text{ }^{\circ}\text{C}$



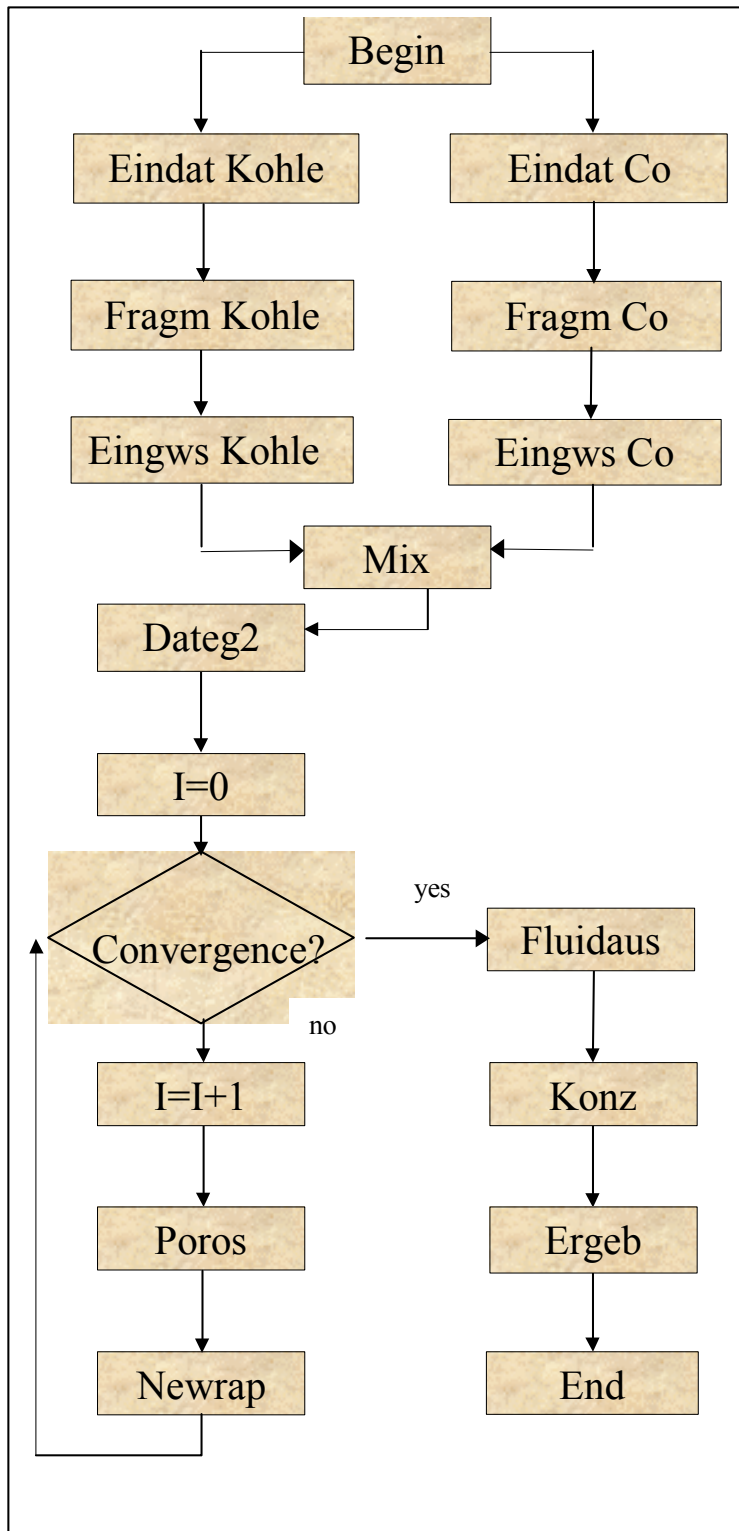
Gas proportions, pyrolysis of lignite and wood, $T = 800\text{ }^{\circ}\text{C}$



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Eindat

Sub for reading input data

Fragm

Sub for calculation of particle distribution in the feed for both fuels

Eingws

Sub for converting and precalculation of degassing, solid distribution and solid flow

Mix

Precalculation of mixture

Dateg2

Output of data

Poros

Sub for calculation of porosity along the reactor

Konz

Calculation of gaseous flows and concentrations for all elements (cells)

Newrap

Sub for Iteration of the nonlinear set of equations consist of 22 variables for 24 elements (cells)

Fluidaus

Output of the flowmechanical calculation

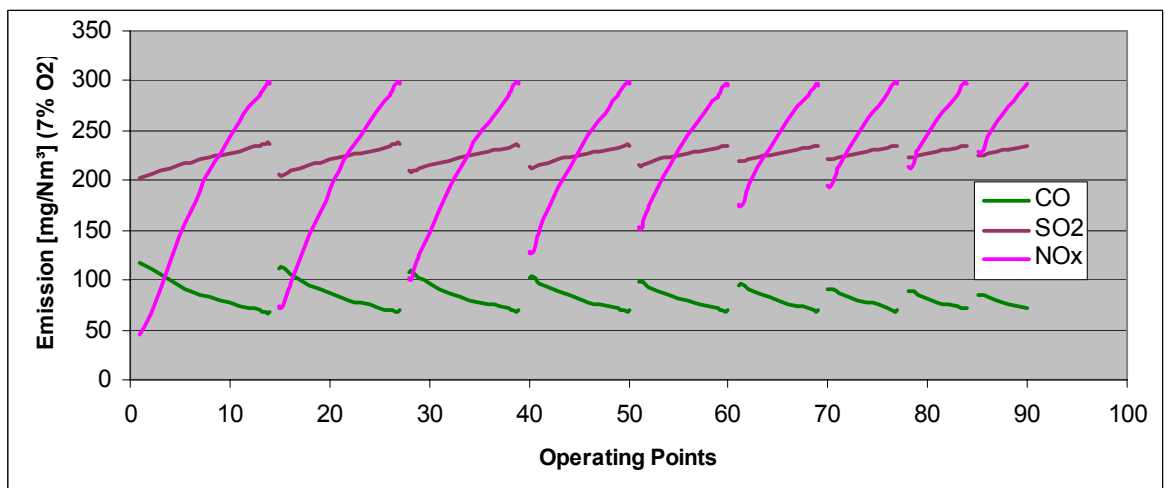
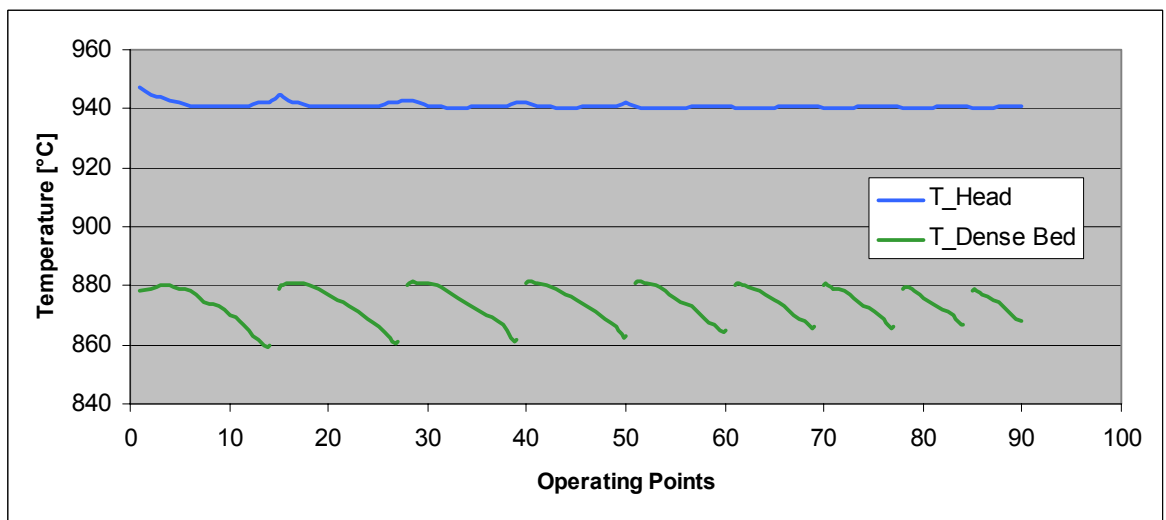
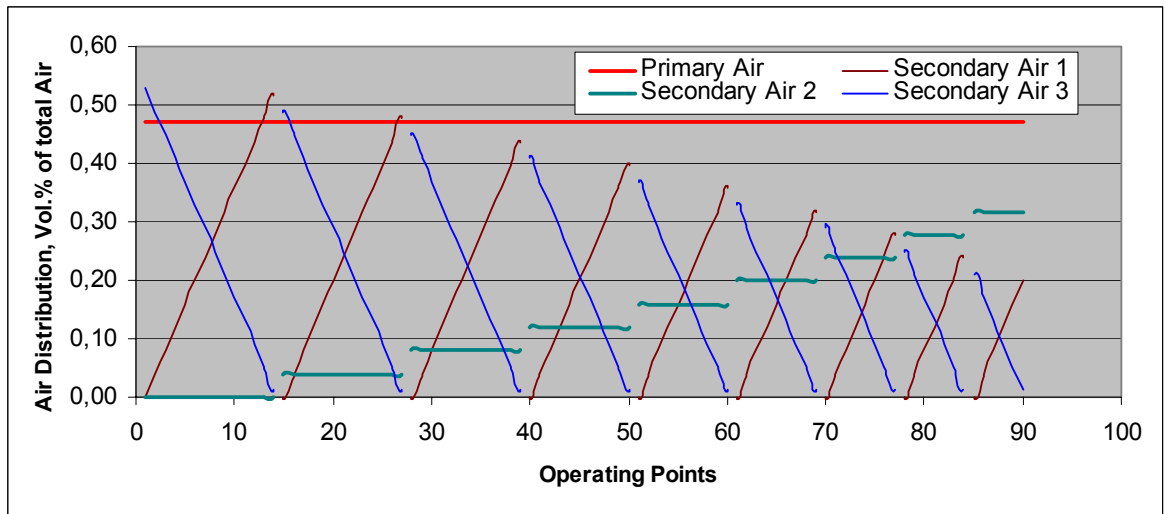
Ergeb



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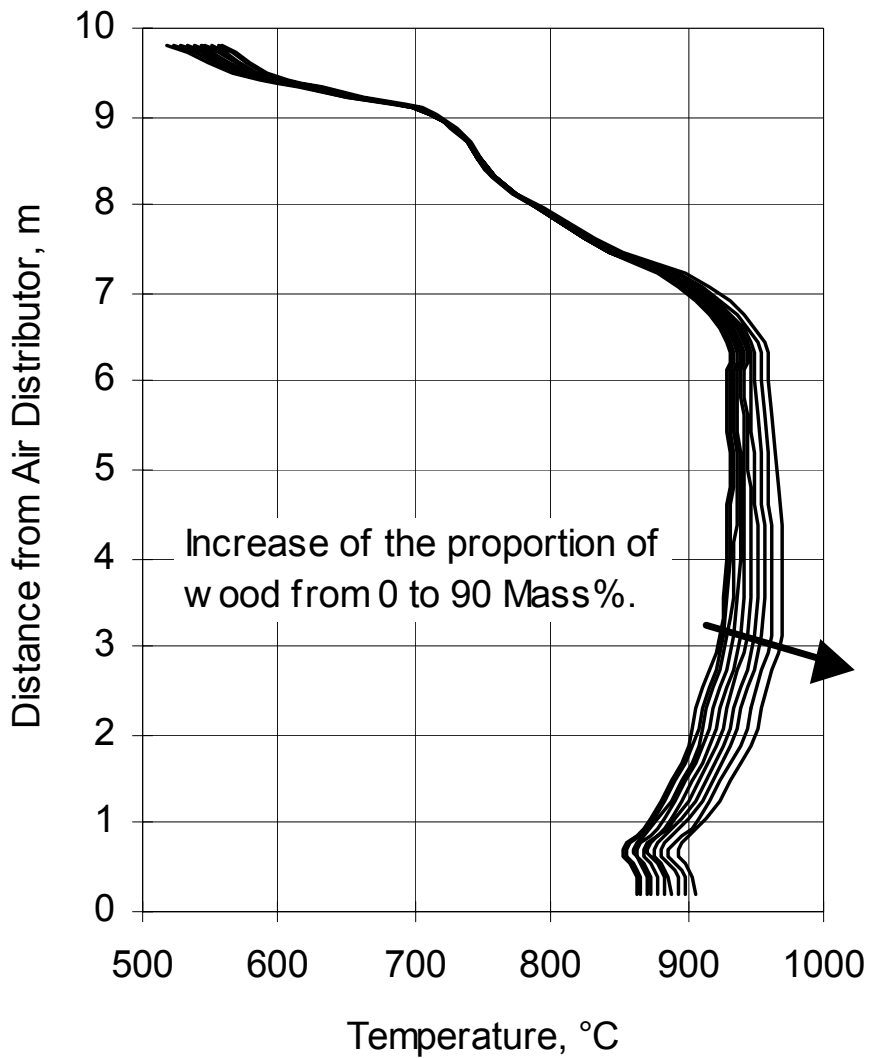
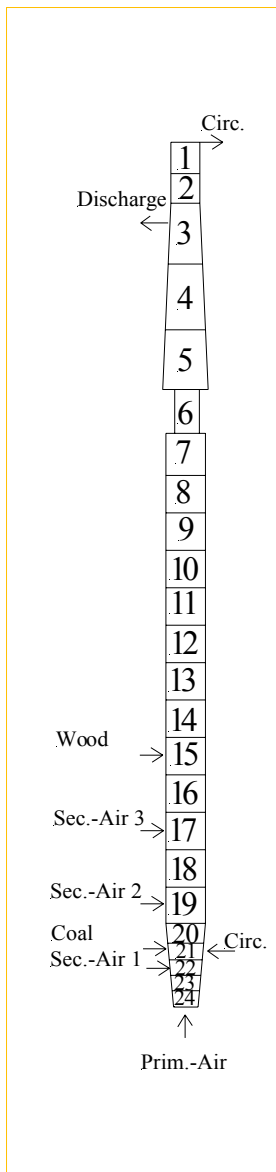
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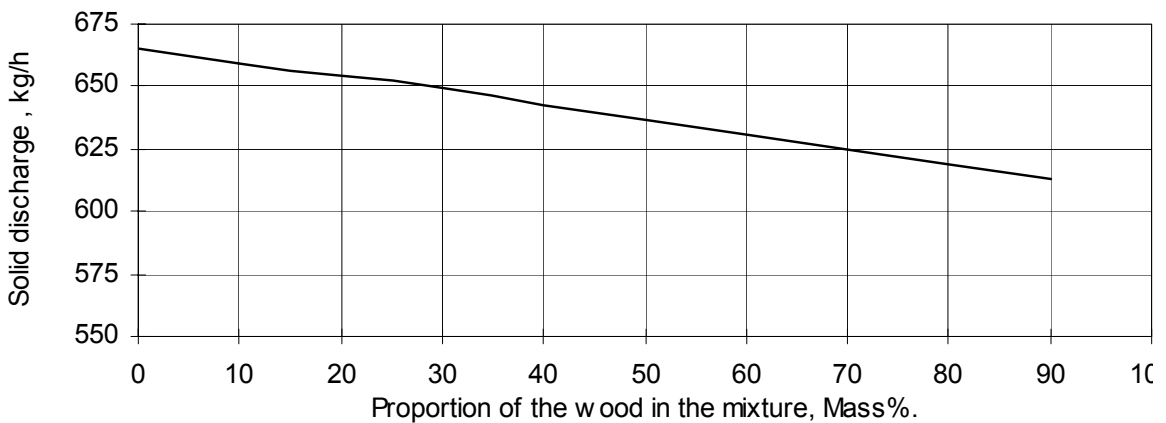
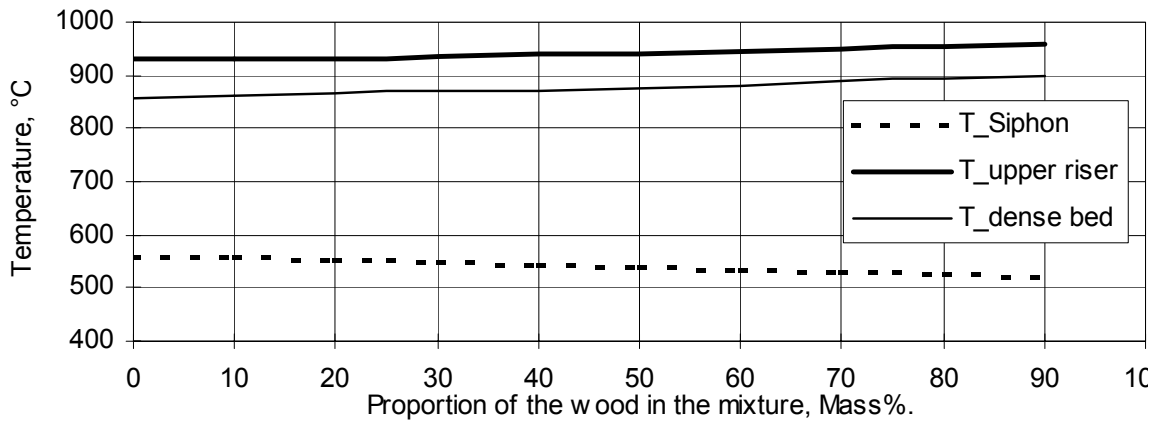
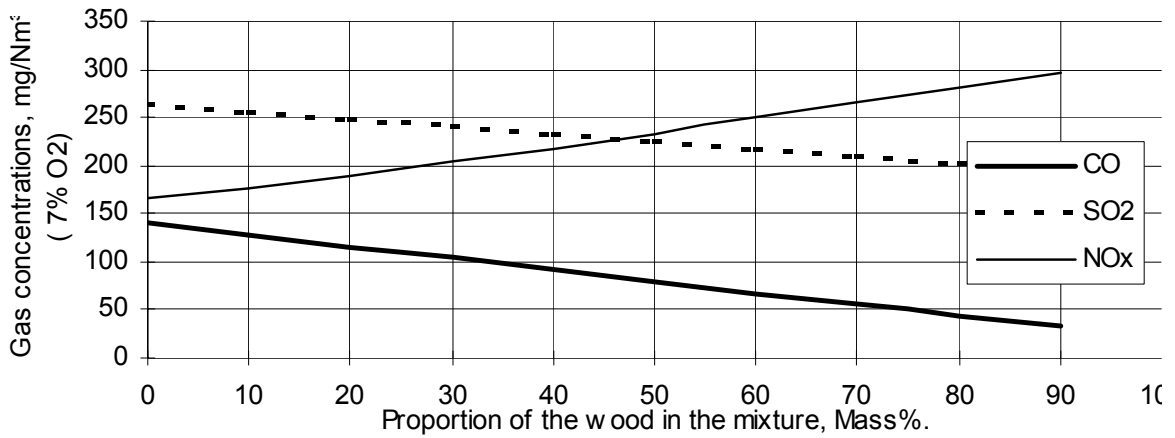
with superproportional wood application the danger of agglomerations is given, special with the Co-fuel waste wood appears this with 950 °C occur by the higher alkali contents, those the ash melting point lower



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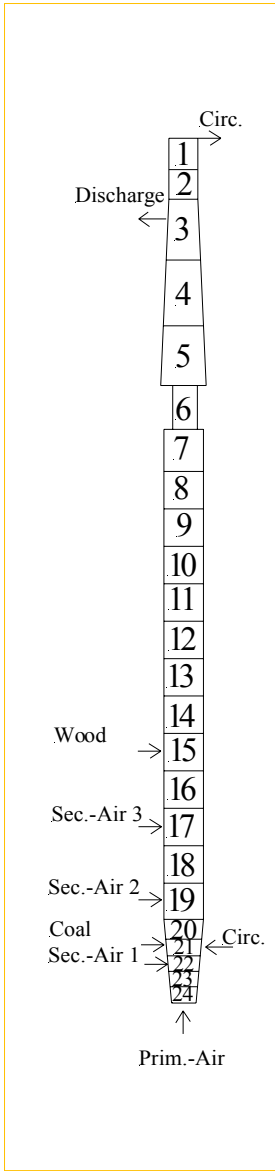
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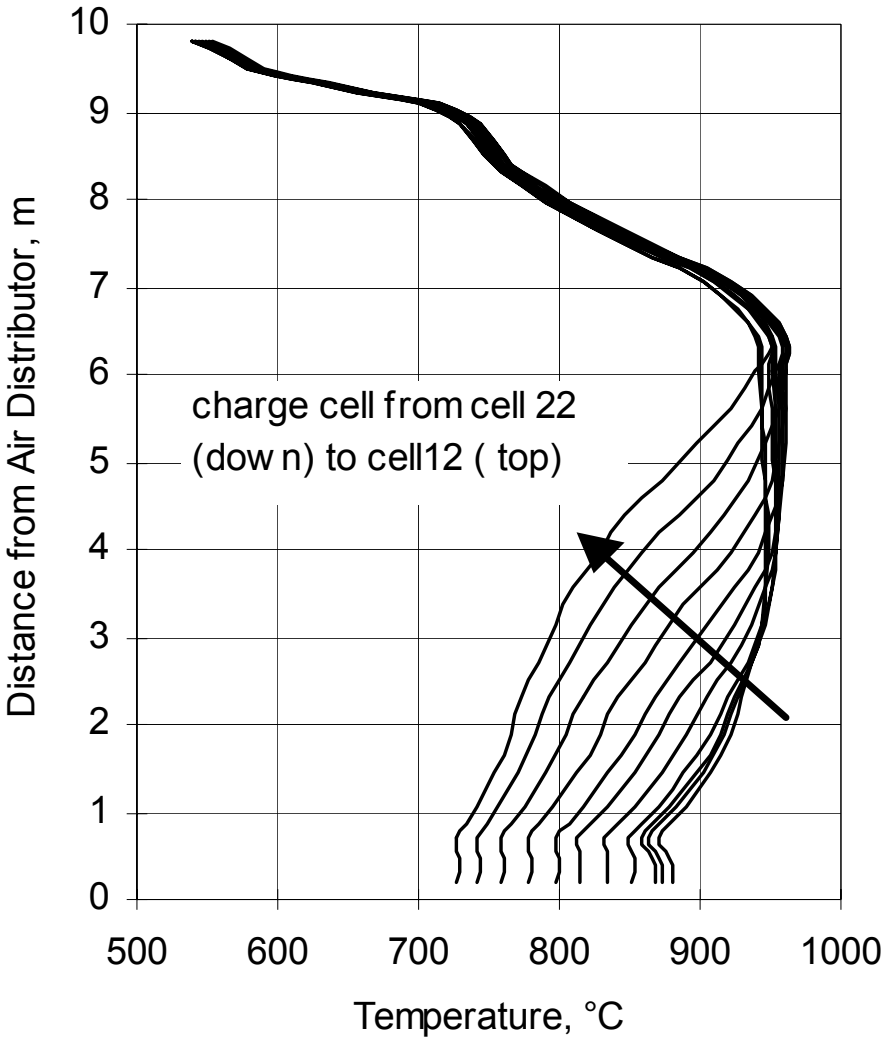
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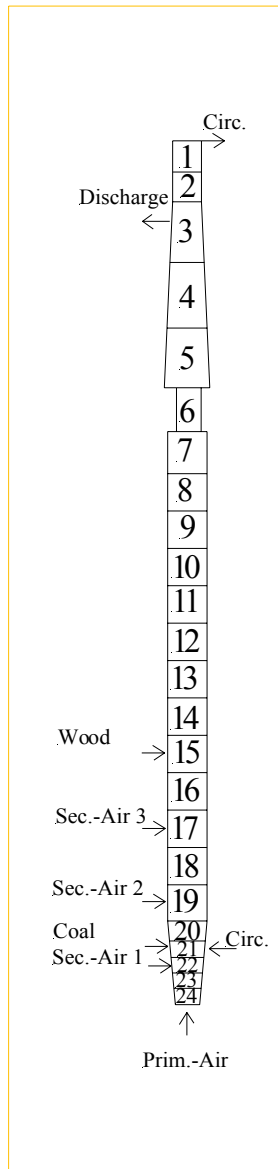
Temperature Course with different charge spots of the Co-fuel



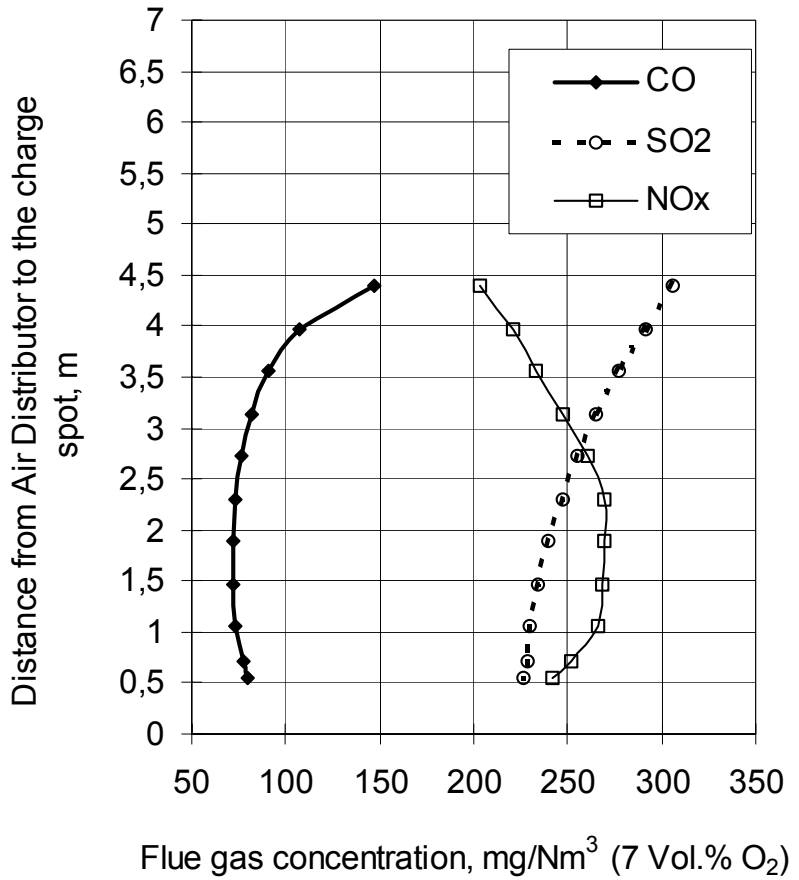
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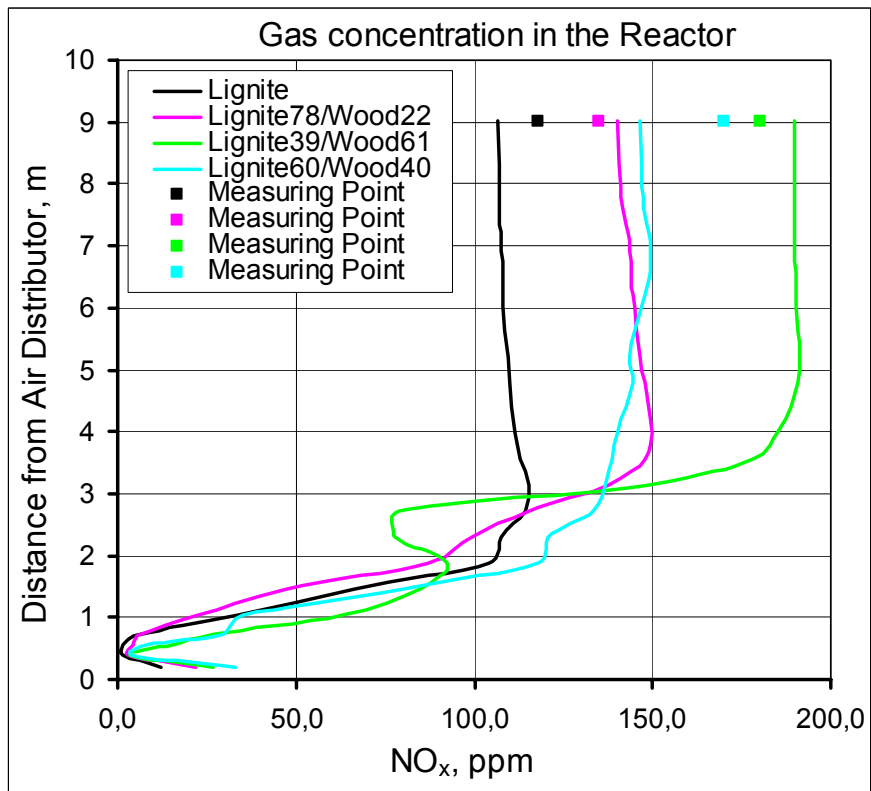
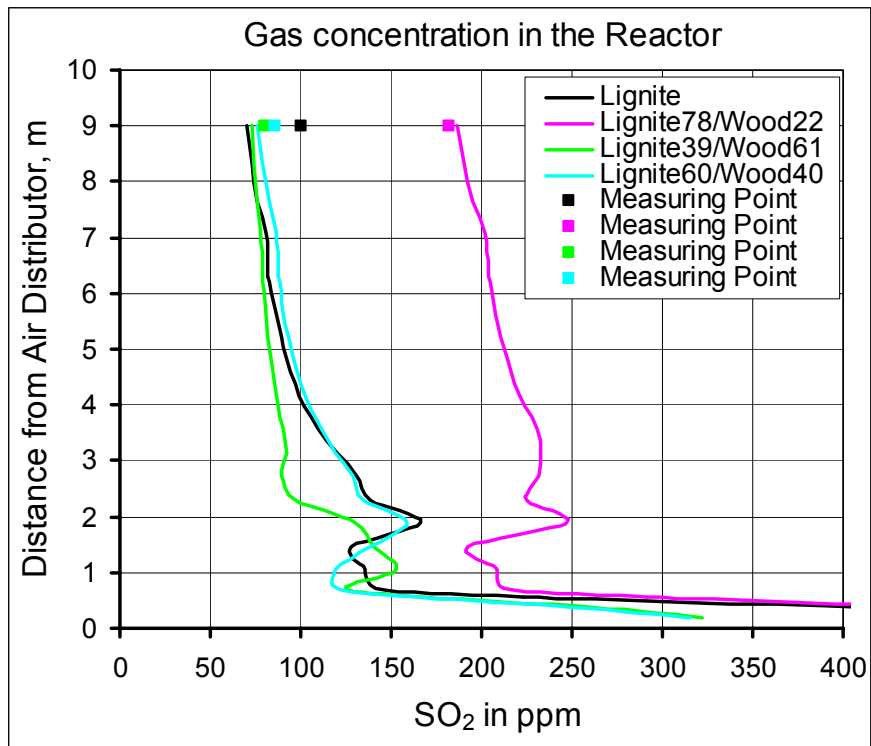
Emission course with different charge spots of the Co-fuel



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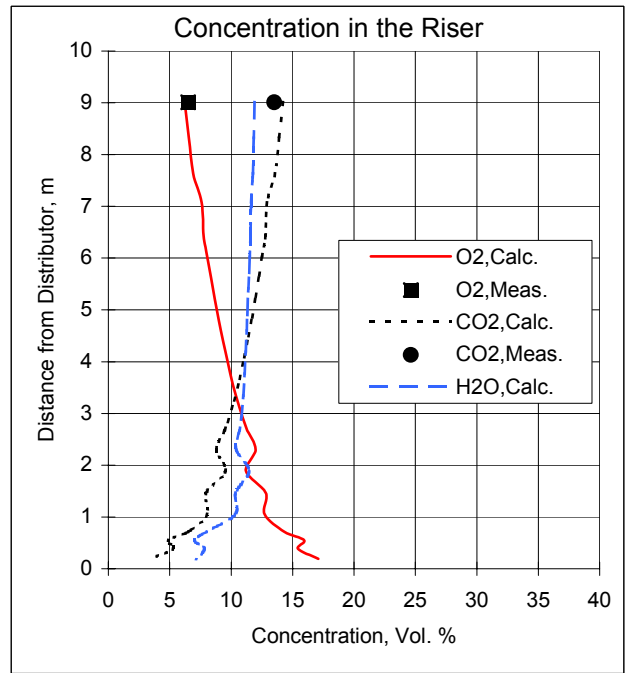
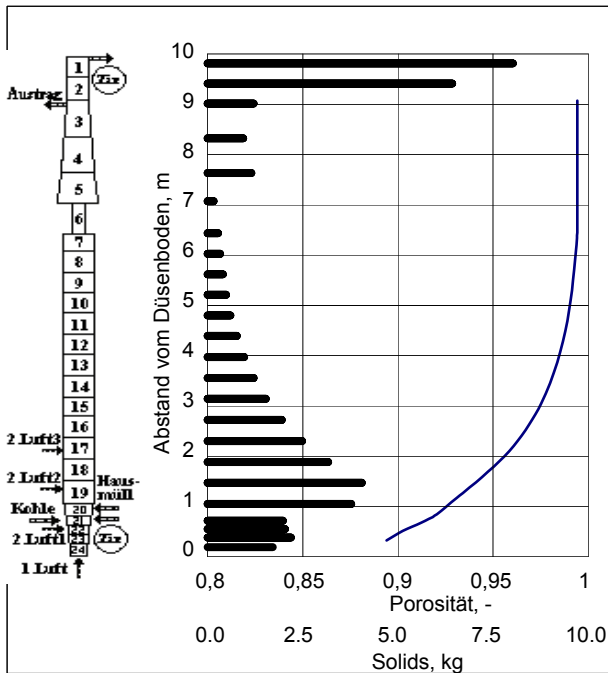
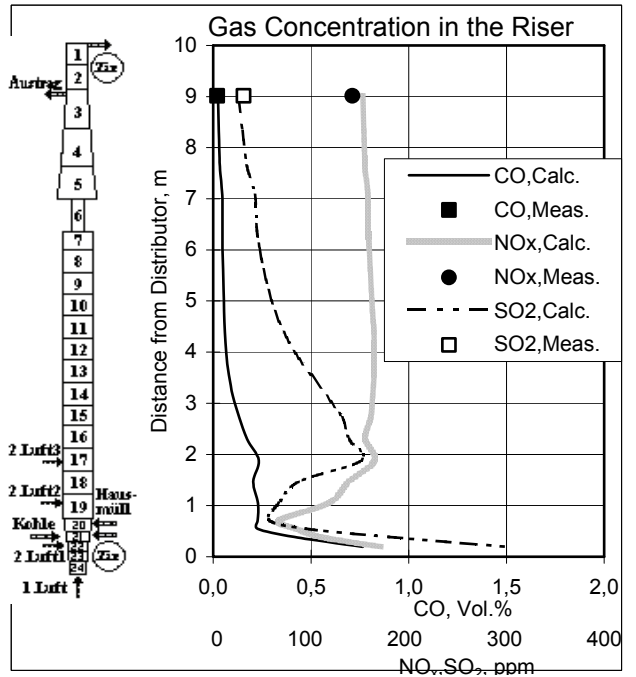
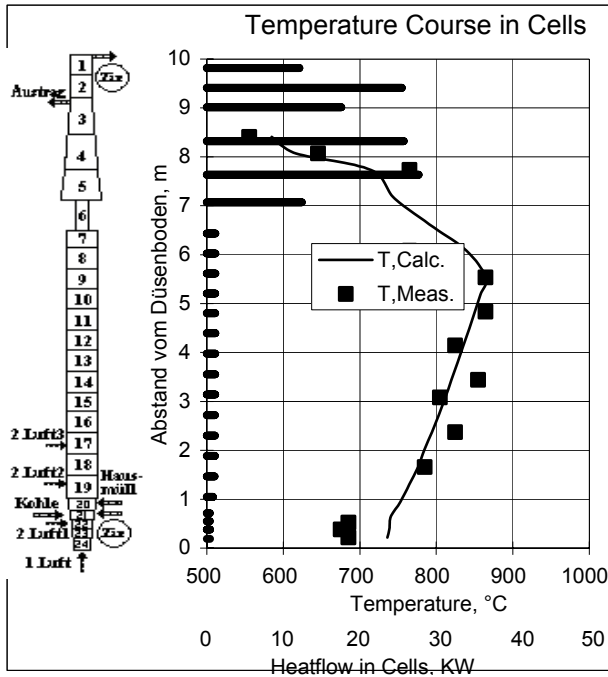
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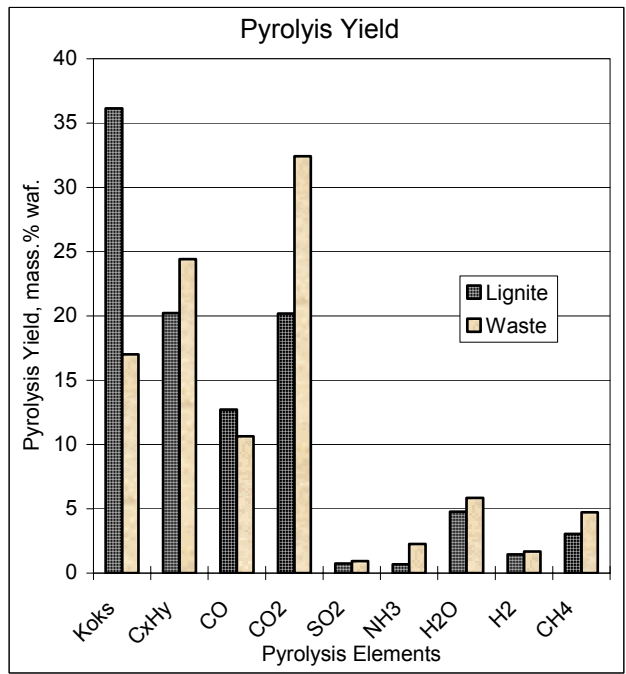
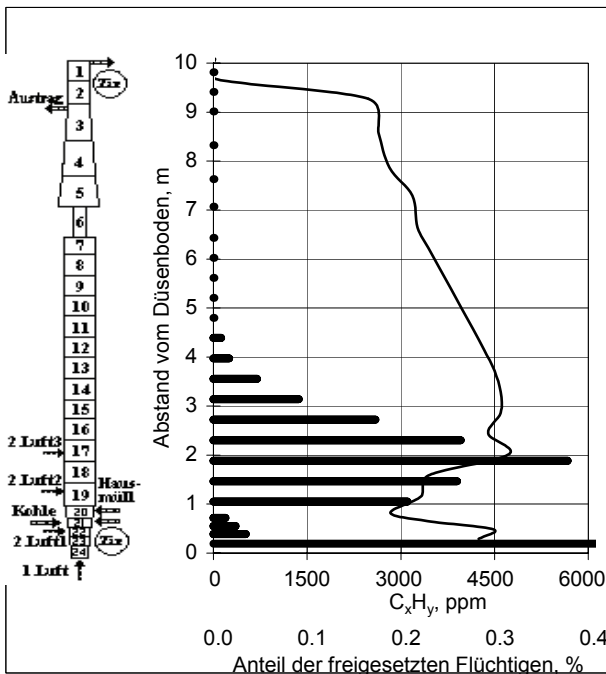
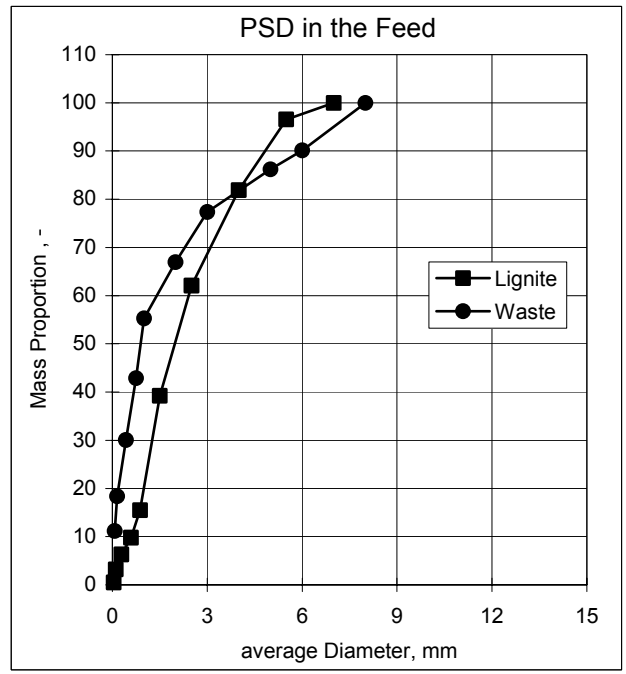
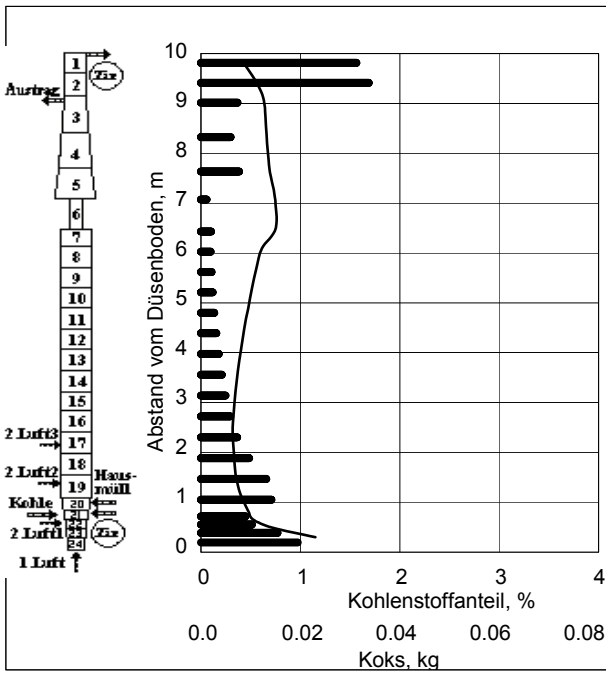
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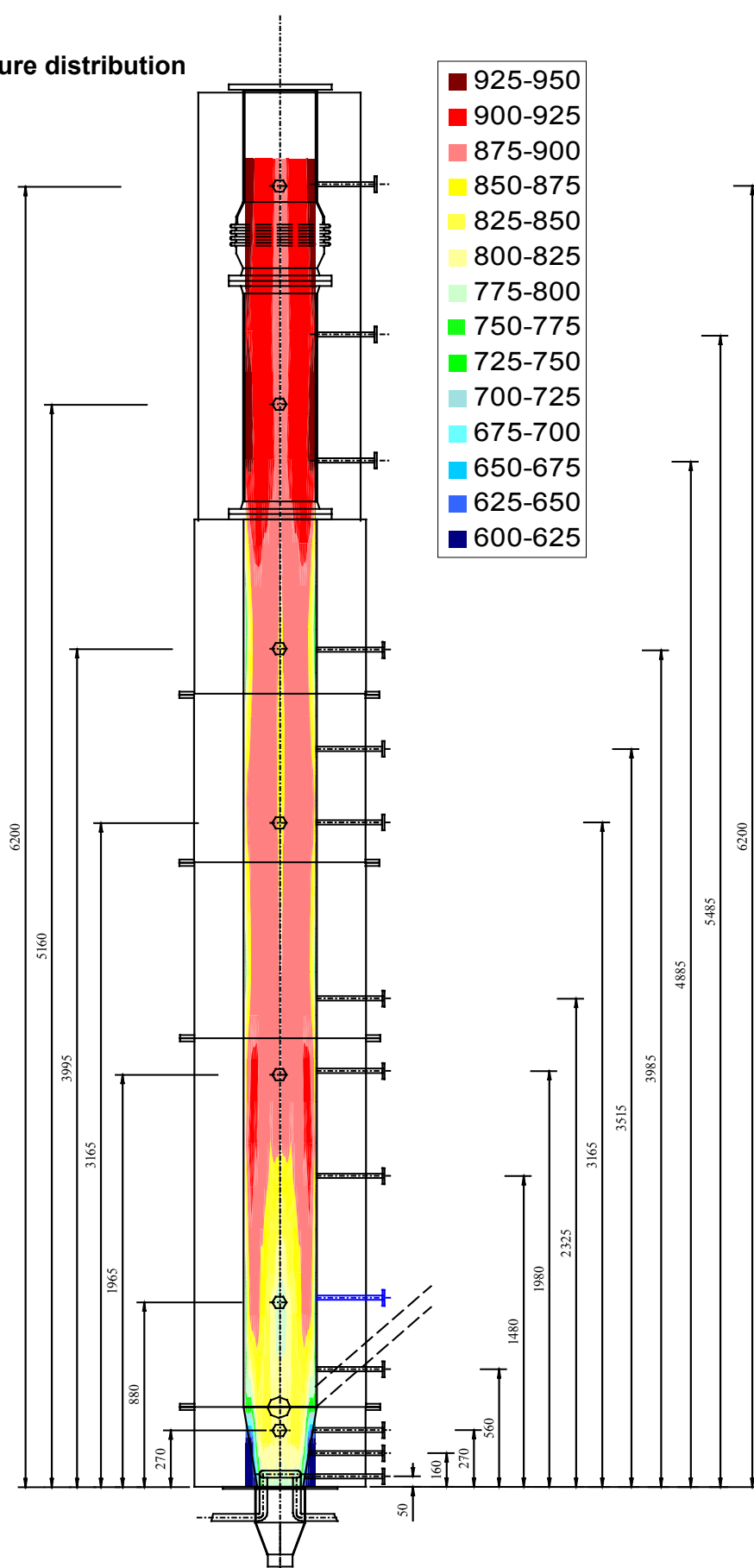
Combustion in ZWSF II		Versuch: tkhm254	Date: 02.07.98			
Participants:	Number: 50	Nr.	Parameter	Unit	Input Cell	
Brunne, T.		1	Lignite	50,0	kg/h	21
Hiller, A.		2	Waste	13,4	kg/h	
Quang, N.T.		3	Air Ratio	1,18	-	-
		4	Th. Load	285	KW	-
TU Dresden		5	Primary Air	128	Nm ³ /h	24
Institut für Energietechnik		6	Second. Air 1.	45	Nm ³ /h	22
Lehrstuhl Kraftwerktechnik		7	Second. Air 2.	52	Nm ³ /h	19
Prof. Dr. -Ing. habil. W. Bernstein		8	Second. Air 3.	58	Nm ³ /h	17
	9	Siphon Air	37	Nm ³ /h	1: 28; 2: 9	



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Temperature distribution

Fuel: Lignite

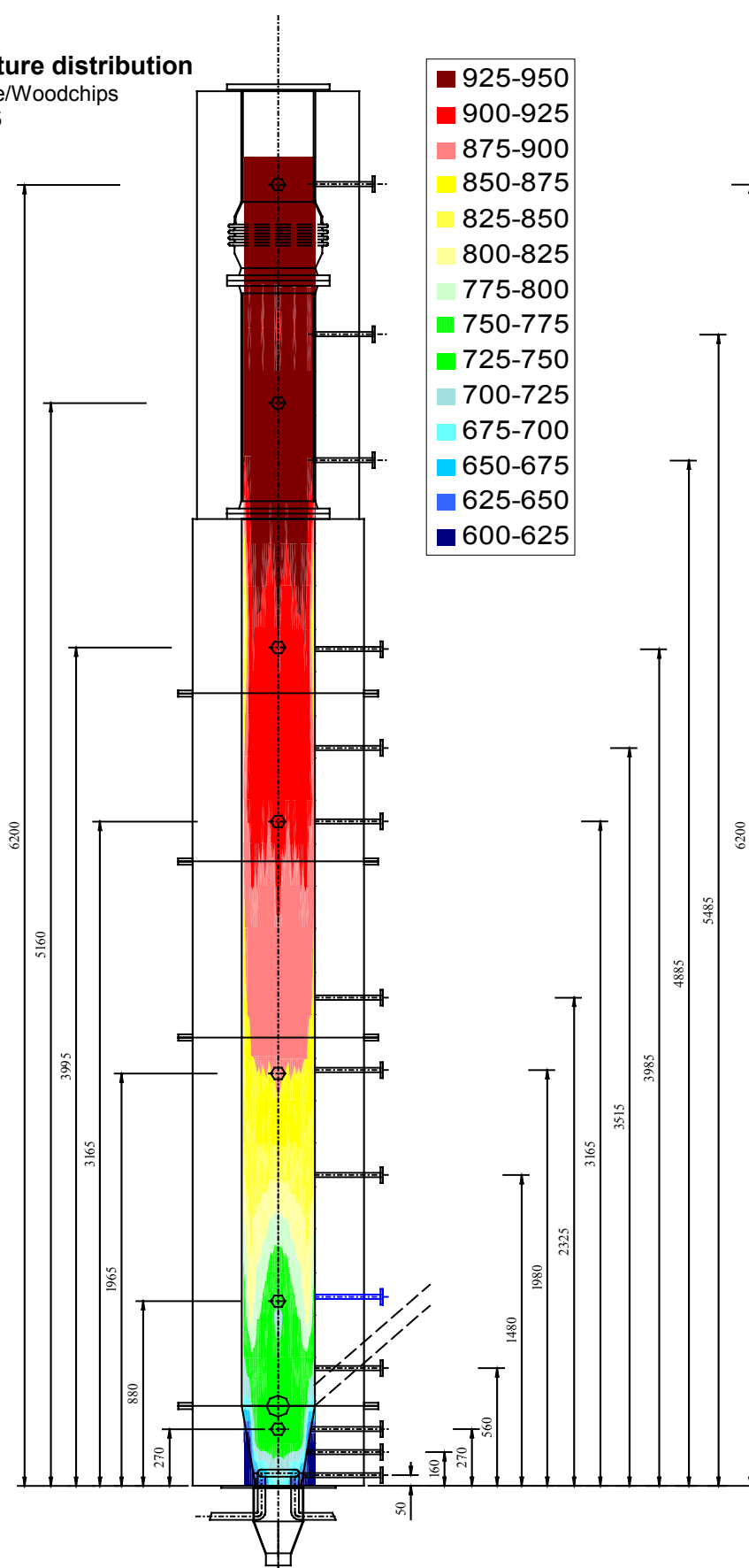


Temperature distribution

Fuel: Lignite/Woodchips

Load: 85/15

260 kW



Outlook

- Determine emission profiles of NO_x and later NH₃,
- In result: optimisation of input spots for co-fuel (reburning),
- Consider dissymmetries as consequence of the charge spots (back-leg and co-fuel),
- Modelling of hard coal,
- Use of further Different Co-fuels: oil sludge, communal waste....,
- Development of the integration of CFB and Gasification (use our co-current fixed bed gasifier),
- Estimate of aspects of economy (and sense electronic waste),
 - Combination of disposal with side effects:

Elementary Analysis	Limestone	?
SiO ₂	2,1 %	2,3 %
Fe ₂ O ₃	0,4 %	3,6 %
Al ₂ O ₃	1,2 %	0,6 %
CaO	53,8 %	46,6 %
MgO	0 %	0,2 %
H ₂ O	0,1 %	6,2 %
S	< 0,01 %	< 0,01 %
Ignition loss 900 °C	43 %	40,5 %

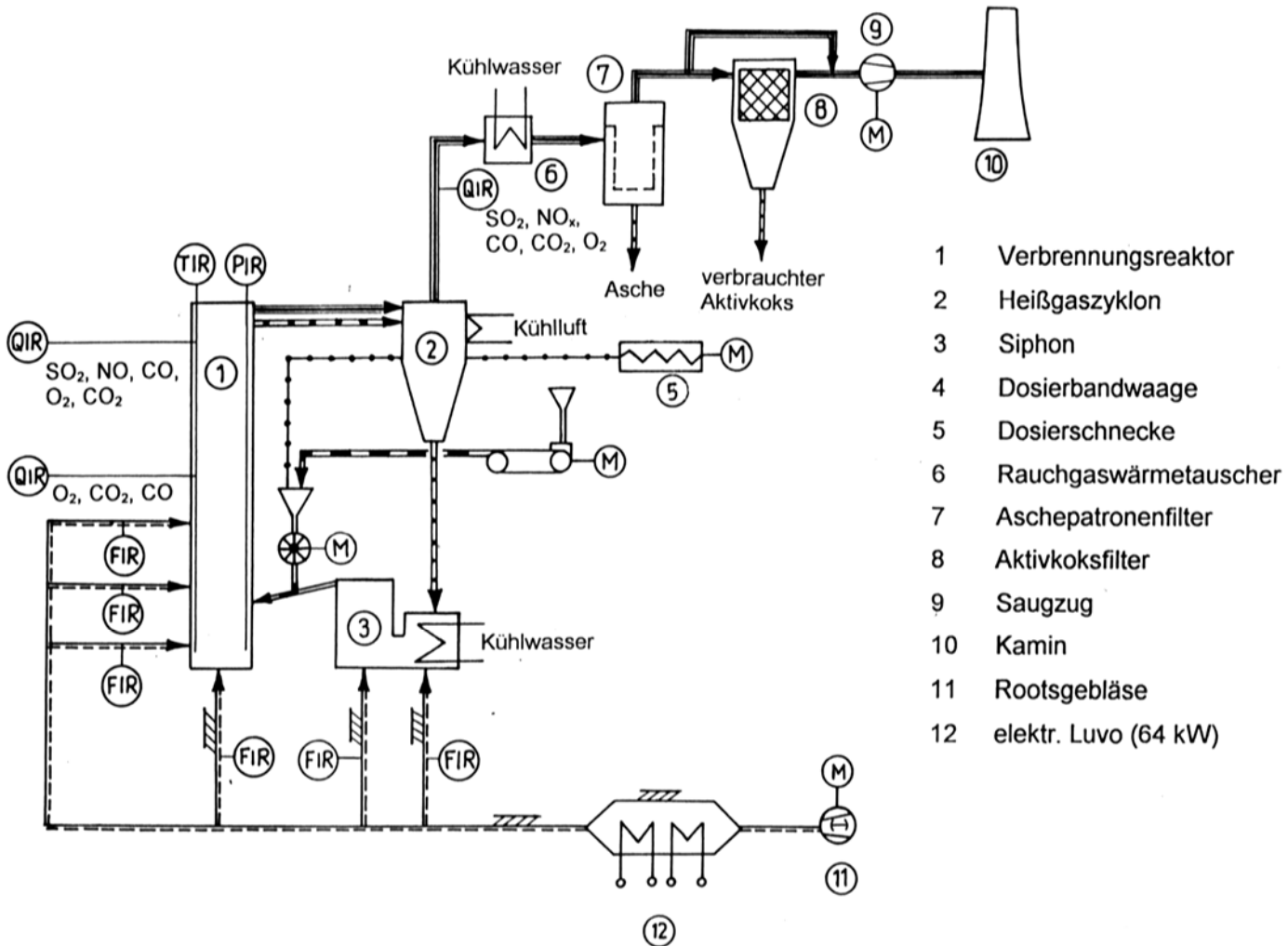
(Limestone and Arrears of the water purification of the power station)



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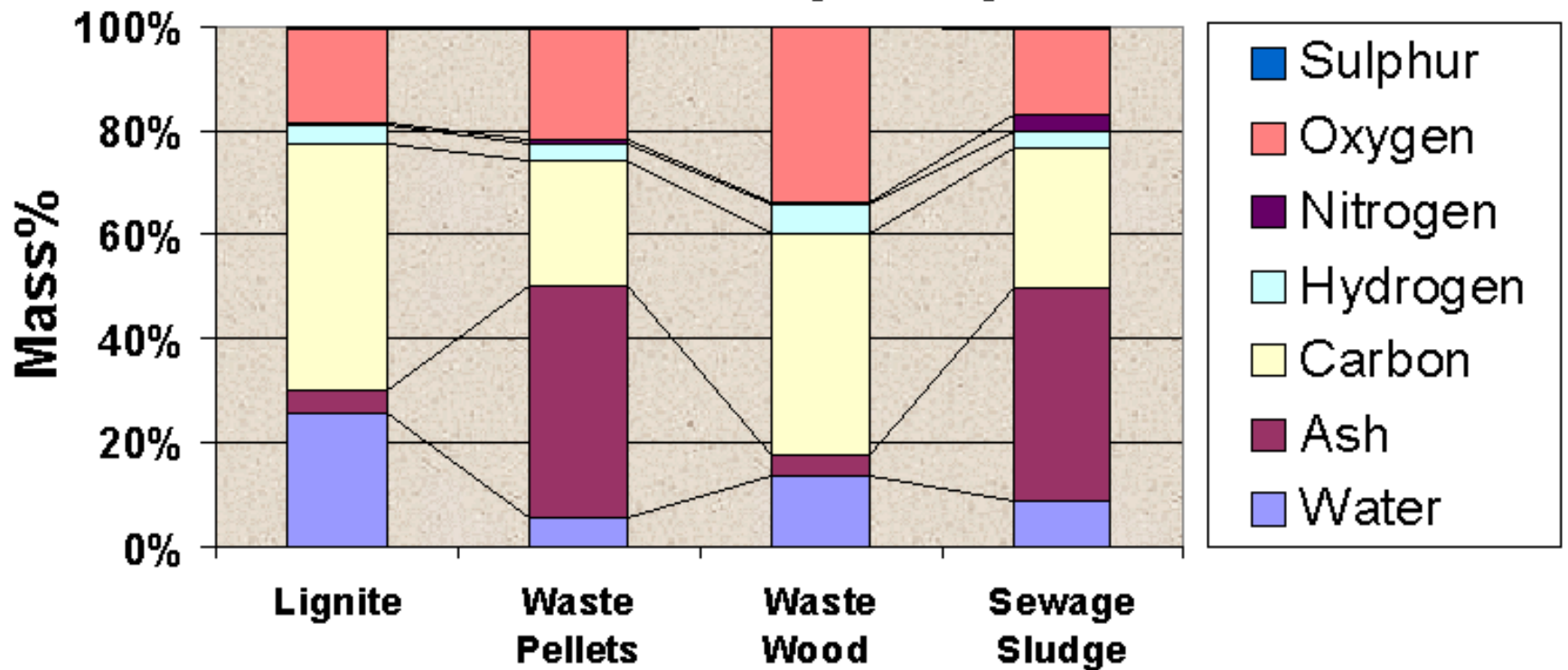
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Fuels		Co-Fuels												
		Ba- gasse	Sewage- Sludge (dry)	Sewage- Sludge (humid)	Com- munal- Waste	Natural- Wood	Waste- Wood	Wood- Pellets	Oil- Sludge	Waste- Sludge (Water- Station)	Paper- Sludge	Paint- Sludge	Elec- tronic- Waste	
		7	9											
Basic Fuels	Lignite-(dry)	18	2	14	15	16	14	10				2	1	2
	Hard-coal	10				4			8	8	9			
	Lignite-(crude)	6												



Elementary Analysis



Volatiles [%]	38,8	41,7	67,3	42,7
Hu [MJ/kg]	17,4	8,7	15,4	10,7

