

Co-combustion of waste with coal in fluidized bed systems - an overview

Joachim Werther


Technical University Hamburg-Harburg
D 21071 Hamburg, Germany

Definition and Scope

What is waste?

‘Waste’ means any substance or object which the holder disposes of or is required to dispose of pursuant to the provisions of national law in force

Waste considered in the context of co-combustion with coal in power power plant:

- waste wood
 - agricultural residues
 - sewage sludge
- 
- ```
graph LR; A[waste wood] --> C[biomass]; B[agricultural residues] --> C; D[sewage sludge] --> C;
```
- biomass

# CONTENTS

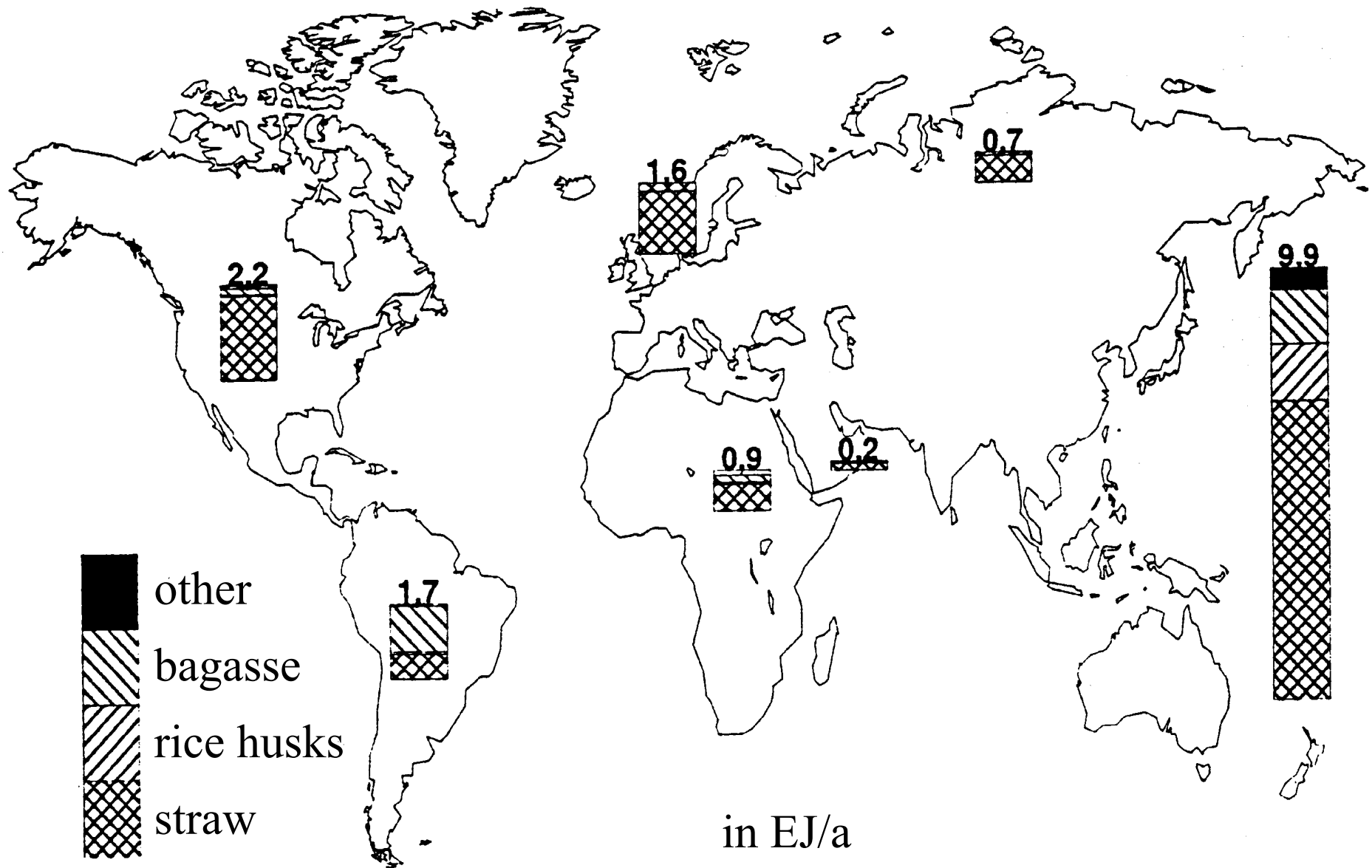
---

1. Introduction
2. Waste fuel characteristics
3. Advantages and disadvantages of co-combustion
4. Co-combustion experiments
5. Legal issues of co-combustion
6. Experience from large-scale plants

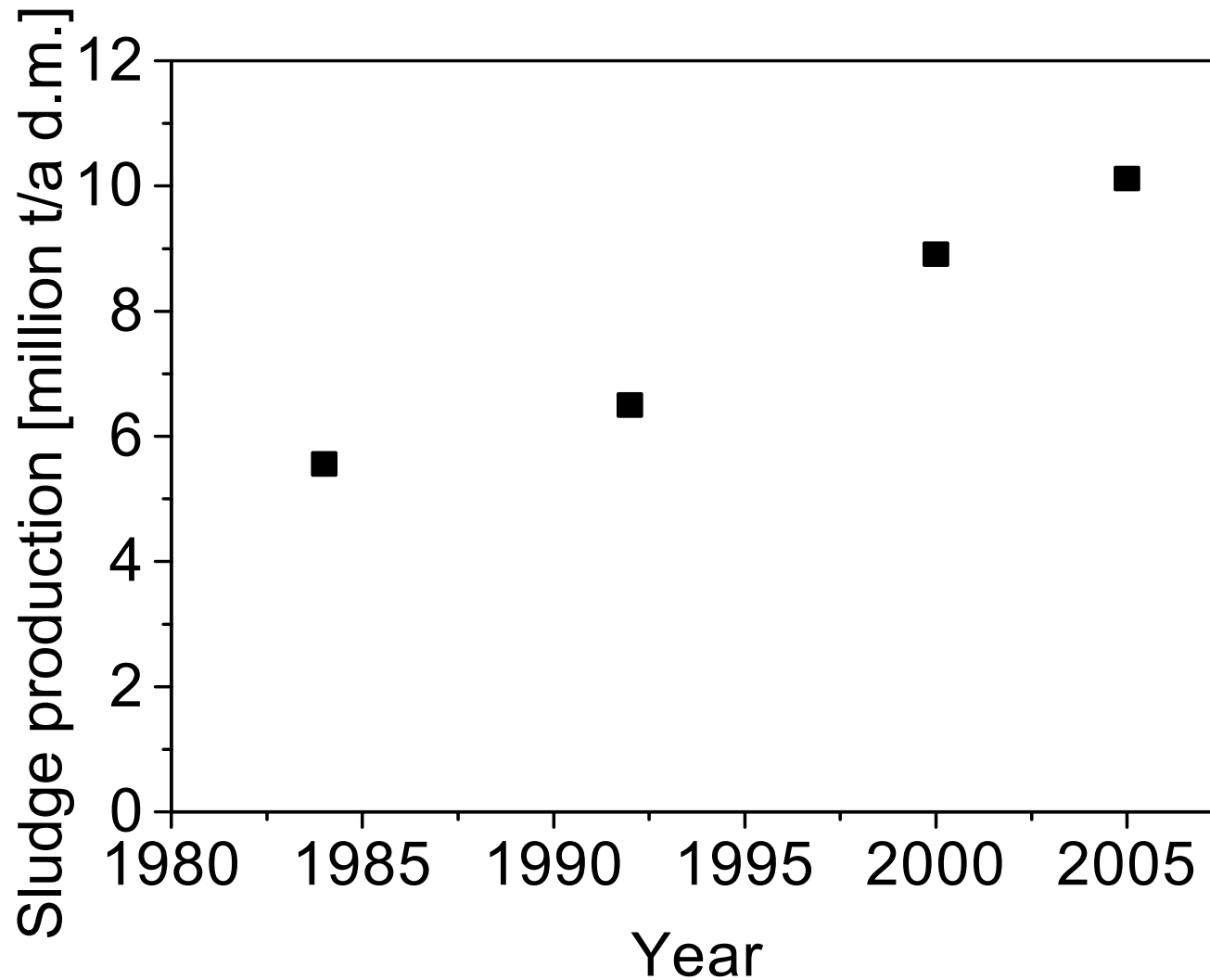
# Motivation for utilization of waste for power generation

- 1. Energy Potential
  - Wood wastes and residues: 41.6 EJ/a
  - Agricultural residues: 17.2 EJ/a
  - Incentives for use of renewable energies
- 2. Kyoto Protocol
  - Overall reduction of GHG by 5 %
  - European Union committed to 8 % reduction
- 3. Sewage sludge
  - Land filling to decrease (e.g. TASI)
  - Incineration to increase
- 4. Governmental actions
  - Germany has decided to step out of nuclear energy
  - ‘Renewable energy law’ promises subsidies (0.1 €/kWh)
  - In Denmark: 1 million tons/year straw to be used for power generation

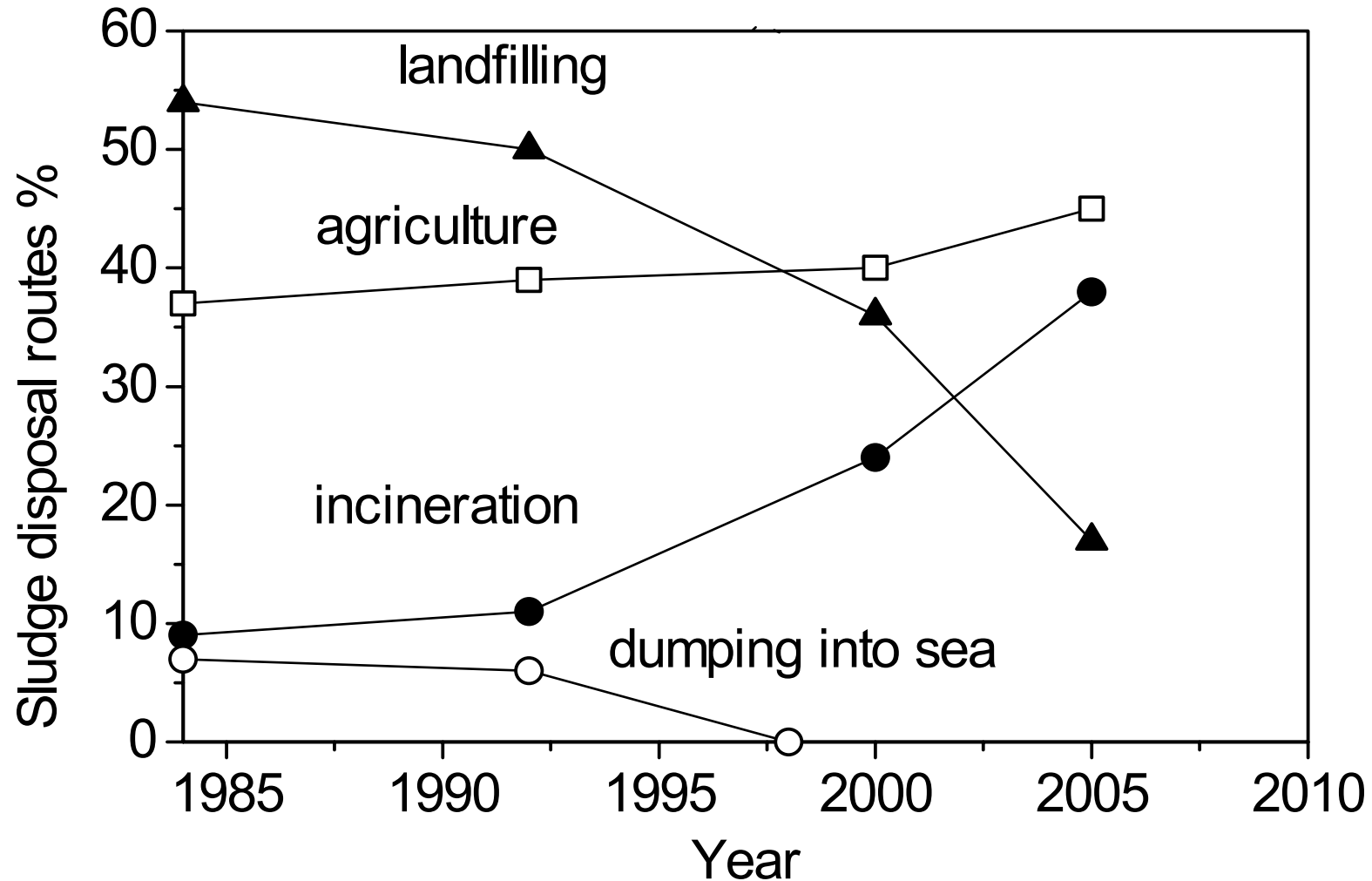
# Energy potential of agricultural residues



# Estimation of sewage sludge production in EU



# Sludge disposal routes in the EU



# Advantages of co-firing waste with coal

- Capitalize on economy of scale
- Useful within economic transport distance of 50-80 km
- Caters well for seasonality of biomass production
- High volatile matter of biomass compensates high carbon content of coal
- Low cost CO<sub>2</sub> reduction technique
- Low SO<sub>2</sub> emission credits
- Coal furnaces improve the efficiency of biomass combustion
- Coal furnaces have efficient flue gas cleaning
- A reliable short term disposal opportunity for wastes



# Problems with utilization of waste as fuels

## 1. For agricultural residues:

- local production + low density → high transport costs
- annual harvest requires large storage space
- storage is associated with loss of biological substance and bears risk of self ignition
- low energy density requires larger plant size → higher investment costs than fossil -fired plants

## 2. For agricultural residues and sewage sludges:

- wide variety of fuel characteristics
- high volatile content
- risk of slagging, fouling, corrosion and agglomeration

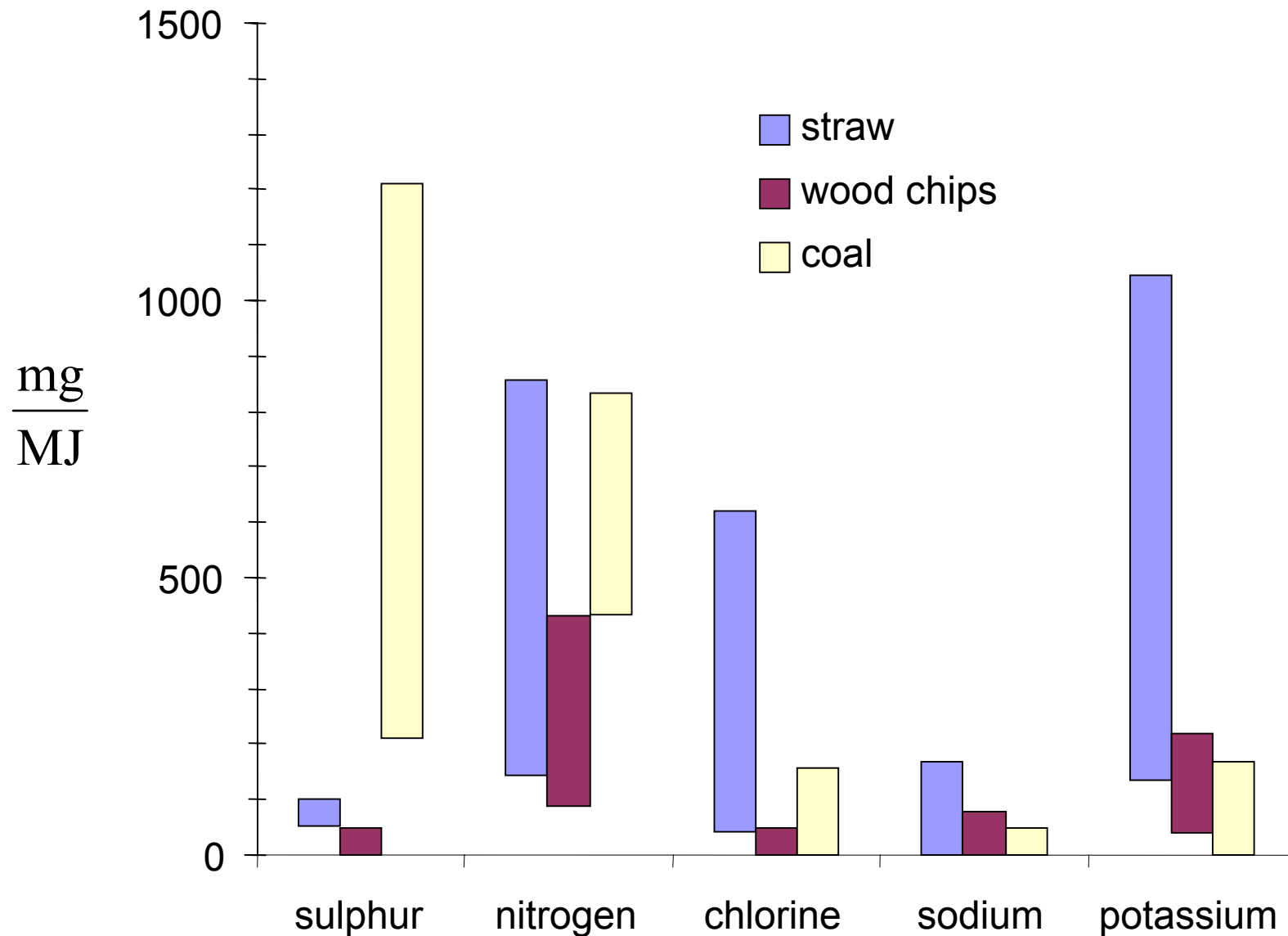
# Fuel characteristics

|                     | proximate analysis |                     |                   |              | ultimate analysis (%waf) |     |      |      |      |
|---------------------|--------------------|---------------------|-------------------|--------------|--------------------------|-----|------|------|------|
|                     | moisture<br>(%raw) | volatiles<br>(%waf) | fixed C<br>(%waf) | ash<br>(%wf) | C                        | H   | O    | N    | S    |
| road side grass     | 16                 | 61.7                | 38.3              | 23.1         | 48.2                     | 6.6 | 41.9 | 1.94 | 0.24 |
| wheat straw         | 20                 | 77.0                | 23.0              | 5.7          | 48.4                     | 6.2 | 45.0 | 0.58 | 0.09 |
| soy husk            | 6.3                | 78.6                | 21.4              | 5.4          | 45.4                     | 6.7 | 46.9 | 0.9  | 0.1  |
| coffee husk (mbuni) | 11.4               | 76.4                | 23.7              | 4.6          | 43.9                     | 4.8 | 49.6 | 1.6  | 0.1  |
| coconut shell       | 4.4                | 76.2                | 23.8              | 3.2          | 51.2                     | 5.6 | 43.1 | 0.0  | 0.1  |
| sewage sludge       | 6.9                | 86.4                | 13.6              | 44.6         | 52                       | 6.3 | 32.1 | 6.3  | 3.1  |
| wood                | 40.0               | 84.9                | 23.3              | 0.8          | 50.7                     | 5.9 | 43.1 | 0.2  | 0.04 |
| peat                | 37.0               | 69.8                | 30.2              | 6.8          | 57.1                     | 5.9 | 43.1 | 2.3  | 0.8  |
| bituminous coal     | 7.5                | 38.8                | 61.2              | 5.3          | 88.0                     | 6.0 | 4.0  | 1.2  | 0.8  |

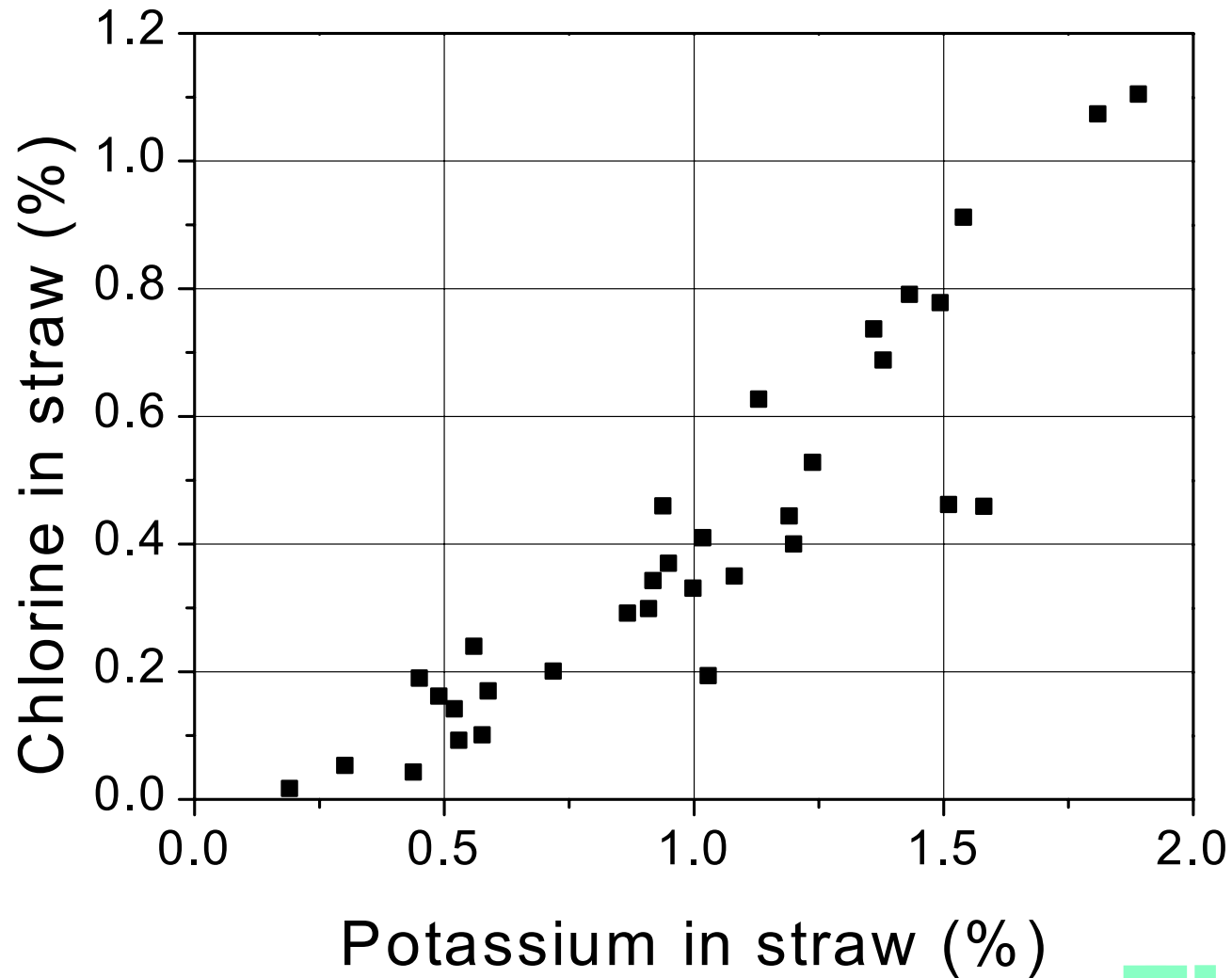
# Characteristics of agricultural residues<sup>11</sup> used as fuels

- moisture content depends on processing (coffee beans are dried before hulling, straw quality depends on weather conditions at time of harvesting)
- volatiles content is significantly higher than in coals
- ash content is low
- oxygen content is high
- low sulfur content but high contents of chlorine ( → HCl emissions, dioxin formation, corrosion), sodium and potassium (→ ash melting behavior, agglomeration)

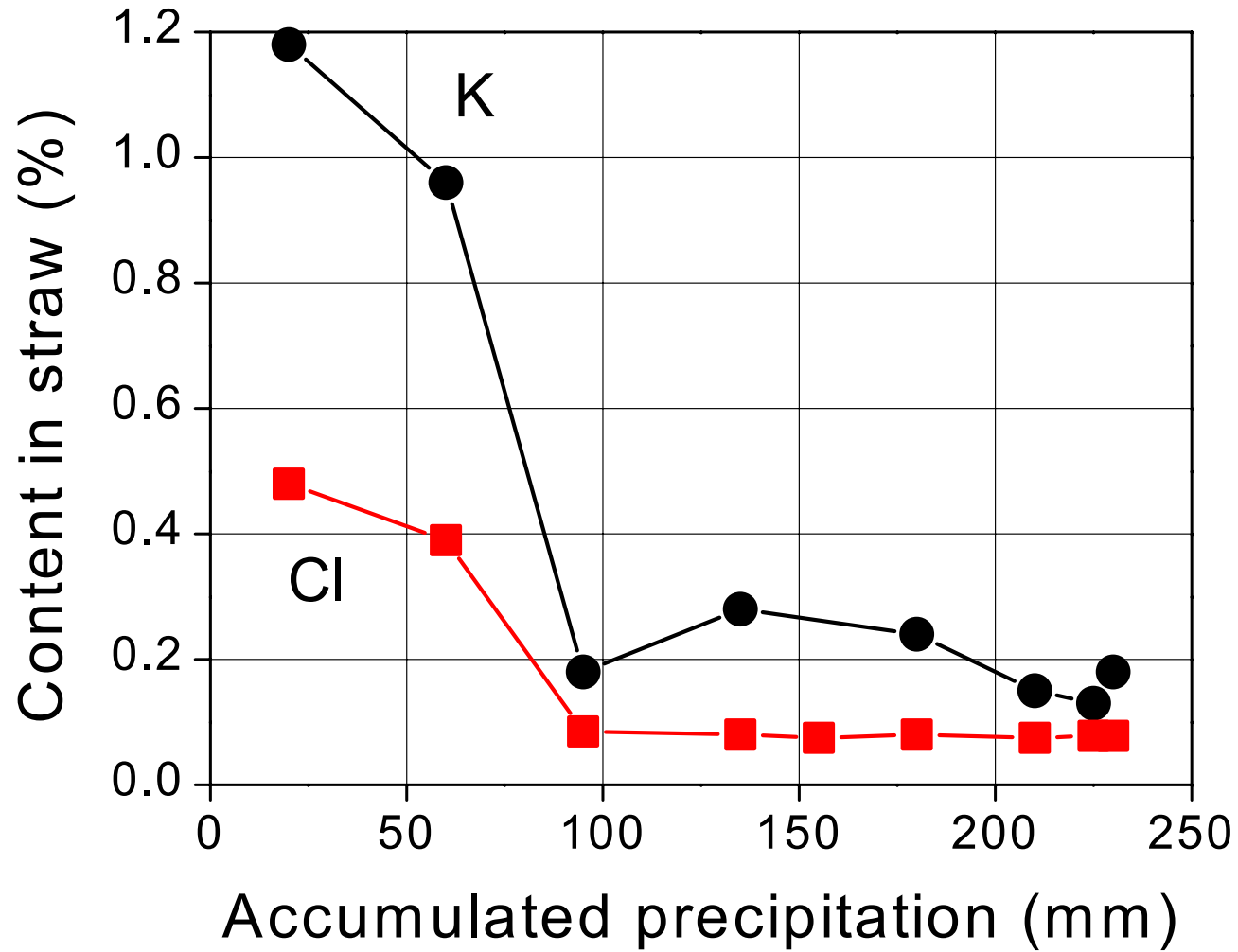
# Biofuel characteristics



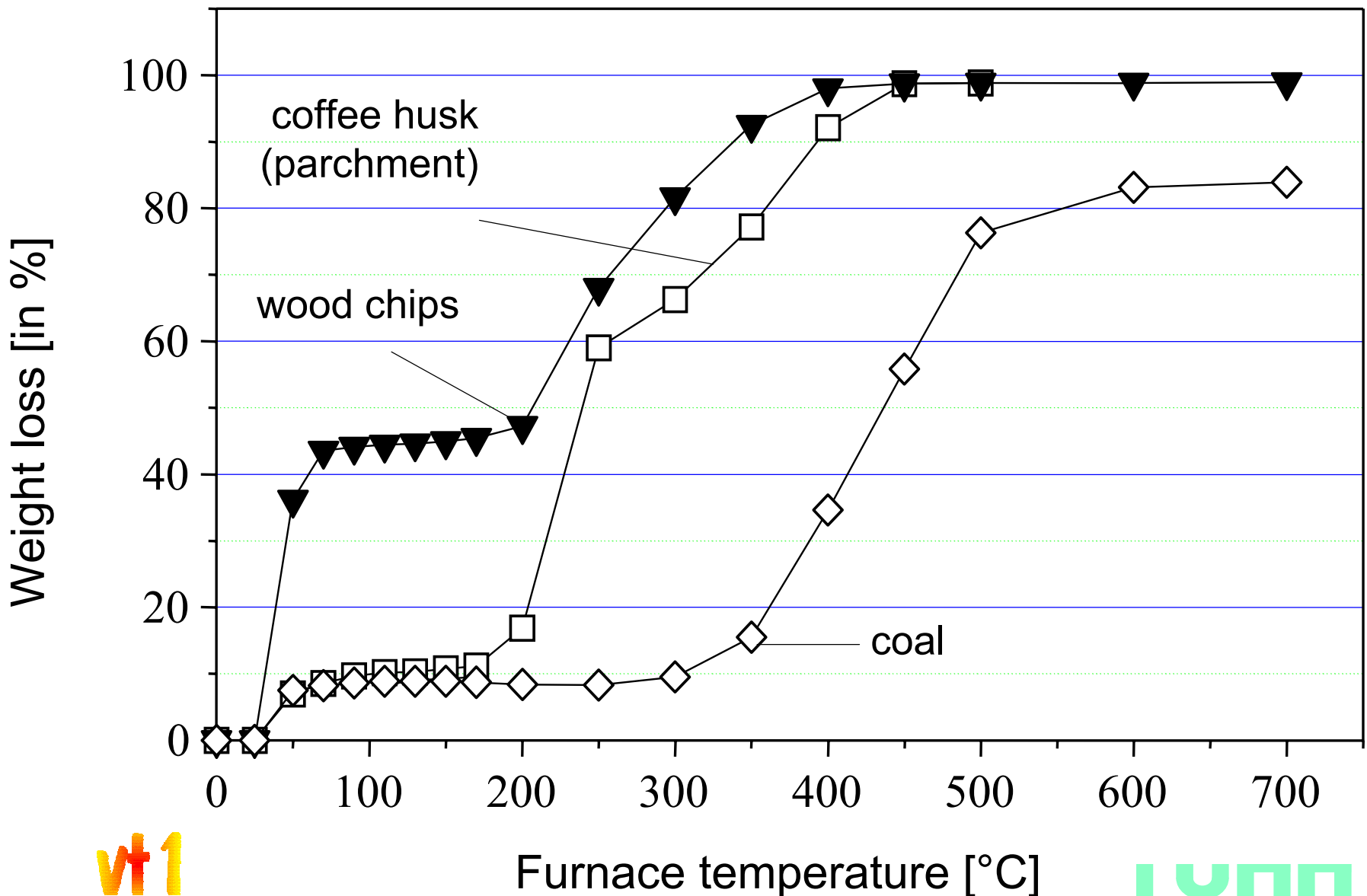
# Content of K and Cl in wheat straw



# Leaching of barley straw on the field



# Weight loss analysis



# Devolatilization characteristics

## 1. Experiments with increasing oven temperature

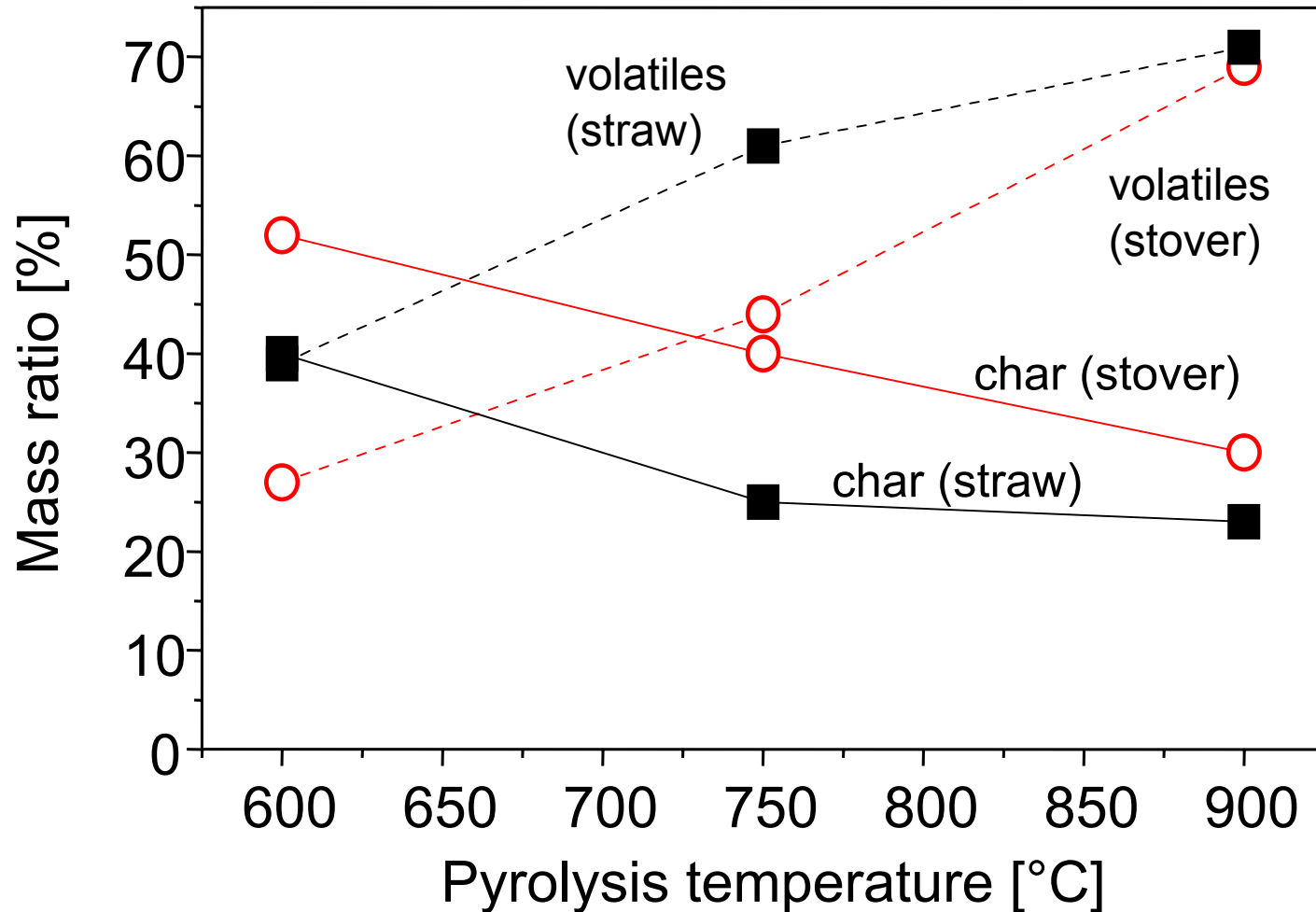
- devolatilization of biomass starts at low temperatures (160-200°C) and is completed around 500°C
- combustion of the char formed can take place at relatively low temperatures

## 2. Composition of the devolatilization products

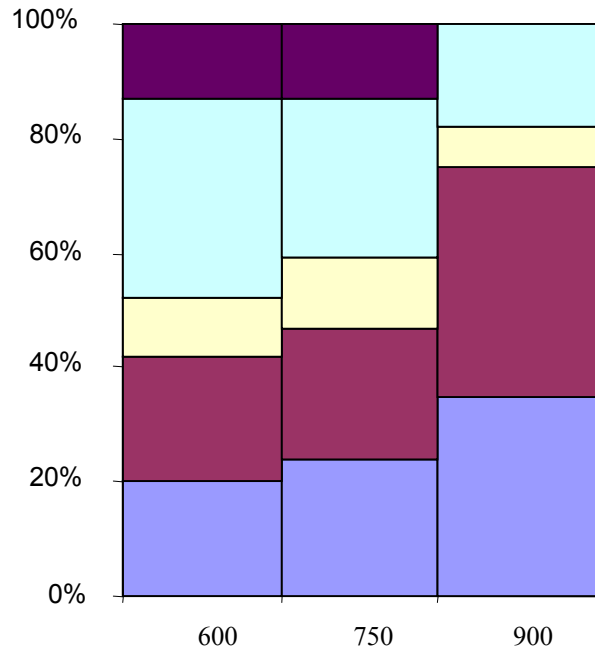
- with increasing pyrolysis temperature the mass fraction of char decreases while the quantity of volatiles increases
- with increasing pyrolysis temperature the fraction of CO<sub>2</sub> in the volatiles decreases whereas (H<sub>2</sub> + CO) increase



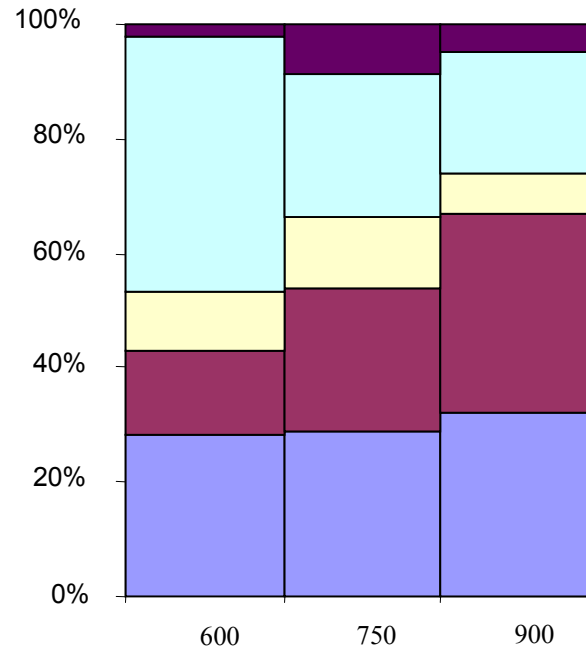
# Influence of the temperature on the relative quantities of char and gas



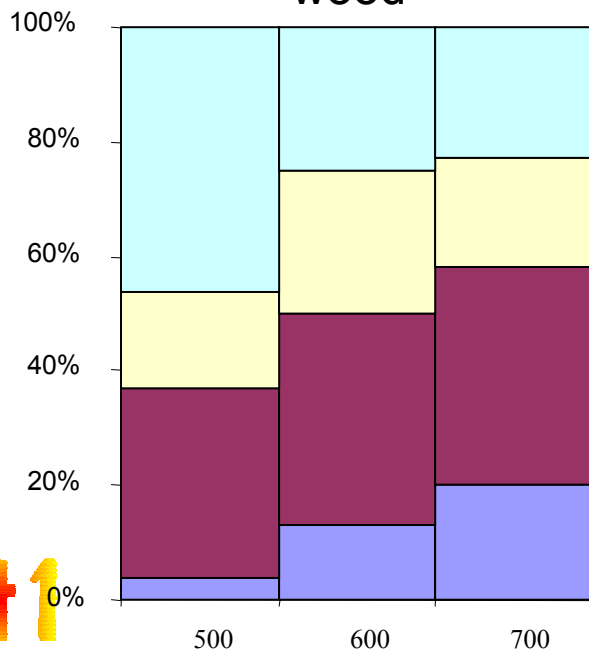
straw



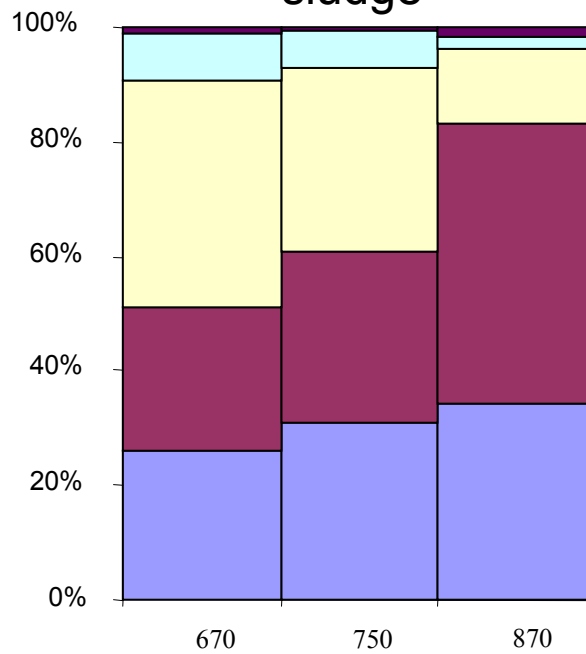
stover



wood



sludge



other

CO<sub>2</sub>

C<sub>x</sub>H<sub>y</sub>

CO

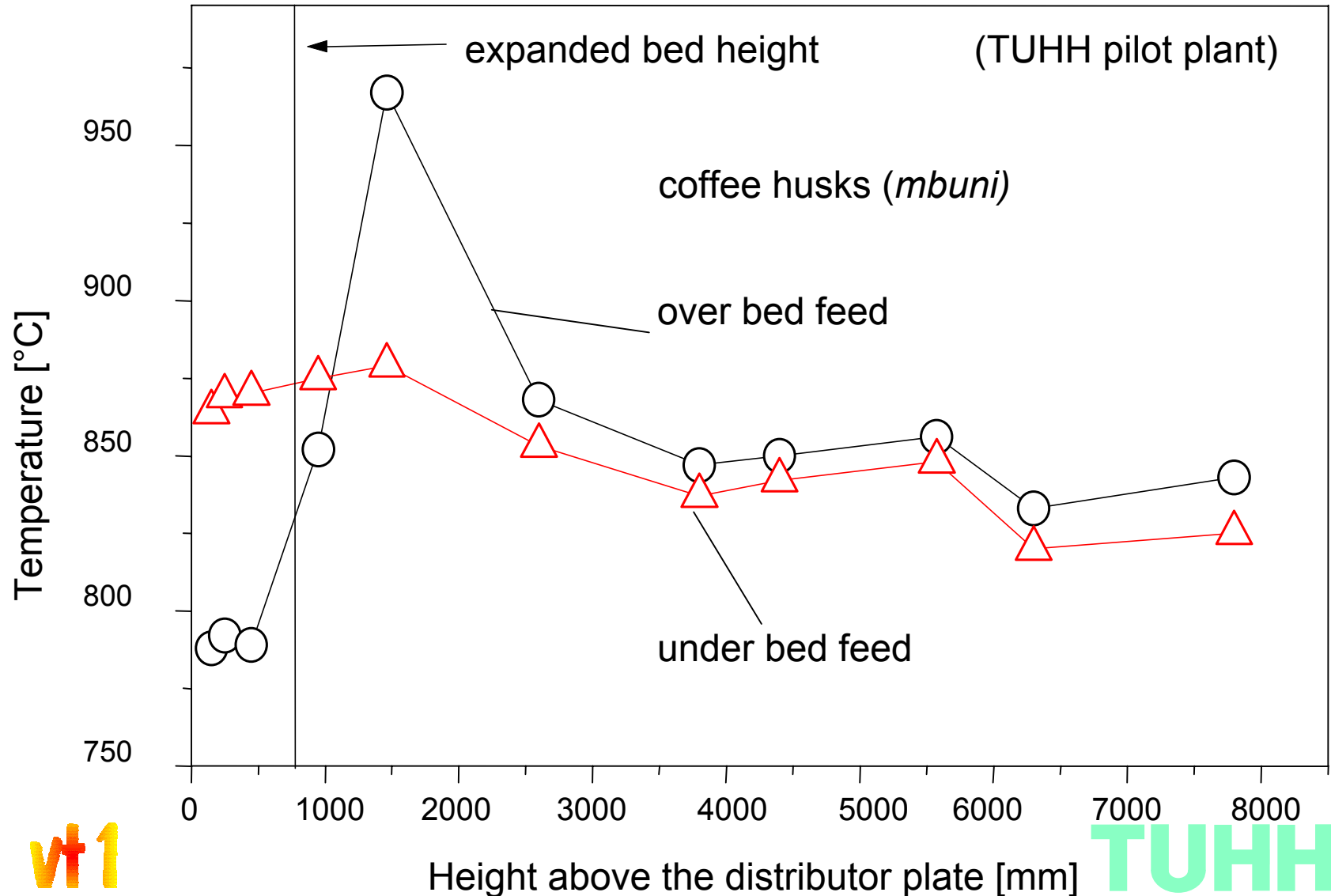
H<sub>2</sub>

# Effects of high volatile matter content

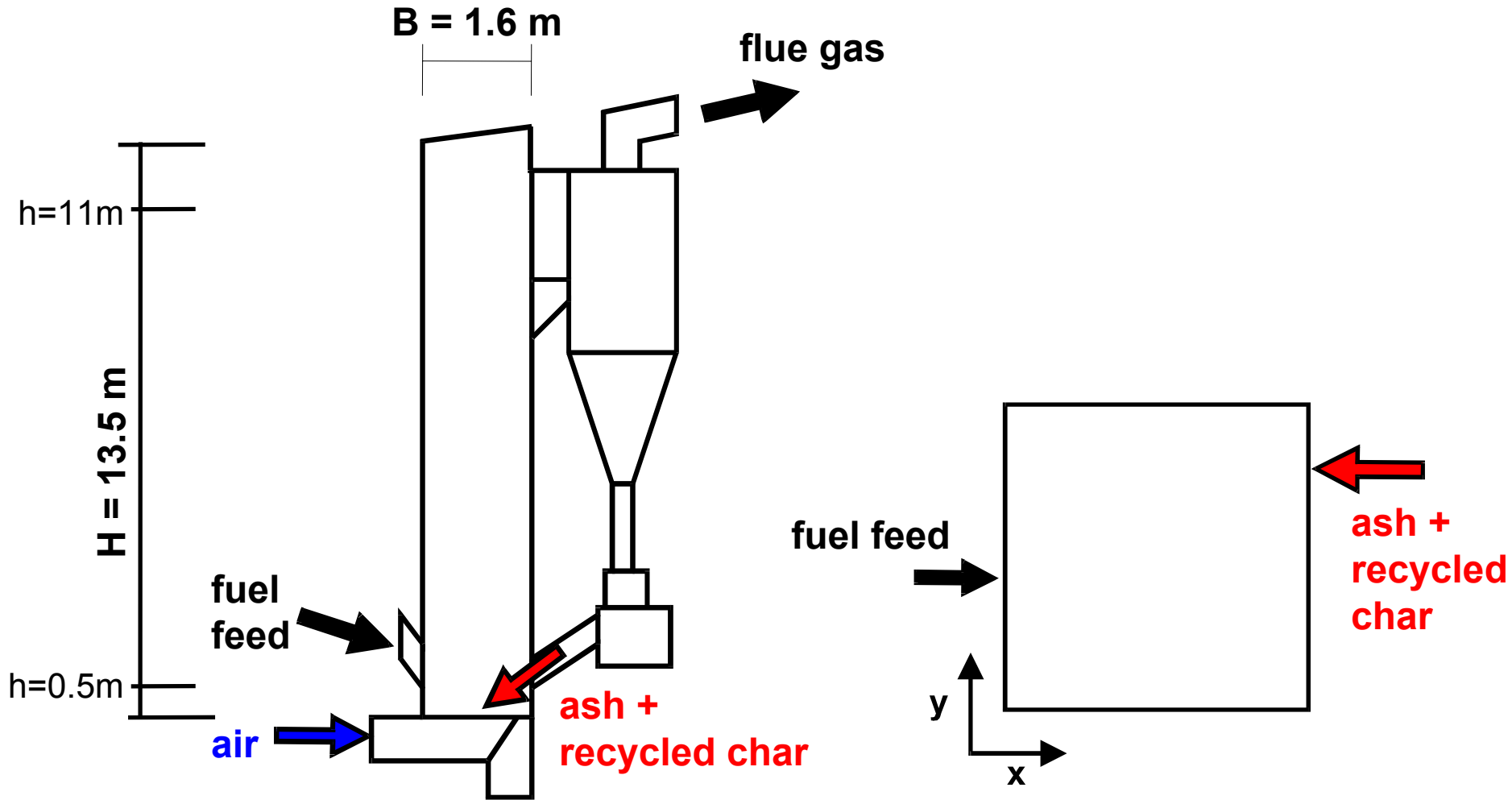
- in bubbling fluidized beds for sewage sludge combustion the freeboard temperature is generally higher than the bed temperature
- experiments with coffee husks show the same phenomenon for overbed feeding
- dry waste materials will release their volatiles close to the feed point since the lateral solids mixing is comparatively low; wet waste material will require more time for drying and devolatilization which provides an opportunity to mix over the bed's cross-section
  - ⇒ the high volatile matter content must be considered in the design of the fuel feeding configuration and the distribution of the combustion air

TUHH plants

# Axial temperature profiles in a bubbling fluidized bed combustor



# CFB-Boiler at Chalmers University, Sweden



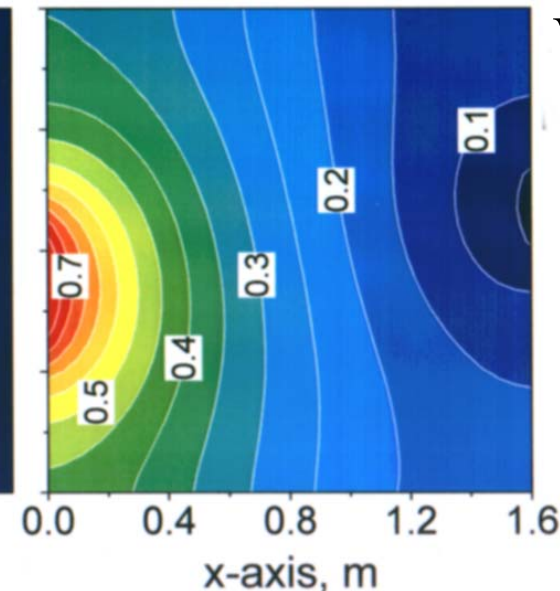
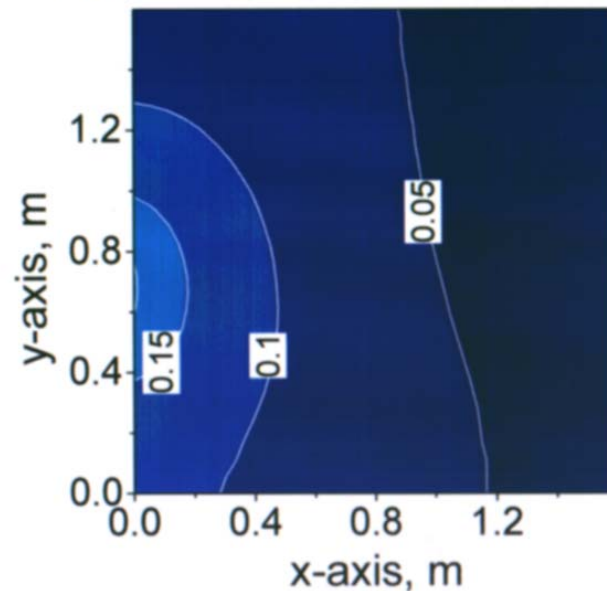
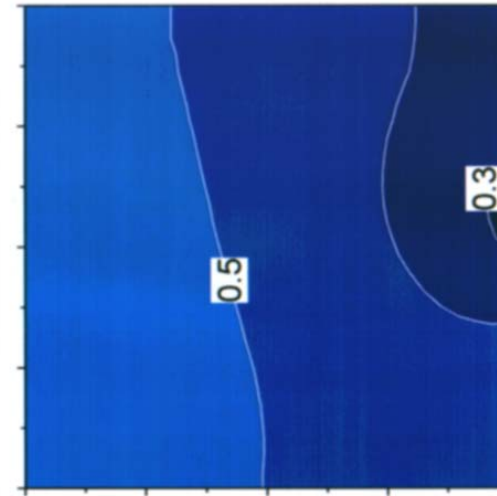
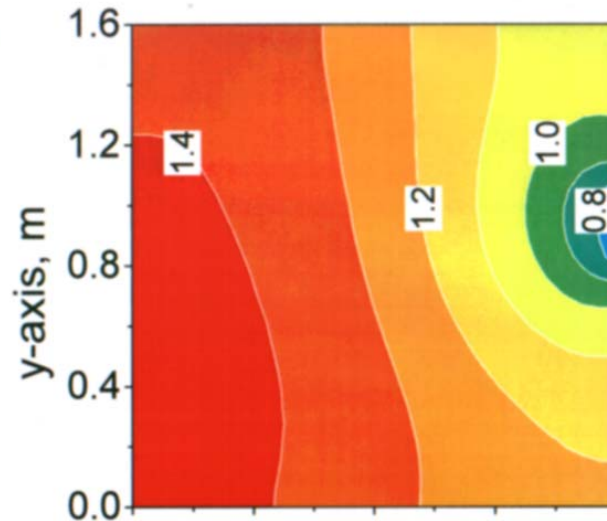
# Chalmers boiler, single stage combustion, $\lambda=1.2$

Bituminous coal  
(39 % volatiles, waf)

Wood  
(80 % volatiles, waf)

Char fraction  
wt.-%

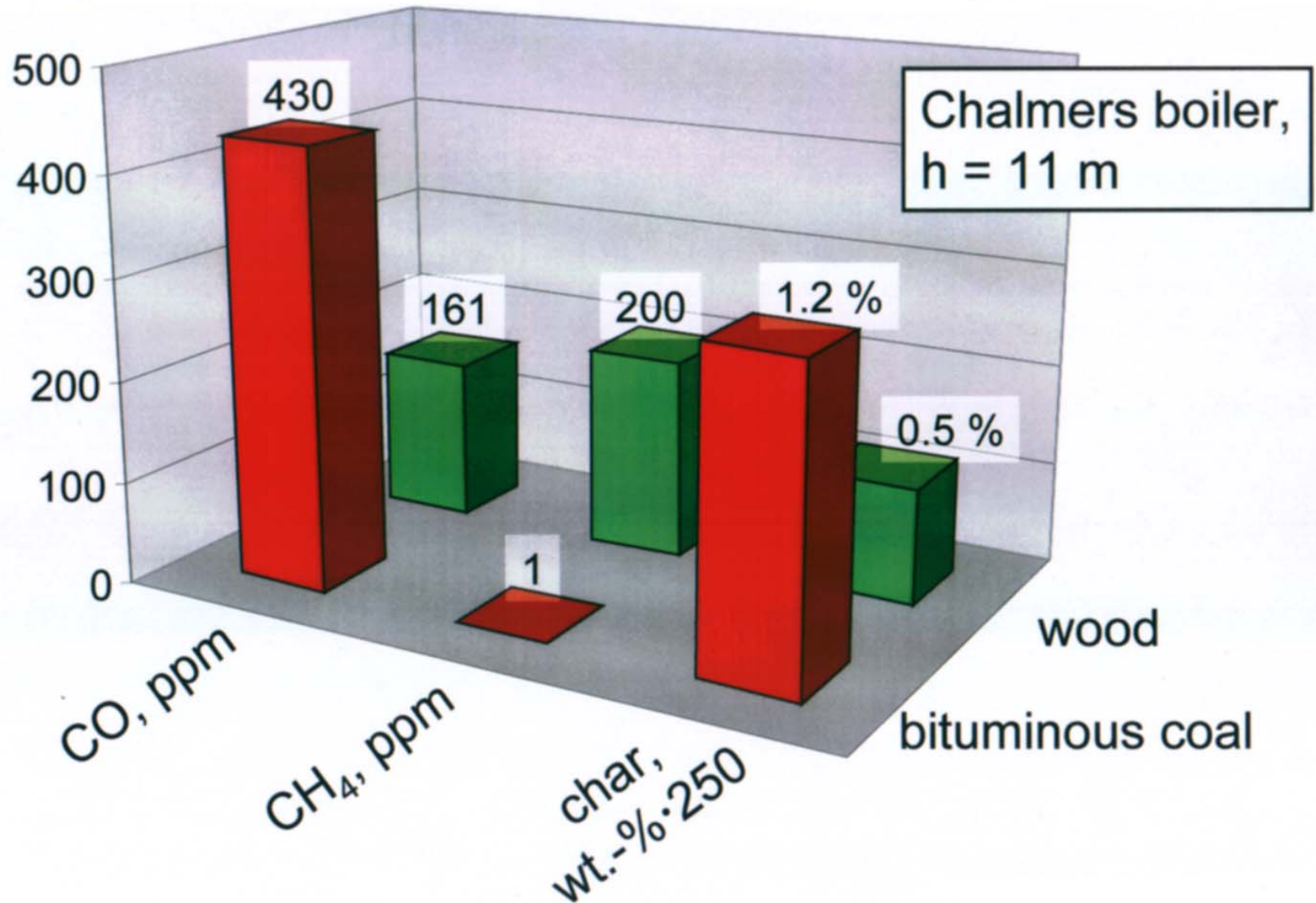
Conditions  
at fuel feed  
level  
 $h=0.5$  m



Volatile sources  
kg/m³/s

# Cross-section averages of concentrations at the combustion chamber outlet (before cyclone)

24





# Problems associated with the ash from agricultural residues

- bed material agglomeration in fluidized bed combustion
- fouling
- corrosion
- in case of co-combustion with coal the composition of the resulting ash is different from that of the coal ash
  - ash can perhaps no longer be used for cement making
- deactivation of the DeNO<sub>x</sub>-catalyst in pf boilers

# Chemical composition of ashes (wt.-%, wf)

|                         | SiO <sub>2</sub> | Fe <sub>2</sub> O <sub>3</sub> | P <sub>2</sub> O <sub>5</sub> | Al <sub>2</sub> O <sub>3</sub> | CaO  | Na <sub>2</sub> O | K <sub>2</sub> O |
|-------------------------|------------------|--------------------------------|-------------------------------|--------------------------------|------|-------------------|------------------|
| cotton husk             | 10.8             | 1.9                            | 4.0                           | 1.3                            | 20.7 | 1.3               | 49.6             |
| soy husk                | 1.7              | 2.5                            | 4.9                           | 7.4                            | 21.4 | 5.3               | 30.5             |
| coffee husk (mbuni)     | 13.5             | 2.2                            | 3.7                           | 3.9                            | 10.7 | 0.4               | 38.1             |
| coffee husk (parchment) | 16.6             | 2.4                            | 3.4                           | 4.5                            | 9.8  | 0.5               | 36.9             |
| sewage sludge           | 38.3             | 12.5                           | 15.4                          | 14.8                           | 9.1  | 2.2               | 2.2              |
| wood                    | 12.8             | 5.2                            | 2.1                           | 4.1                            | 45.2 | 0.6               | 0.5              |
| peat                    | 24.6             | 8.2                            | 5.4                           | 8.1                            | 31.7 | 0.4               | 0.6              |
| bituminous coal         | 43.7             | 10.2                           | 0.3                           | 24.7                           | 5.8  | 0.9               | 3.2              |

→ ashes of agricultural residues are characterized by high contents of K<sub>2</sub>O and (in some cases) Na<sub>2</sub>O

# Ash melting temperatures

|           | K <sub>2</sub> O, wt.-% | initial deformation temperature, °C | hemisphere temperature, °C | flow temperature, °C |
|-----------|-------------------------|-------------------------------------|----------------------------|----------------------|
| straws    |                         |                                     |                            |                      |
| barley    | 40.3                    | 730 – 800                           | 850 – 1050                 | 1050 – 1200          |
| oat       | 40.3                    | 750 – 850                           | 1000 – 1100                | 1150 – 1250          |
| rye       | 19.2                    | 800 – 850                           | 1050 – 1150                | 1300 – 1400          |
| Wheat     | 6.6                     | 900 – 1050                          | 1300 – 1400                | 1400 – 1500          |
| grains    |                         |                                     |                            |                      |
| rye*      |                         | 710                                 |                            | 810                  |
| wheat*    |                         | 687                                 | 887                        | 933                  |
| wood      |                         |                                     |                            |                      |
| spruce*   |                         | 1426                                | 1600                       | 1583                 |
| coals     |                         |                                     |                            |                      |
| Colombian | 2.2                     | 1283                                | 1353                       | 1427                 |
| Russian   | 2.4                     | 1282                                | 1363                       | 1404                 |

\* from Kaltschmitt & Hartmann

# Agglomeration problems in fluidized bed combustion systems

- melting ash may act as a binder for bed particles
- alkali compounds may react with Si compounds of bed particles:  
 $2 \text{SiO}_2 + \text{Na}_2\text{CO}_3 \rightarrow \text{Na}_2\text{O} \cdot 2\text{SiO}_2 + \text{CO}_2$ , melting point  $874^\circ\text{C}$   
 $4 \text{SiO}_2 + \text{K}_2\text{CO}_3 \rightarrow \text{K}_2\text{O} \cdot 4\text{SiO}_2 + \text{CO}_2$ , melting point  $764^\circ\text{C}$
- if sufficient  $\text{Fe}_2\text{O}_3$  is available the risk of agglomeration is reduced since reaction with the alkali compounds according to  
 $\text{Fe}_2\text{O}_3 + \text{X}_2\text{O} \rightarrow \text{X}_2\text{Fe}_2\text{O}_4$   
 $\text{Fe}_2\text{O}_3 + \text{X}_2\text{CO}_3 \rightarrow \text{X}_2\text{Fe}_2\text{O}_4 + \text{CO}_2$   
 leads to compounds with melting temperatures  $> 1135^\circ\text{C}$
- beginning of agglomeration may be detected by fluctuations in bed temperature and pressure differential

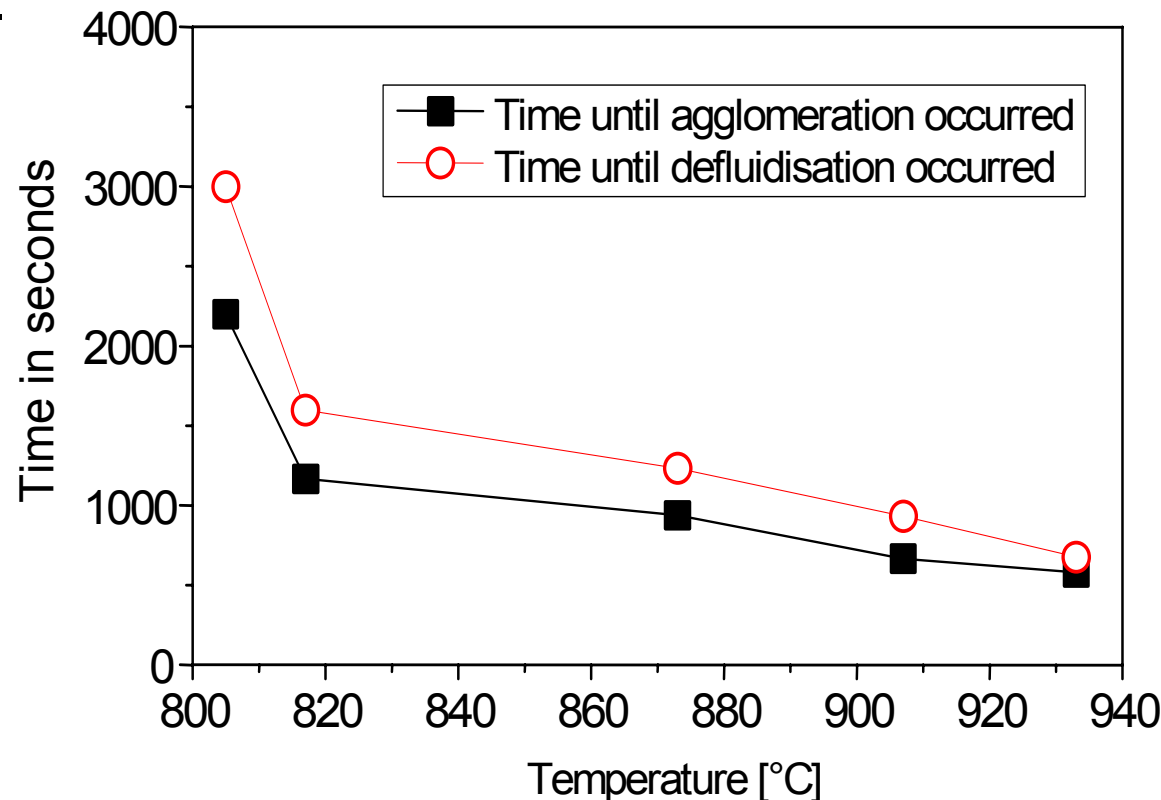
# Combustion of coffee husks in the fluidized bed



overbed-feeding, view from the distributor level after bed has been taken out, agglomerate layer has formed in the freeboard

# Agglomeration in fluidized bed combustors

- tendency towards agglomeration may be counter-acted - to a limited extent - by
  - vigorous movement of the bed particles at high fluidizing velocity (circulating rather than bubbling fluidized bed)
  - use of coarser bed material
- increasing bed temperature increases the risk of agglomeration



# How to reduce the risk of agglomeration?

- ash melting may be estimated from analysis:
- flow temperature  $T_{FP} = 1369 - 43.4 K + 192.7 Ca - 698 Mg$ , °C, where K, Ca, Mg are the contents of the elements in wt.-%
  - Ca-containing additives may be used to increase  $T_{FP}$  of fuel ash (limestone injection into the splash zone of a rice stalk fluid bed combustor prevented fouling)
  - other additives have also turned out to be effective (addition of 3 wt.-% kaolin to chopped oat straw increased deformation temperature from 770 to 1200-1280 °C)
- Change of bed material sand against other materials (dolomite, alumina, ...)
  - Saxena et al. cofired peanut hull pellets with propane: agglomeration with silica sand, no agglomeration with alumina

# How to reduce the risk of agglomeration?

- blending of risky fuels with other fuels (e.g. blending of coffee husks with 20 % coal containing 40 % ash reduced K in fuel ash from 43.8 to 13.5 %)
- leaching of K from straw by rain increases initial deformation temperature by 100-150 K



# Slagging and fouling

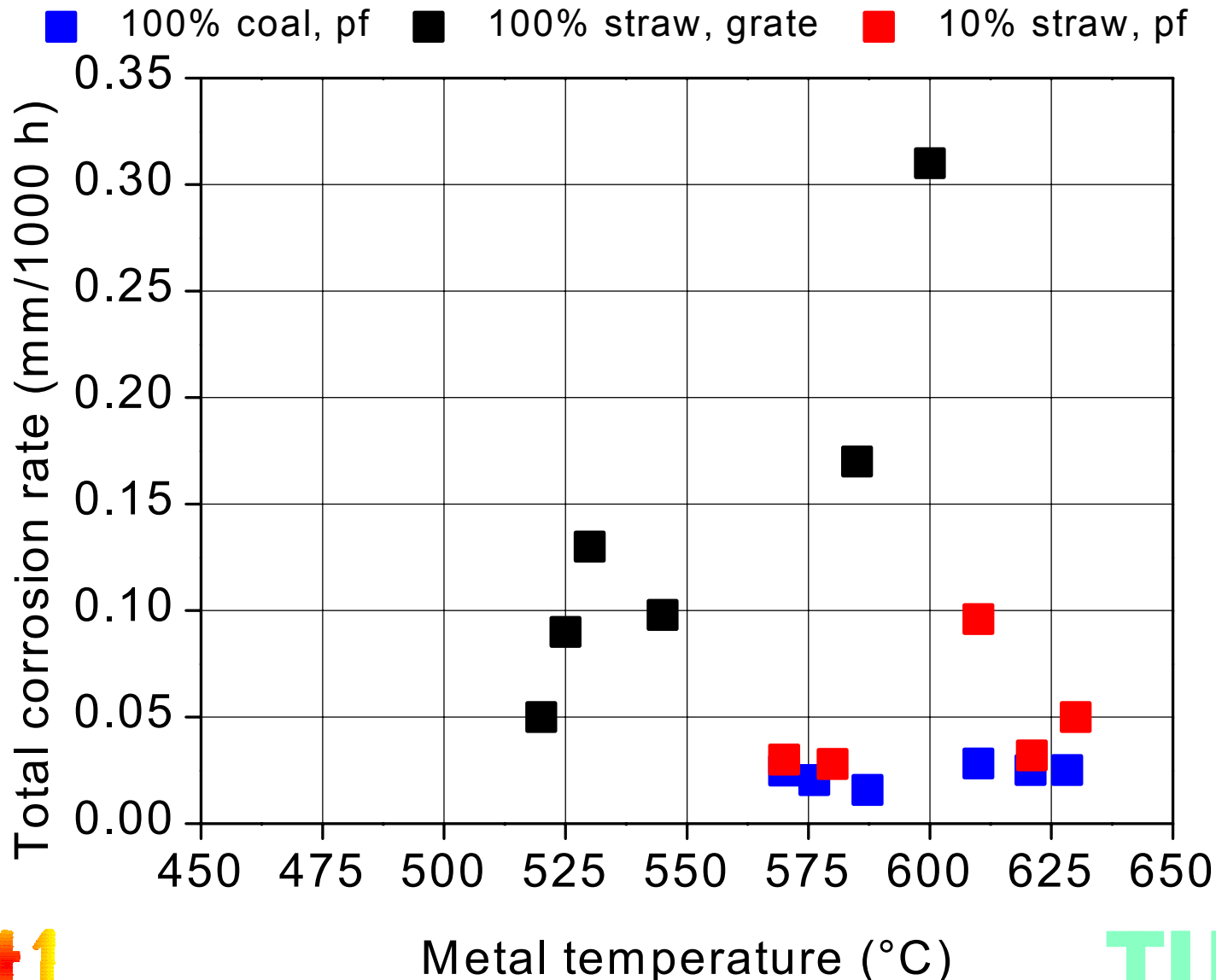
- is quite often experienced in combustors firing 100 % straw
  - Muthukrishnan et al. (1995) in 10 MW FBC plant firing rice stalk experienced ash deposition on superheater tubes in flue gas path
  - for straw combustion in grate-fired boilers in Denmark a lot of trouble (ash fusion on the grate, slagging in the furnace, fouling on convective surfaces) is reported
- in coal industry an alkali index is used
  - above 0.17 kg alkali/GJ: fouling probable –
  - e.g. 0.07 kg/GJ for a polish coal, but 1.0 kg/GJ for rice hulls, 1.1 kg/GJ for wheat straw, 1.6 kg/GJ for rice straw
- fouling must be expected with agricultural residues and has to be taken into account in the design and operation of the combustion system

# Corrosion

- is a problem in the mono-combustion of straw: in Danish straw-fired district heating boilers underneath ash deposits red layers of corrosion products found on convection tubes
- corrosion rates may be acceptable for co-combustion with coal if the energy share of the straw is about 10-20 %
- potassium and chlorine are the key elements

# Corrosion tests at superheater probes

Material: X20CrMo V12.1 (Spliethoff 1999, Sander 2000)



# Emissions from the combustion of agricultural residues

- $\text{NO}_x$ : single-stage combustion of
 

|             |         |                   |
|-------------|---------|-------------------|
| rice husk   | 200-370 | mg/m <sup>3</sup> |
| straw       | 400     | mg/m <sup>3</sup> |
| coffee husk | 400     | mg/m <sup>3</sup> |

 →staged combustion may be necessary
- $\text{SO}_2$ : due to low content in fuel no special measures required
- $\text{HCl}$ : typically 20-120 mg/m<sup>3</sup>,  
whether  $\text{HCl}$  washing is necessary depends on legal conditions  
(EU directive for waste combustion applies for co-combustion  
but not for mono-combustion, special regulations for biomass  
fired boilers)
- dioxins: generally less than 0.1 ng TE/m<sup>3</sup>  
(Danish straw fired power plants < 0.01 ng)
- particulate emissions:  
80 % of the ash particle mass released from the combustion of  
wood and grass is associated with particle sizes < 1  $\mu\text{m}$

# Legal issues associated with co-firing of wastes

## Legal basis for EU:

Directive 2000/76/EC of December 4, 2000  
on incineration of waste

- Directives 86/369/EEC, 89/429/EEC, 94/67/EC are replaced as from December, 2005  
→one directive now for waste and hazardous waste incineration and co-incineration
- air emission limit values for co-incineration  
→emission limit values depending on the plant size and plant process
- emission limit values for discharge of waste water from cleaning of exhaust gases

# Operating Conditions

- For slag and ashes: Total Organic Carbon (TOC) content  $< 3 \%$  or loss of ignition (LOI)  $< 5 \%$  of the dry weight of the material
- Waste incineration: temperature of the gas resulting from the process  $\geq 850 \text{ }^{\circ}\text{C}$  for two seconds after the last injection of combustion air
- Co-incineration: temperature of the gas resulting from the process  $\geq 850 \text{ }^{\circ}\text{C}$  for two seconds
- For hazardous wastes with a content of halogenated organic substances  $> 1 \%$  the temperature has to be raised to  $> 1100 \text{ }^{\circ}\text{C}$
- Automatic system required to prevent waste feed
  - during start up
  - when  $T \leq 850 \text{ }^{\circ}\text{C}$  or  $\leq 1100 \text{ }^{\circ}\text{C}$
  - when exceeding any emission limit values

# Emission limit values

| <b>Polluting substances</b>           | <b>EU-Directive</b>                                                                                                                  | <b>17<sup>th</sup> BImSchV</b>                                                                                                       | <b>13<sup>th</sup> BImSchV</b>                                                                                                |
|---------------------------------------|--------------------------------------------------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------------------------------------------|-------------------------------------------------------------------------------------------------------------------------------|
|                                       | waste incineration<br>Daily average values<br>in mg/m <sup>3</sup><br>standard conditions,<br>dry basis,<br>11 vol.-% O <sub>2</sub> | waste incineration<br>Daily average values<br>in mg/m <sup>3</sup><br>standard conditions,<br>dry basis,<br>11 vol.-% O <sub>2</sub> | power plants<br>Daily average values<br>in mg/m <sup>3</sup><br>standard conditions,<br>dry basis,<br>7 vol.-% O <sub>2</sub> |
| TOC                                   | 10                                                                                                                                   | 10                                                                                                                                   | 20                                                                                                                            |
| CO                                    | 50                                                                                                                                   | 50                                                                                                                                   | 250                                                                                                                           |
| NO <sub>x</sub> (as NO <sub>2</sub> ) | >6t/h: 200<br><6t/h: 400                                                                                                             | 200                                                                                                                                  | 800                                                                                                                           |
| SO <sub>2</sub>                       | 50                                                                                                                                   | 50                                                                                                                                   | 400                                                                                                                           |
| HCl                                   | 10                                                                                                                                   | 10                                                                                                                                   | 50                                                                                                                            |
| HF                                    | 1                                                                                                                                    | 1                                                                                                                                    | 2                                                                                                                             |
| Hg                                    | 0,05                                                                                                                                 | 0,03 (since 1999)                                                                                                                    | -                                                                                                                             |
| Dioxins and furans                    | 0,1·10 <sup>-3</sup>                                                                                                                 | 0,1·10 <sup>-3</sup>                                                                                                                 | -                                                                                                                             |

# German legislation for sludge combustion

- If heat release by sludge  $> 25 \%$   
→ 17<sup>th</sup> regulation of the German federal immission law (17<sup>th</sup>BImSchV)
- If heat release by sludge  $< 25 \%$   
→ averaged emission limits of the 13<sup>th</sup>BImSchV regulation (for power plants) and 17<sup>th</sup> regulation  
→ mixing rule
- If heat release by sludge  $< 25 \%$   
→ emission limits have to be kept as if the heat release of the waste were exactly 10 %



# Determination of the heat release

$$\frac{\dot{m}_{\text{waste}} \cdot H_{\text{u,waste}} + \dot{m}_{\text{support}} \cdot H_{\text{u,support}}}{\left( \dot{m}_{\text{waste}} \cdot H_{\text{u,waste}} + \dot{m}_{\text{support}} \cdot H_{\text{u,support}} \right) + \dot{m}_{\text{fuel}} \cdot H_{\text{u,fuel}}}$$

$\dot{m}_{\text{waste}}$ : mass flux of waste

$\dot{m}_{\text{fuel}}$ : mass flux of normally authorised fuel

$\dot{m}_{\text{support}}$ : mass flux of support fuel

$H_{\text{u,waste}}$ : heating value of waste

$H_{\text{u,fuel}}$ : heating value of normally authorised fuel

$H_{\text{u,support}}$ : heating value of support fuel

# Mixing rule

$$C_{\text{mix}} = \frac{V_{\text{fuel}} \cdot C_{13} + V_{\text{waste}} \cdot C_{17}}{V_{\text{fuel}} + V_{\text{waste}}}$$

$C_{\text{mix}}$

total emission limit value

$C_{13}$

emission limit value 13thBImSchV

$C_{17}$

emission limit value 17thBImSchV

$V_{\text{fuel}}$

exhaust gas volume from the incineration of waste at standard oxygen concentration of 7 %

$V_{\text{waste}}$

exhaust gas volume from the incineration of the fuel normally used in the plant at standard oxygen concentration of 11 %

# Reference conditions

Total emission limit values are referred to reference oxygen concentrations:

$$O_{2,\text{ref}} = \frac{V_{\text{fuel}} \cdot O_{2,13\text{thBImSchV}} + V_{\text{waste}} \cdot O_{2,17\text{thBImSchV}}}{V_{\text{fuel}} + V_{\text{waste}}}$$

It is forbidden to convert an emission value which is measured at a lower  $O_2$  value in off gas than the reference  $O_2$  value. If the  $O_2$  concentration in the exhaust gas is higher than the reference concentration it must be corrected.

# EU-Directive

- for co-combustion of waste:
  - ⇒ emission limit values for co-incineration plants (Annex II) according to mixing rule
  
- for co-combustion of hazardous waste:
  - heat release of hazardous waste  $< 10 \%$ :  
emission limits have to be kept as if the heat release of the waste was exactly  $10 \%$
  
  - heat release of hazardous waste  $\geq 10 \%$ :  
emission limit values for co-incineration plants (Annex II) according to mixing rule
  
  - heat release of hazardous waste  $> 40 \%$ :  
emission limit values for incineration plants (Annex V)

# Mixing rule for EU-Directive - (Annex II)

Emission limit value  $C$  depends on volume fraction of the exhaust gas from waste incineration, the plant size and the plant process.

$$C = \frac{V_{\text{waste}} \cdot C_{\text{waste}} + V_{\text{proc}} \cdot C_{\text{proc}}}{V_{\text{waste}} + V_{\text{proc}}}$$

$C$  : total emission limit value and oxygen content for standardisation respecting the partial volumes

$V_{\text{waste}}$  : exhaust gas volume resulting from the incineration of the waste

$C_{\text{waste}}$  : emission limit values set for incineration plants (Annex V)

$V_{\text{proc}}$  : exhaust gas volume resulting from the combustion of fuels normally used in the plant

$C_{\text{proc}}$  : emission limit values set for certain industrial sectors (Annex II)

# Standard conditions (EU)

- 273 K, 101,3 kPa, 11 % oxygen, dry gas in exhaust gas of incineration plants
- 273 K, 101,3 kPa, 3 % oxygen, dry gas in exhaust gas of incineration of waste oil
- 273 K, 101,3 kPa, 10 % oxygen, dry gas in exhaust gas of cement kilns

co-incineration:

- in the case of co-incineration, the results of the measurement shall be standardised at a total oxygen content as calculated in Annex II
- oxygen content for calculation of  $C_{\text{proc}}$ :
  - solid fuels: 6 %, dry gas
  - biomass: 6 %, dry gas
  - liquid fuels: 3 %, dry gas

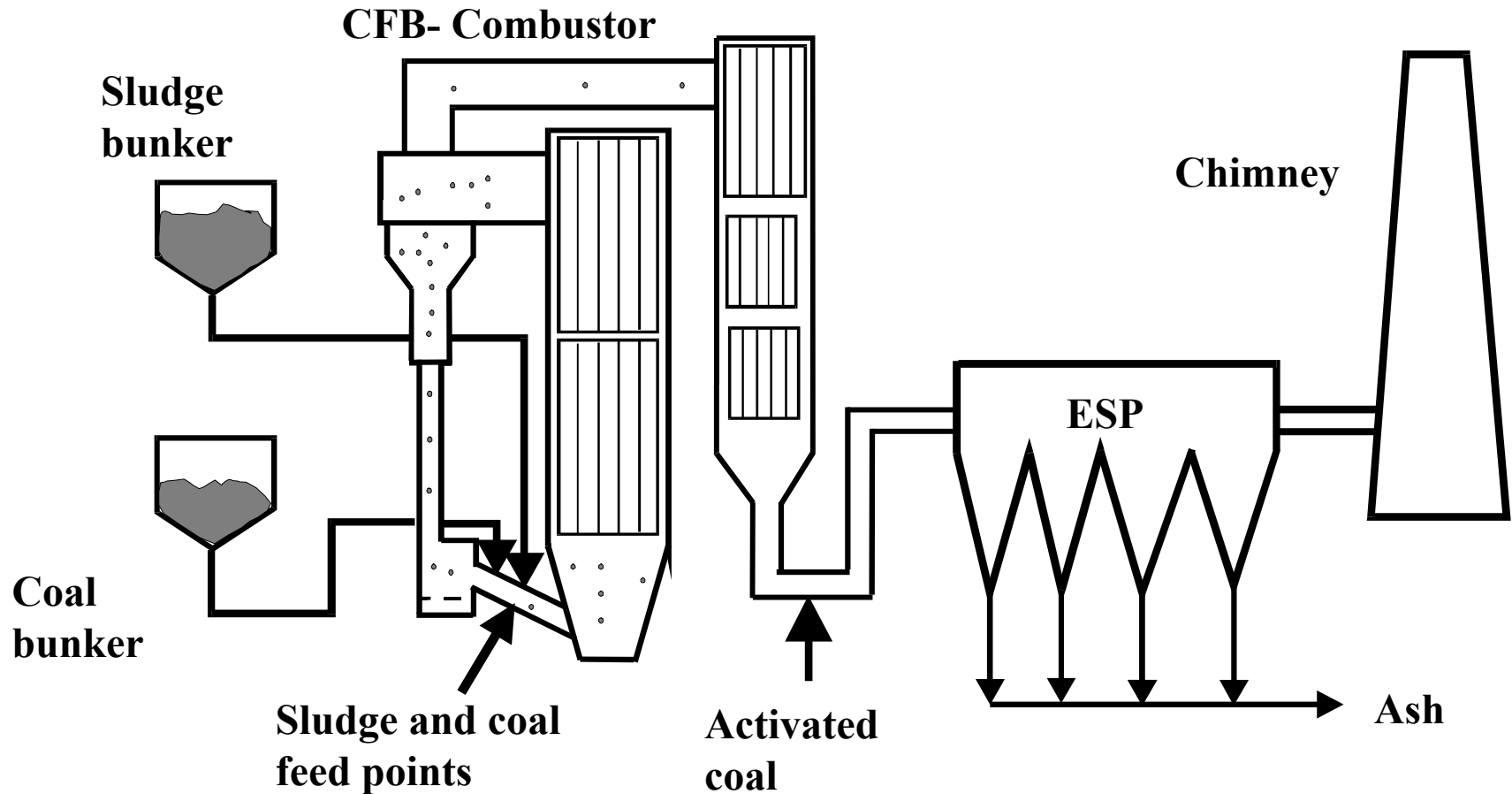
- Hochkant

# Grenaa





# Co-combustion of sewage sludge in Rheinbraun's Berrenrath Plant



# Co-combustion experience from Rheinbraun's Berrenrath Plant

- 120,000 t/a mechanically dewatered sludge is incinerated (about 40,000 t d.m. /a)
- Permission granted to burn 65,000 t d.m./a
- no problems with emissions after injection of active coke before the ESP was installed
- Hg emission is now  $4\mu\text{g}/\text{m}^3$

# Tacoma steam power plant - USA



## Details

Owner - Tacoma Public  
Utilities

Capacity - 50 MWe

Furnace - FBC

Fuel - coal: 20 %  
wood: 60 %  
RDF: 20 %

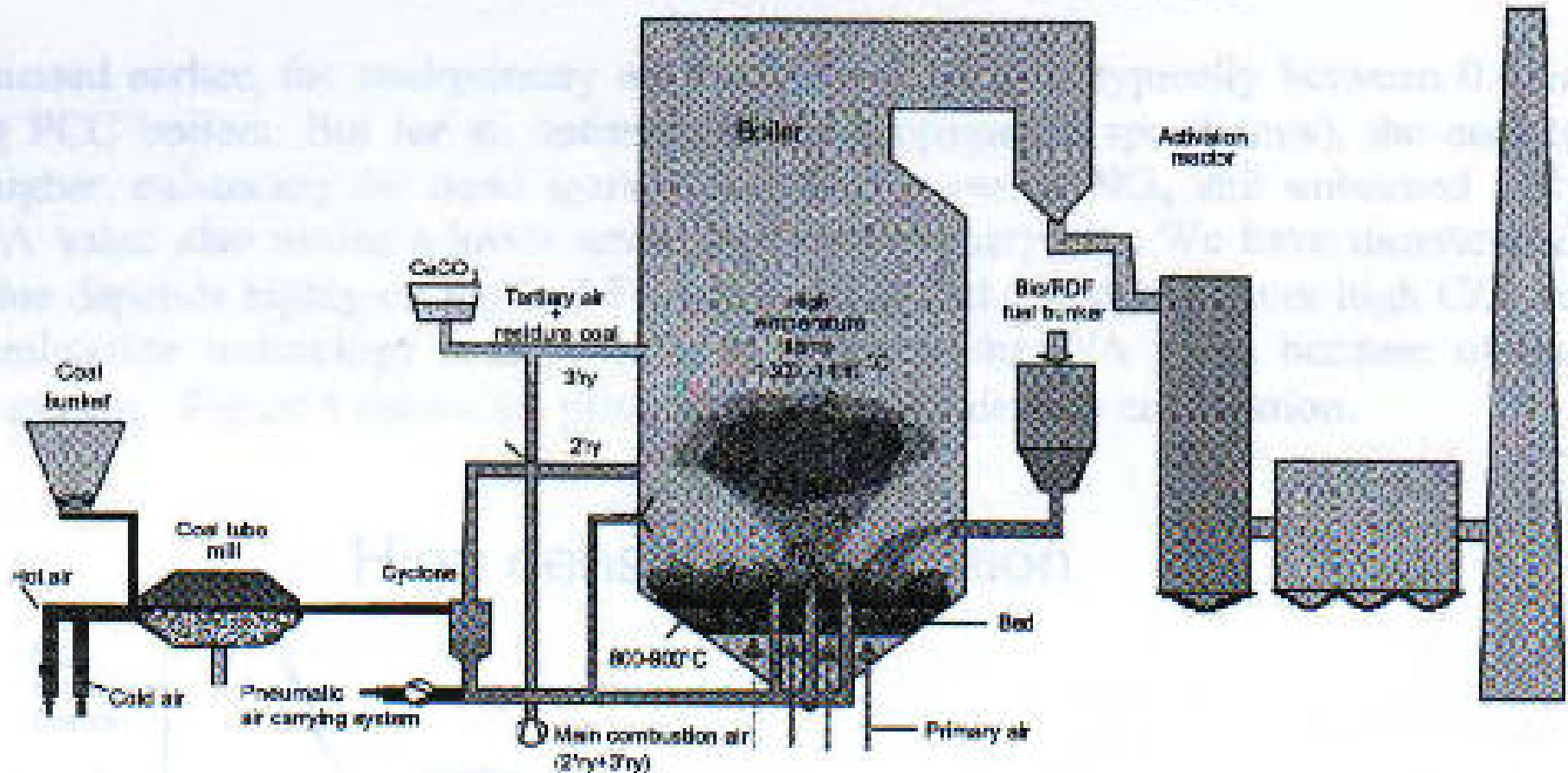
## Special features

- periodical removal of agglomerated bed media, glass, metals etc from bed
- 93 % HCl removal

# Experience gained at Tacoma steam power plant

- Co-combustion from 1991 to 1998
- All emission limits met
  - FBC+limestone injection led to low  $\text{SO}_2$  and  $\text{NO}_x$
  - 90 % HCl removal
- All byproducts recycled
  - Fly ash used for sludge stabilization and cement production
  - Aggregates used as road base material
- Economics
  - the renovation of the plant cost \$45m million
  - Project received grant from state of \$ 15 million
  - Plant was saving municipality of \$ 1 million per year by burning RDF
  - From 1997 the plant was making losses of \$ 2 million per year due to unfavorable costs of wood and cheap electricity from public utilities

# HAT-BFB multifuel combustion



# HAT-BFB multifuel combustion

- Biomass and part of coal burned in bed at low temperature (800-850 °C) to avoid agglomeration
- Remaining part of coal burned in freeboard at 1350-1450
- High freeboard temperature
  - Leads to higher combustion efficiency
  - Low NO<sub>x</sub> based on the reburning technique
- Large scale demonstration plant
  - at Rauhalahti Power Plant in Finland
  - Fuel: wood, bark and coal

# Experts ranking of co-firing technologies

By 80 experts at EU Seminar in Cottbus, 2000

| Fuel mixture             | PF  | FBC | CFB | PFBC | IGCC | GC  |
|--------------------------|-----|-----|-----|------|------|-----|
| Coal and biomass         | 2,6 | 3,3 | 3,5 | 1,9  | 2,3  | 2,6 |
| Coal and residual wood   | 2,3 | 3,4 | 3,8 | 1,7  | 2,0  | 2,6 |
| Coal and straw           | 1,9 | 2,5 | 3,0 | 1,6  | 1,6  | 1,9 |
| Coal and sewage sludge   | 2,7 | 2,9 | 3,2 | 2,4  | 2,5  | 2,4 |
| Coal and petroleum coke  | 2,1 | 2,0 | 2,5 | 2,1  | 3,7  | 2,3 |
| Coal and RDF             | 2,1 | 2,6 | 3,1 | 1,9  | 3,0  | 2,7 |
| Coal and municipal waste | 1,7 | 2,6 | 2,9 | 1,6  | 2,7  | 2,7 |
| Coal and plastics        | 1,9 | 2,3 | 2,3 | 2,0  | 2,7  | 2,9 |
| Average of ranking       | 2,2 | 2,7 | 3,0 | 1,9  | 2,6  | 2,5 |
| Position of ranking      | 5   | 2   | 1   | 6    | 3    | 4   |

# Conclusions

- Agricultural residues and sewage sludges are CO<sub>2</sub>-neutral fuels. Their use contributes to the global reduction of CO<sub>2</sub> emissions.
- Agricultural residues are difficult fuels due to high alkali contents and handling properties.
- Co-combustion with coals in existing power stations is an interesting option.
- Sewage sludges provide income for power stations
- Agricultural residues will only be used to a greater extent if economic incentives are sufficient.  
→ governmental programs, tax privileges and subsidies are necessary



# Acknowledgment

---

The contributions made by

Markus Fehr, TUHH

Markus Sanger, TUHH

Ernst-Ulrich Hartge, TUHH

Tom Ogada (Moi University, Eldoret/Kenya)

Zachary Siagi (Moi University, Eldoret/Kenya)

are gratefully acknowledged.