

CO₂ CAPTURE BY OXY-COMBUSTION IN CFB

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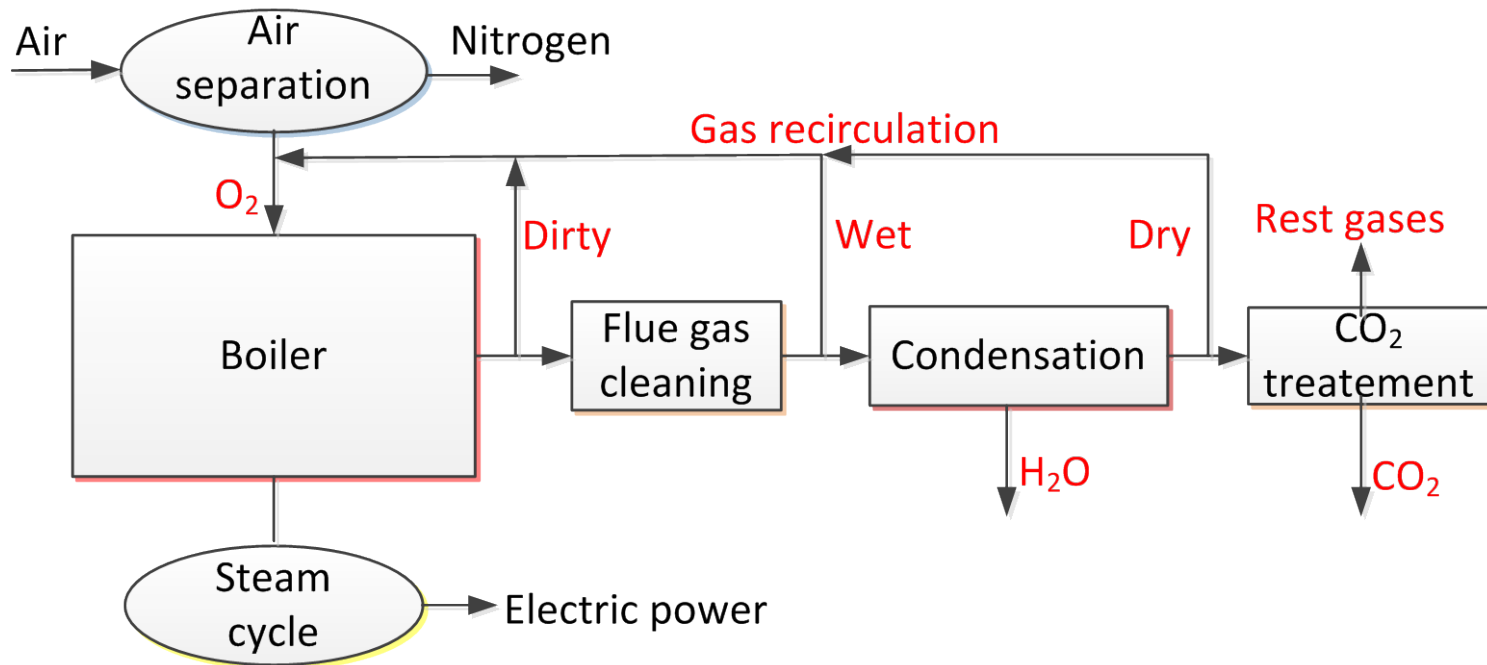
CONDITIONS

CO₂ should be removed in a pure form for deposition.

This is achieved by replacing N₂ in air by CO₂, eg by using pure O₂ diluted by CO₂ from the flue gas instead of air.

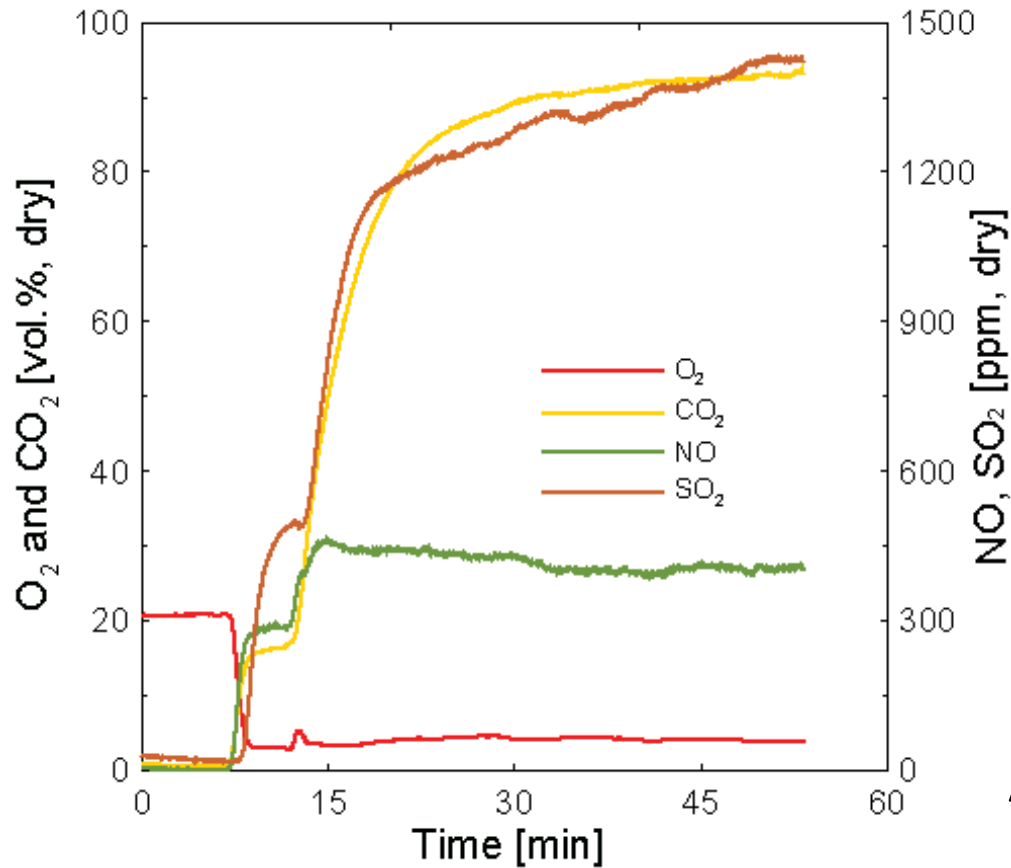
OXY-FUEL CFB BOILERS

Air (O_2+N_2) is replaced by O_2+CO_2 -----dry recirculation
or by $O_2+CO_2+H_2O$ ----wet recirculation .



FLUE GAS RECIRCULATION

Flue gas recirculation leads to enrichment of gases



A start-up sequence taken from flame combustion

GAS DATA

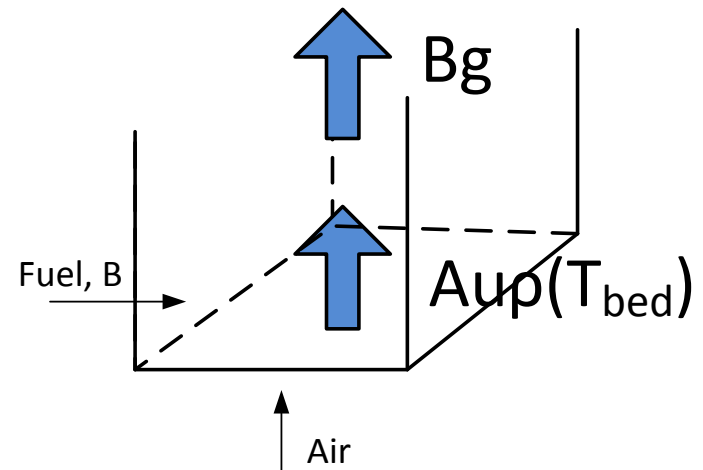
The data for the gas in the air and the oxy-fuel cases are slightly different.

Gas	Density ρ kg/m ³	Specific heat, C_{pg}		Mean specific heat, C_{pgm}	Molecular mass, M kg/kmol
	900 °C	0°C	900 °C	0-900 °C	
O ₂	0.328	0.917	1.100	1.025	32.0
N ₂	0.287	1.041	1.200	1.098	28.0
CO ₂	0.451	0.827	1.275	1.090	44.0
H ₂ O	0.202	2.200	2.343	2.118	18.0

THE GAS FLOW THROUGH AN FBC COMBUSTION CHAMBER

The flue gas flow and the fluidization gas flow are identical

$$Bg = Aup$$



- B feed rate of fuel (kg combustibles/s)
- g flue gas yield at a given air ratio (kg/kg comb.)
- A cross section of the bed (m²)
- u fluidisation velocity (m/s)
- ρ gas density (kg/m³)

CFB: TYPICAL DATA

In all cases the same "reasonable" CFB conditions should be maintained:

- Bed temperature 900 °C
- Fluidization velocity 5 m/s
- Excess oxygen 3-4 %O₂
- Fuel input (power of the boiler) B kg combustibles/s

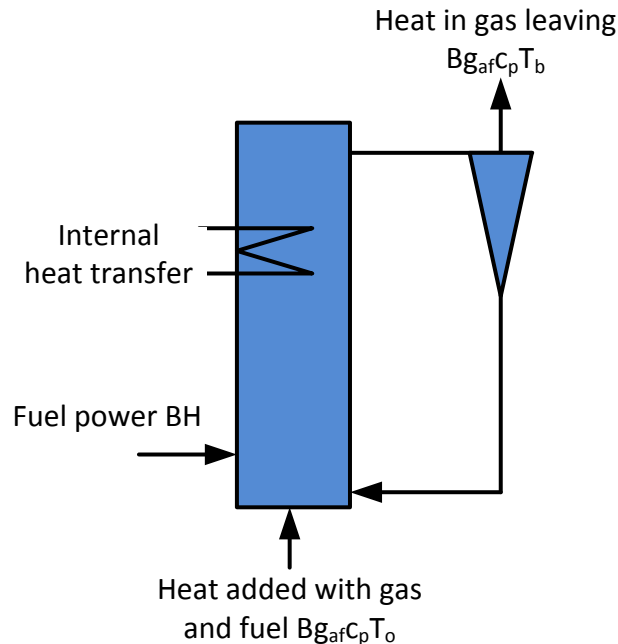
TWO OPTIONS FOR OXY-CFB

The oxy-CFB application for CO₂ capture may have two forms:

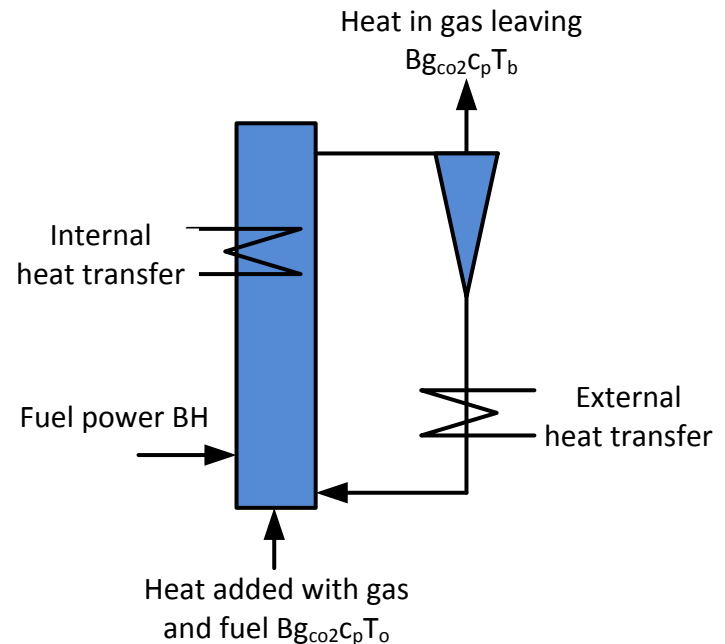
- 1) **"CO₂ capture ready"** → similar to present air-fired CFB boilers
- 2) **New development** → higher oxygen concentration

HEAT RELEASE/ABSORPTION

Exit gas temperature is equal to bed temperature.



Air-fired case (CO₂-capture ready). Combustion heat is transferred to gas and internal heat transfer surfaces.



60% oxy case. Smaller furnace with external heat exchangers, cooling the circulating particles and so the combustion chamber.

THE HEAT BALANCE (Equation 1)

Heat supplied: $BH =$

Heat spent:

$Bgc_p\Delta T_{bed}$ on heating of air and fuel (=heating of combustion gas)+

+heat absorption by furnace heat transfer surfaces+

+and by external heat transfer surfaces $AFc_{ps}\Delta T_{ext}$ (if any)+

+losses (neglected)

1) THE CO₂-CAPTURE-READY CASE

a) HEAT BALANCE (Equation 1)

At constant temperatures and heat transfer, the heat balance becomes $[Bg_{af}\bar{c}_{p,af}\Delta T_{bed}] = [Bg_{co2}\bar{c}_{p,co2}\Delta T_{bed}]$ and

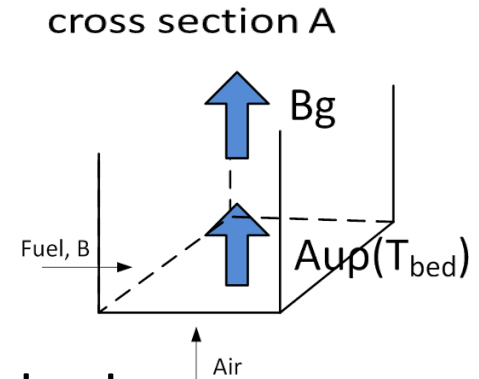
$$g_{af} = \frac{hM_{H_2O}}{2M_h} + \frac{cM_{CO_2}}{M_c} + \frac{1 - y_o}{y_o} \ell_o M_{N_2} + w/b$$

$$g_{CO_2} = \frac{hM_{H_2O}}{2M_h} + \frac{cM_{CO_2}}{M_c} + \frac{1 - y_{o,co2}}{y_{o,co2}} \ell_o M_{CO_2} + w/b$$

are equal at the oxygen concentration of

$y_{o,co2} = 0.28$ in the dry case and about the same in the wet case (Simplified calculation without excess air).

a) GAS BALANCE, $u = \text{const}$



Gas produced by combustion = gas fluidizing the bed

$$B_g = Au\rho(T_{bed}, gas) [kg / s]$$

The gas density is $\rho = 0.29 \text{ kg/m}^3$ in the air case and 0.45 in the CO_2 case. The remaining quantities in the gas balance should be constant in the air and oxy-fuel case, which is not possible.

Instead, optimization between T_{bed} and u is needed (ρ and heat transfer), supported by the choice of primary/secondary air (affecting the local u).

2) THE NEW-DEVELOPMENT CASE

The design is about equal to a conventional CFB

Unchanged: Operation conditions, u and T_{bed} ,
and so the heat transfer coefficient bed-to-surface (same bed suspension density ($\rho_{bed}(u)$) and circulation rate $F(u)$).

Changes: Oxygen concentration: between 0.21 and 1.0 (0.60 is chosen as a maximum)

- The corresponding flue gas recirculation ($\text{CO}_2, \text{H}_2\text{O}$)
- The corresponding heat release per x-section area $\text{MW}_{\text{fuel}}/\text{m}_2$.

Consequences: x-section surface A is reduced, and additional heat removal (in the return loop) is needed.

THE FURNACE CROSS-SECTION AREA

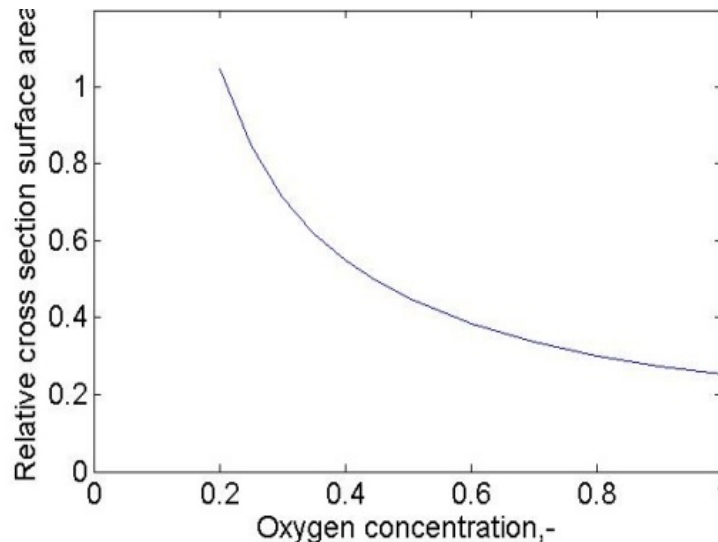
Gas yield from combustion Bg = Fluidization gas $uA\rho_{gas}(T)$

B fuel feed rate (kg fuel/s)--given

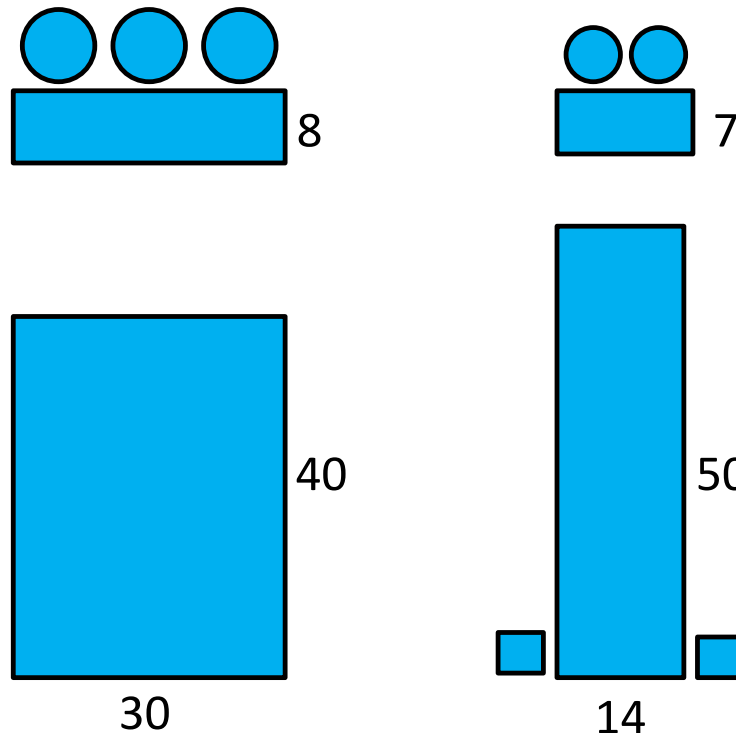
u fluidisation velocity (m/s)--given

ρ_{gas} gas density (kg/m³)---given @ given temp

$$A_{CO_2}/A_{af} = (g_{CO_2}/g_{af})_{\text{volume}}$$



AIR-FIRED BOILER OF SIZE 300 MW_e COMPARED WITH AN OXY-FIRED @ 60% O₂ (Dimensions in m).



The volumes of the two furnaces

$$V_{af} = 9600 \text{ m}^3$$

$$V_{co_2} = 4900 \text{ m}^3$$

The air fired boiler has $A_{af} = 240 \text{ m}^2$ and $g_{af} (y_o = 0.21, N_2)$.

The oxy-fired one has $A_{co_2} = 98 \text{ m}^2$ and $g_{co_2} (y_{o,co_2} = 0.60, CO_2)$

HEAT BALANCE CALCULATION (Equation 1)

The air-fired case (no external heat exchangers):

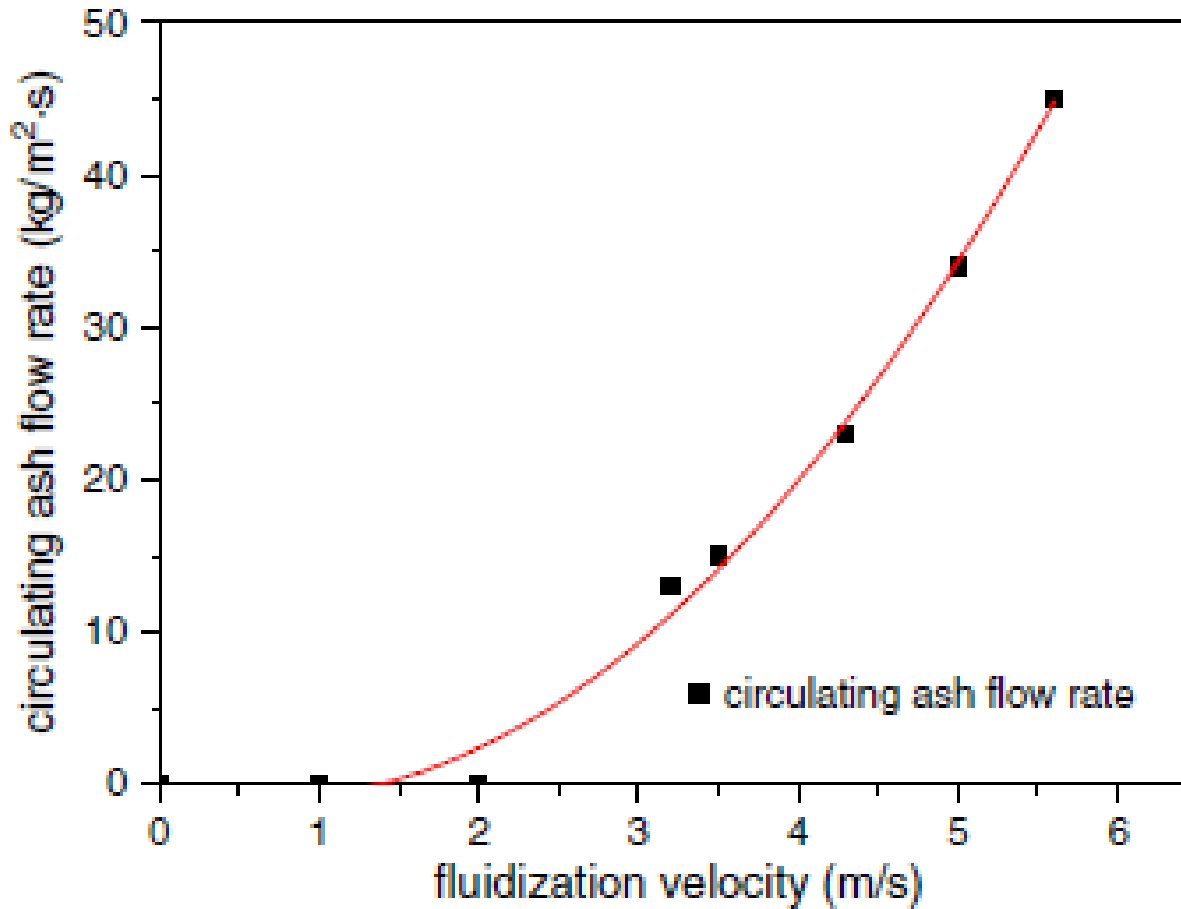
$$\text{Internal heat transfer} = BH - Bg_{af} c_{paf} \Delta T_{bed}$$

The oxy-fired case, including external heat exchangers

$$AFC_{psolids} \Delta T_{external} = BH - Bc_{pco2} g_{co2} \Delta T_{bed} - (V_{co2}/V_{af})(BH - Bg_{af} c_{paf} \Delta T_{bed})$$

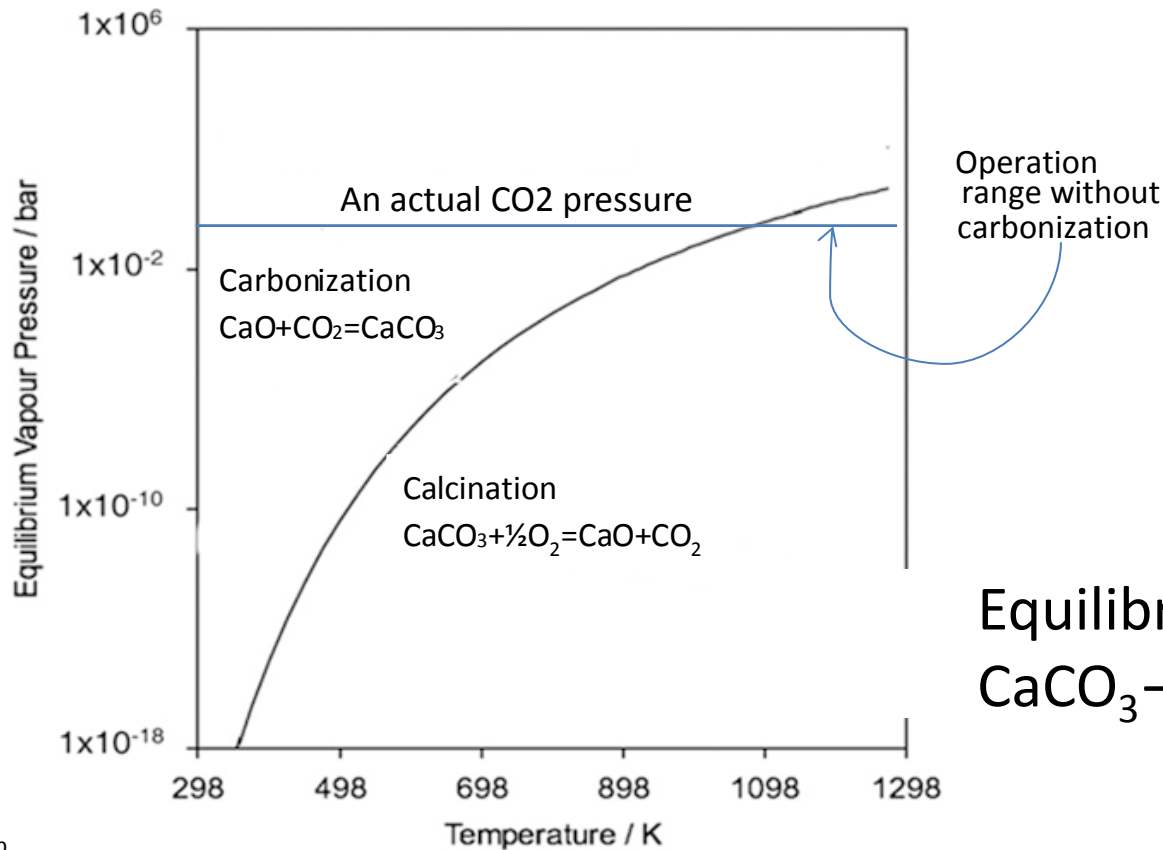
With a circulation flux of $F=35 \text{ kg/m}^2\text{s}$ this gives $\Delta T_{external} = 90 \text{ K}$

CIRCULATING FLUX VS VELOCITY MEASURED ON 300 MW_e CFB BOILERS



CONSEQUENCES OF EXTERNAL HEATING

Even considering the uncertainty of the recirculation rate, the cooling of the external particle flow is moderate, but with limestone in the bed recarbonization may occur.



Equilibrium diagram
 $\text{CaCO}_3 - \text{CaO} - \text{CO}_2$

b

SUMMARY ON OXY-CFB DEVELOPMENT

At present there is **no commercial oxy-CFB in operation.**

The heat balance **cannot** be satisfied at the same time as $u=\text{const}$ in the CO₂-capture-ready case. Some (minor) adjustment is needed.

The maximum oxygen concentration that can be used is unknown at present.

The moderation of temperature by the bed material in oxy-CFB boilers is an advantage compared to oxy-PC.